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Comprendre le monde,
construire l'avenir®

Recombination, excitation and dissociation of hydride molecular cations in low energy electron collisions

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A. Bultel¹², K. Hassouni², D. Talbi¹³, A. P. Hickman¹⁴, A. Faure¹⁵, J. Tennyson⁶
and I. F. Schneider^{1,3}

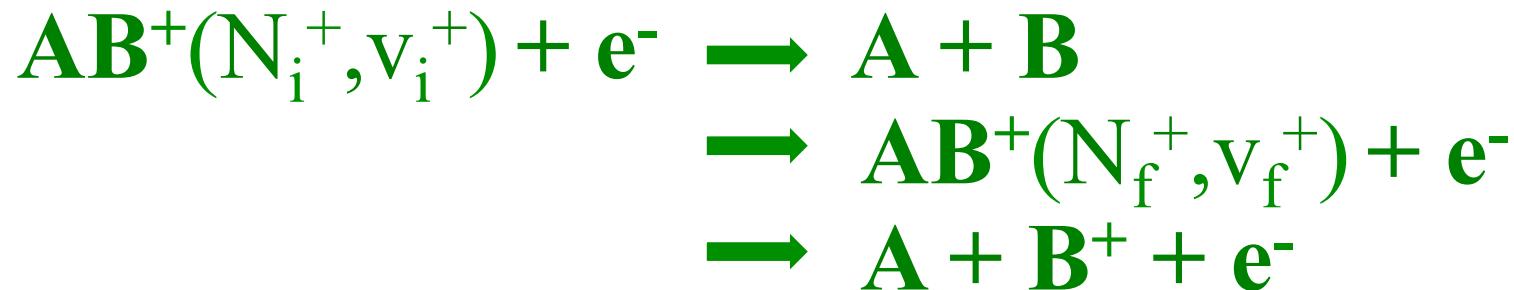
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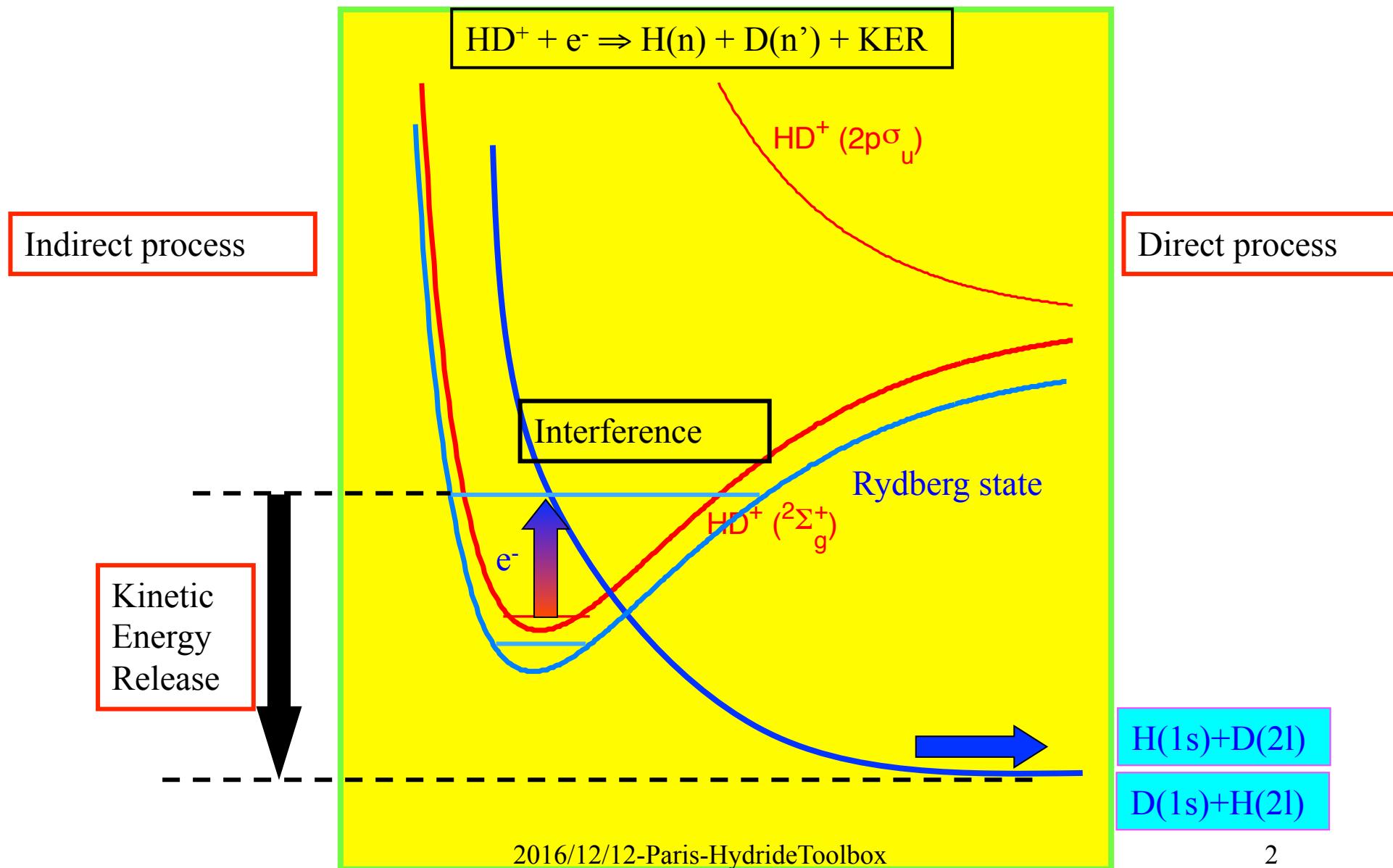
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Electron-cold molecular ion reaction: Dissociative Recombination



Storage ring measurements: concept

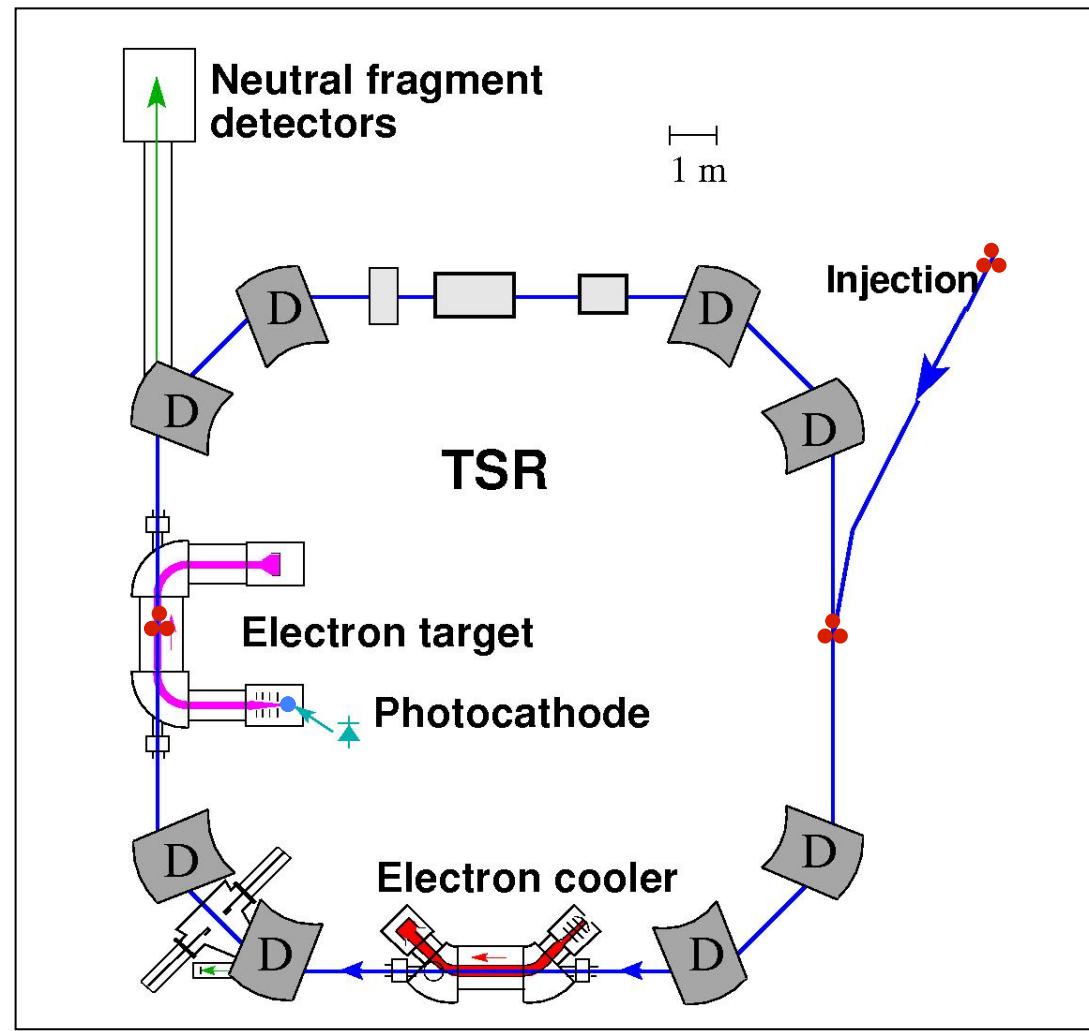
Advantages

- radiative relaxation of internal excitations (rotation, vibration)
- direct measurement
- 100% detection efficiency
- high resolution

• Vibrations cool



• Rotations stay forever



Benchmarking: a NOT-SO-Hydride system, H_2^+ (or isotopologues) + e⁻: maximum ACCURACY

HD⁺ + e⁻

Rotational transitions induced by collisions of HD⁺ ions with low-energy electrons

O. Motapon,^{1,2} N. Pop,³ F. Argoubi,⁴ J. Zs Mezei,^{2,5,6} M. D. Epee Epee,¹ A. Faure,⁷ M. Telmini,⁴ J. Tennyson,⁸ and I. F. Schneider^{2,5}

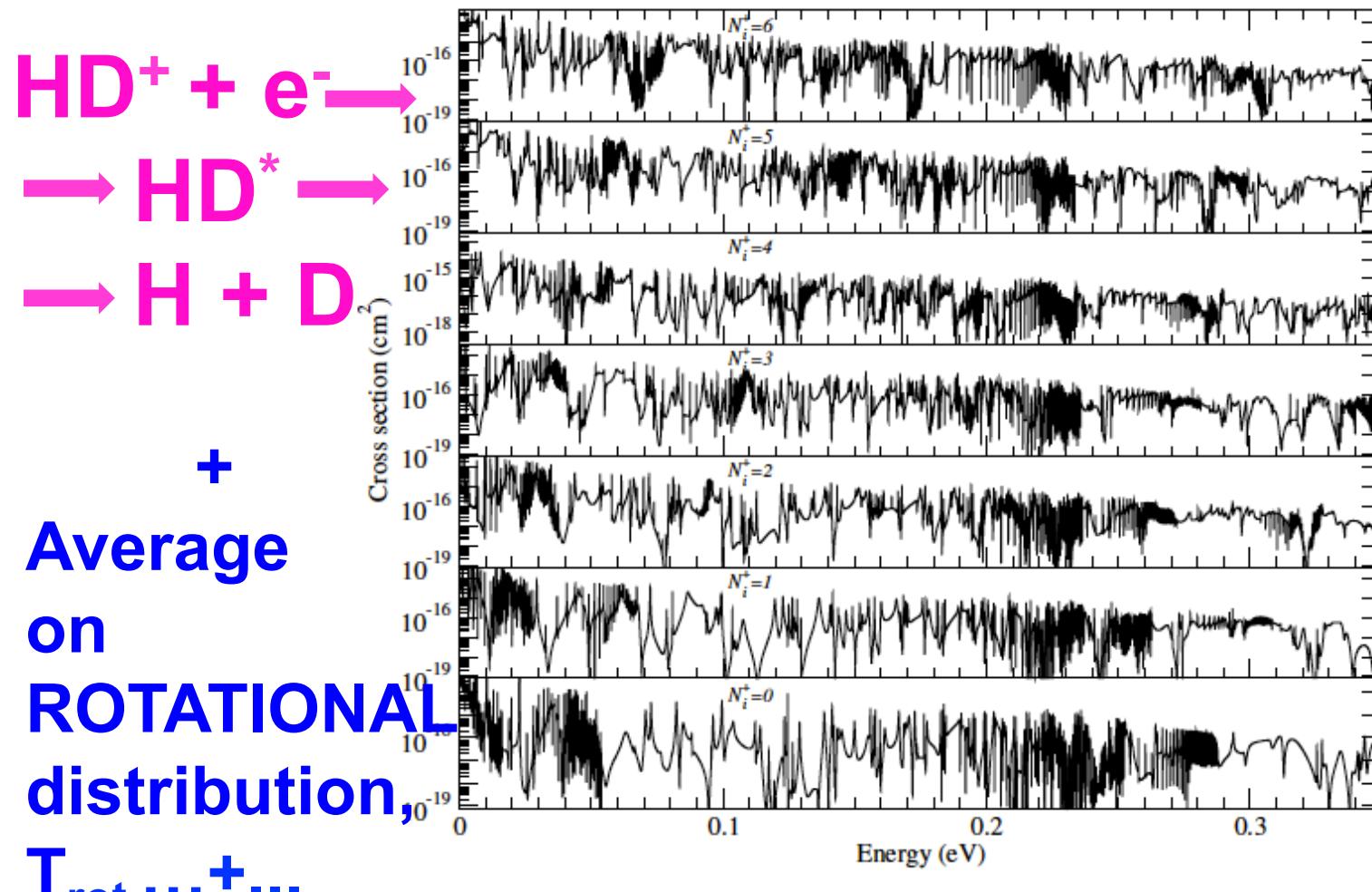
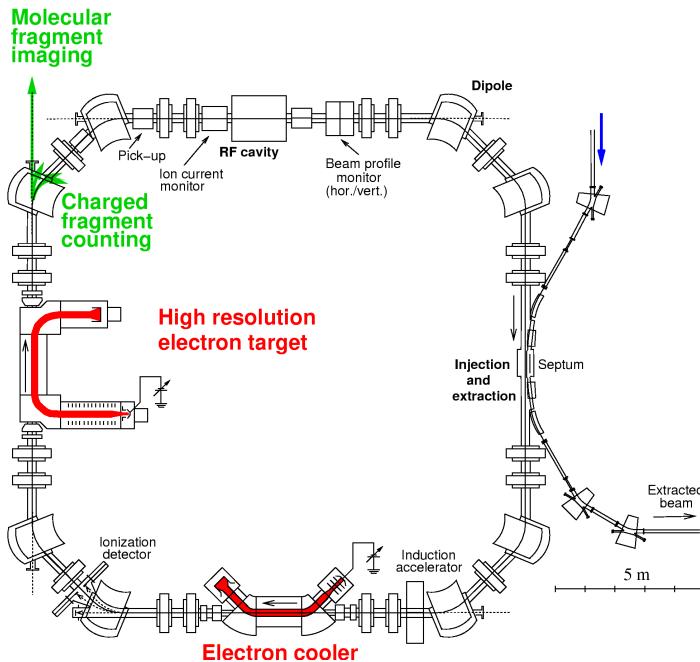


Figure 3. DR cross sections of HD⁺ initially in one of its lowest rotational level N_i^+ (vibrational ground state).



...+
**Convolution
with
ANISOTROPIC
Maxwell
Distribution:**

$$\alpha = \langle v\sigma \rangle = \iiint \sigma(v) v f(v_d, v) dv \quad (1)$$

$$f(v_d, v) = \frac{m}{2\pi k T_{e\perp}} \exp\left(-\frac{mv_\perp^2}{2k T_{e\perp}}\right) \sqrt{\frac{m}{2\pi k T_{e\parallel}}} \exp\left(-\frac{m(v_\parallel - v_d)^2}{2k T_{e\parallel}}\right) \quad (2)$$

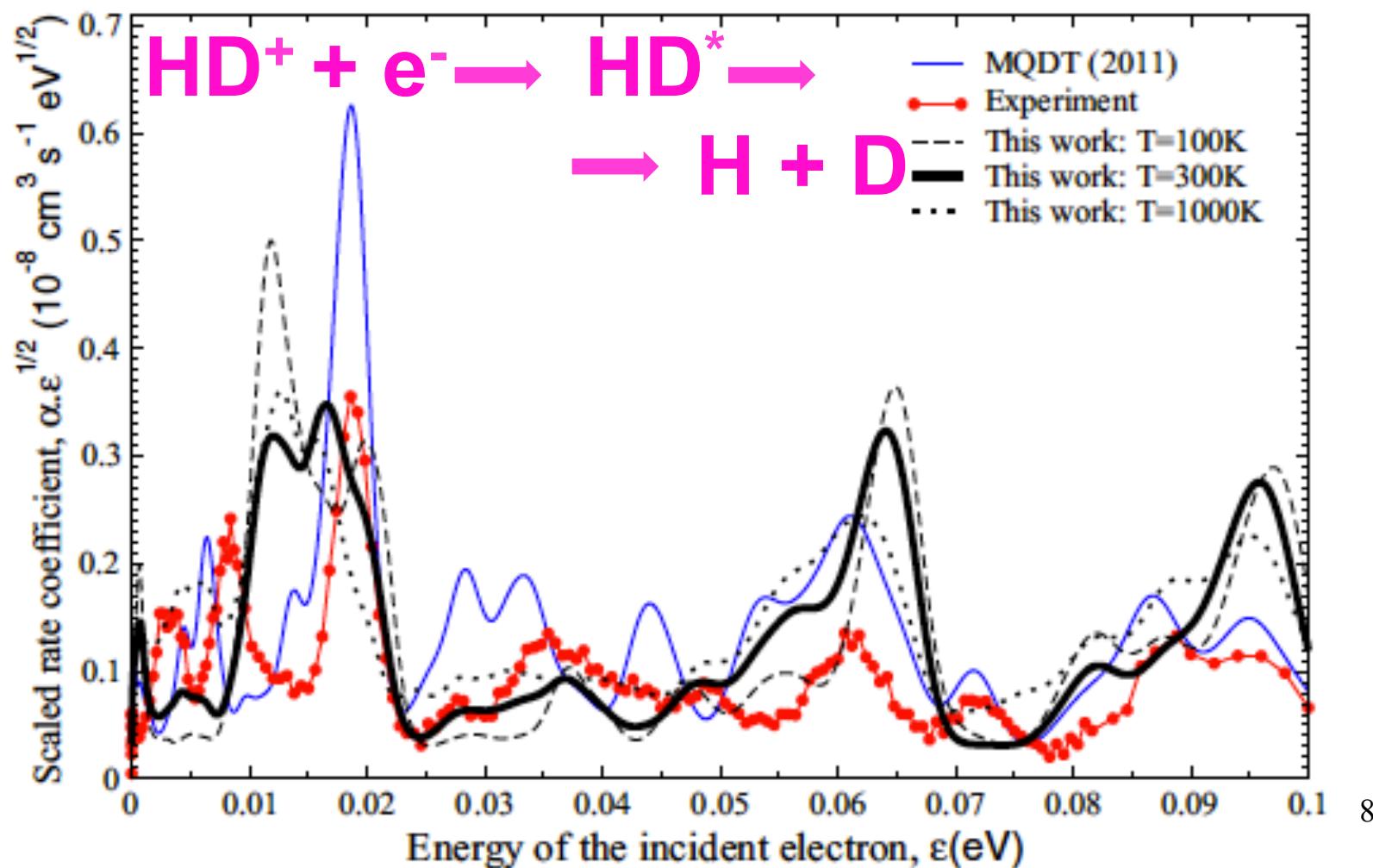
Best parameters:

$$T_{\text{long}} = 20 \mu\text{eV} = 0,23 \text{ K} = 0,16 \text{ cm}^{-1}$$

$$T_{\text{trans}} = 500 \mu\text{eV} = 5,80 \text{ K} = 4,03 \text{ cm}^{-1}$$

Rotational transitions induced by collisions of HD⁺ ions with low-energy electrons

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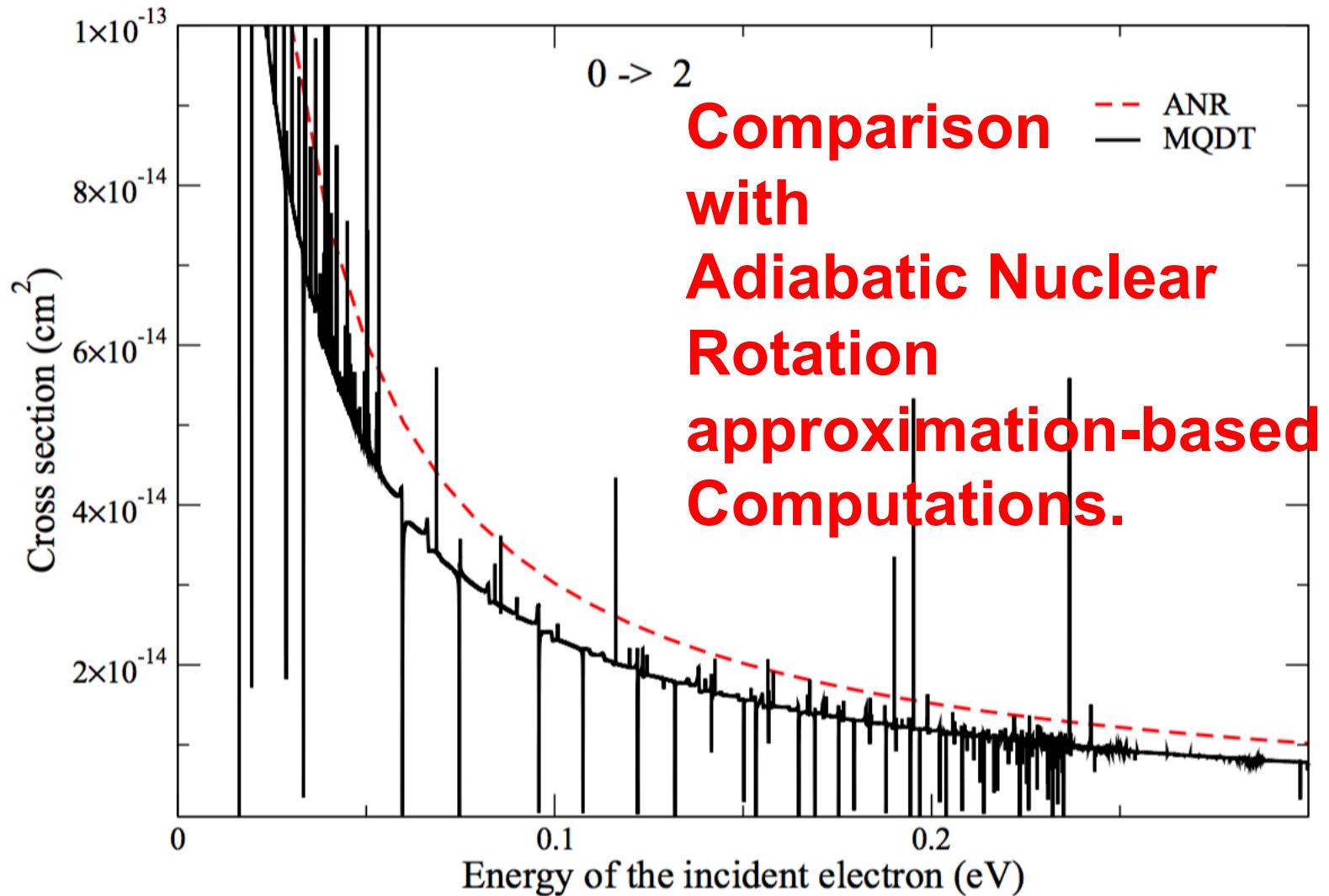


FIG. 2. (Color online) Zoom in the linear scale of the cross section for the transition $0 \rightarrow 2$, in the ground vibrational state of $\text{HD}^+(\text{X}^2\Sigma_g^+)$. Black solid curve: MQDT computations; red dashed curve: ANR approximation-based computations.

For the modelers: **Maxwell rate-coefficients**

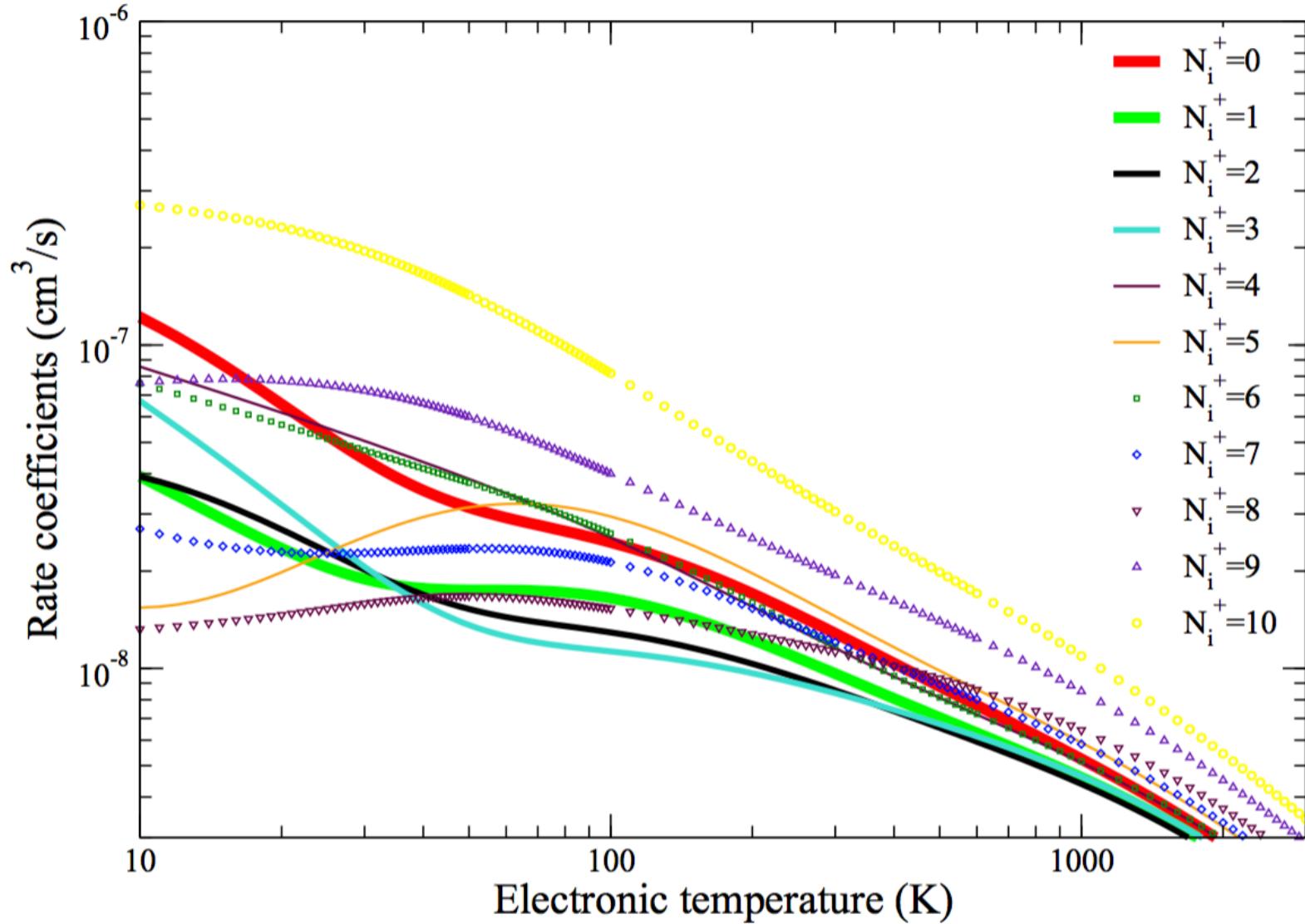


FIG. 9. (Color online) Maxwell isotropic rate coefficients for the dissociative recombination $\text{HD}^+(\text{X } ^2\Sigma_g^+)$ with $v_i^+ = 0$ as a function of initial rotational level, $N_i^+ = 0$ to 10.

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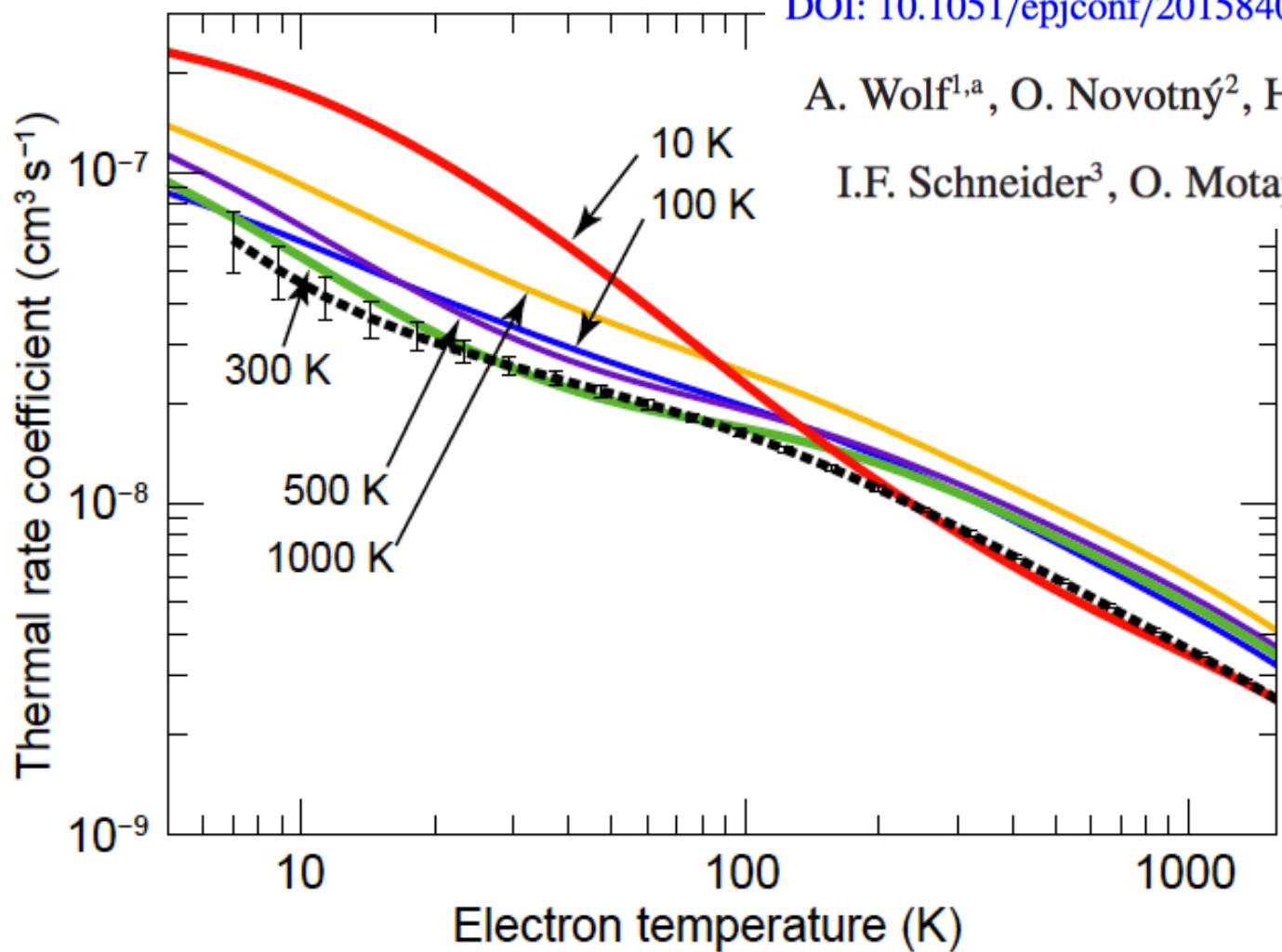
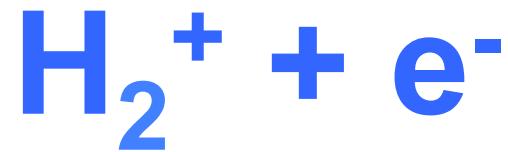
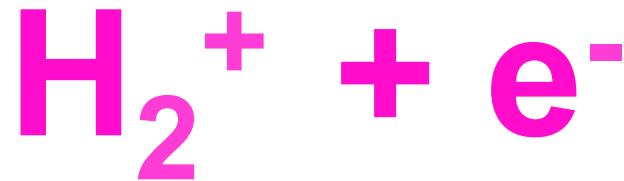


Fig. 2. Thermal rate coefficient for HD⁺ DR as a function of the electron temperature. Experiment is given by a the thick dashed line; 1σ estimated experimental errors due to uncertainties in the electron beam temperatures are also indicated. Theory is obtained for the given rotational temperatures.





Reactive collisions of very low-energy electrons with H_2^+ : rotational transitions and dissociative recombination

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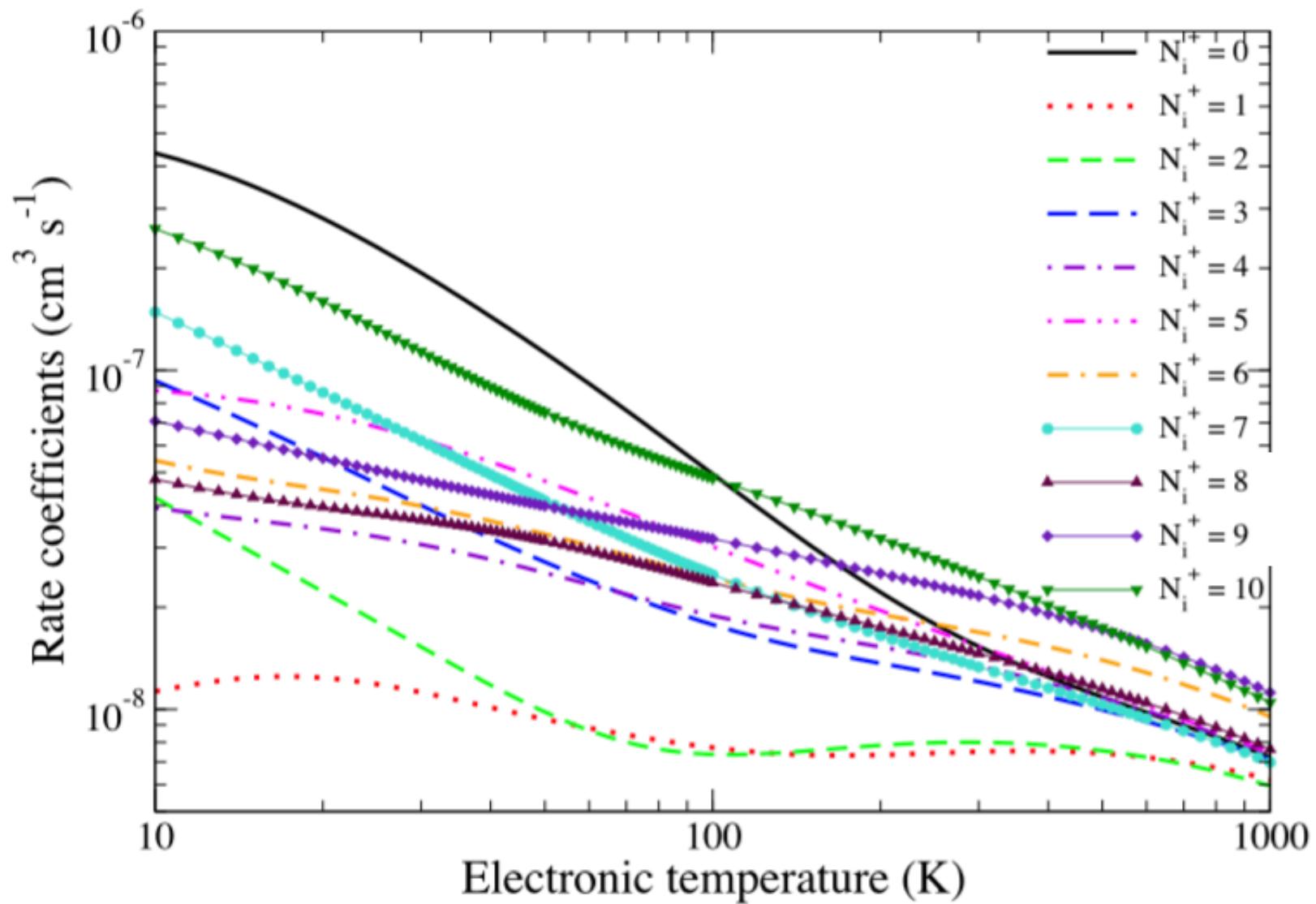


Figure 4. Maxwell rate coefficients for the DR of $\text{H}_2^+(X^2\Sigma_g^+)$ on its ground vibrational level $v_i^+ = 0$, as a function of its initial rotational level, N_i^+ , to 15

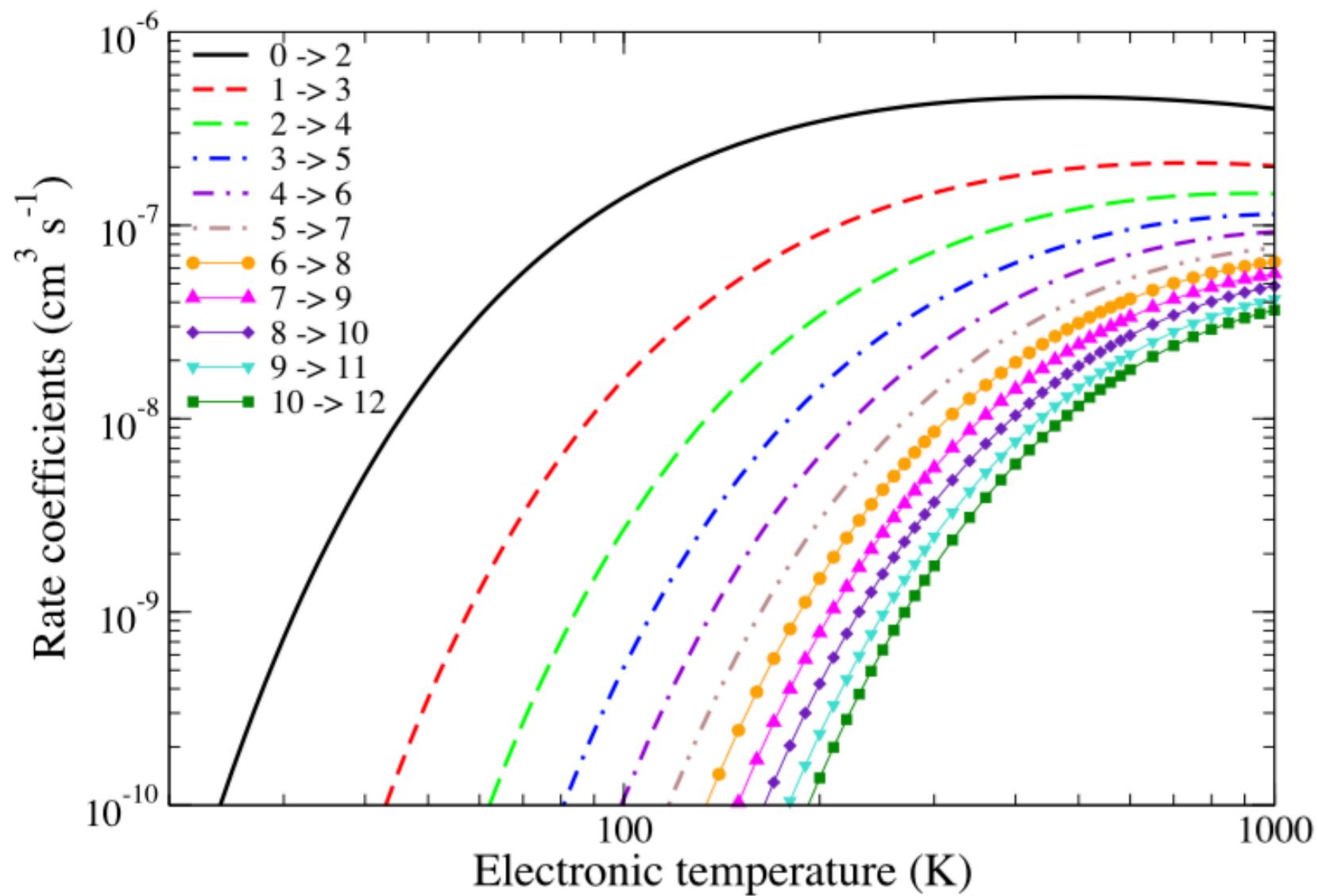
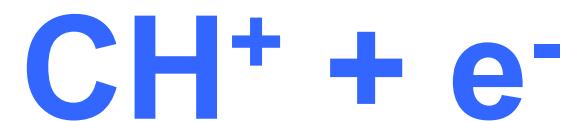
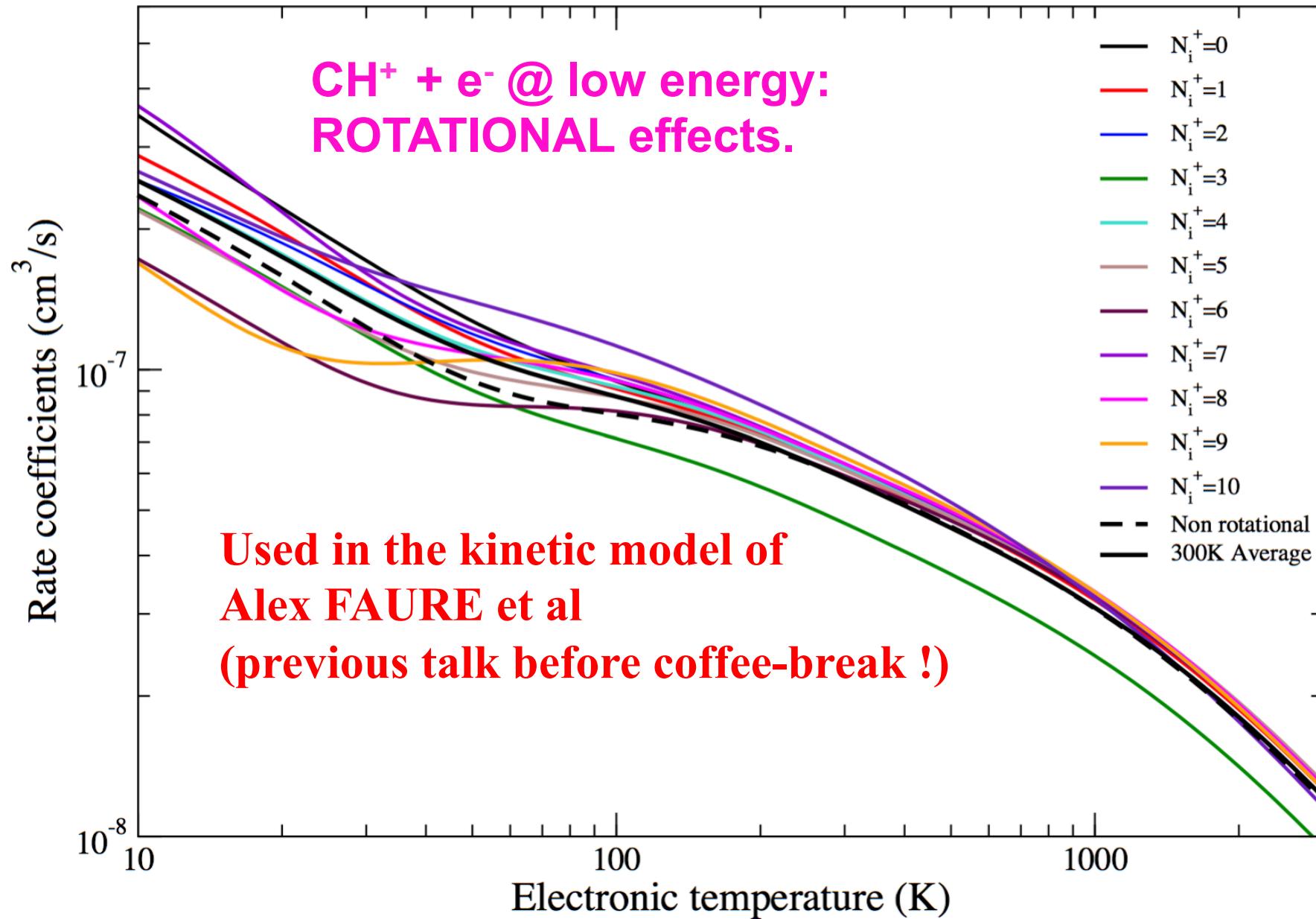


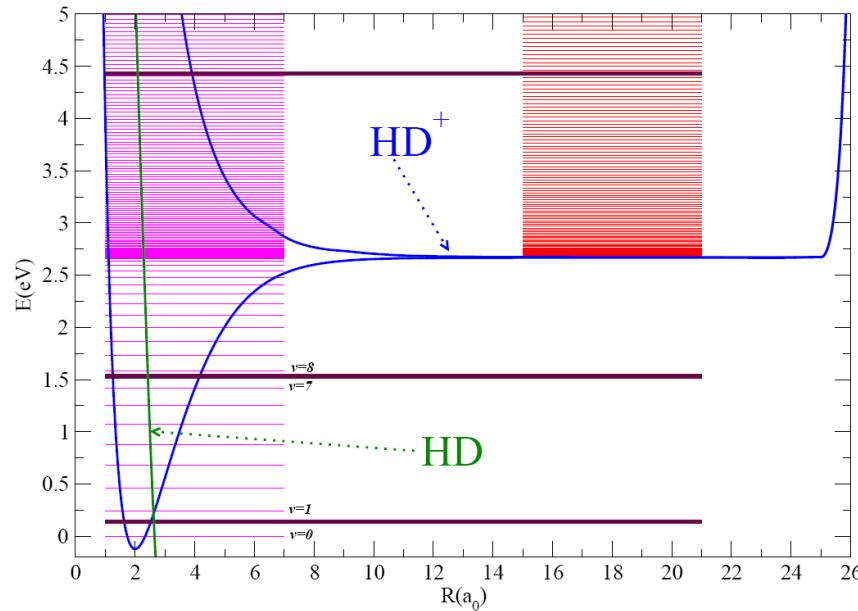
Figure 3. Maxwell rate coefficients for rotational excitation $N_i^+ \rightarrow N_i^+ + 2$, with $N_i^+ = 0$ to 10 of $\text{H}_2^+(X^2\Sigma_g^+)$ on its ground vibrational level $v_i^+ = 0$.¹⁶



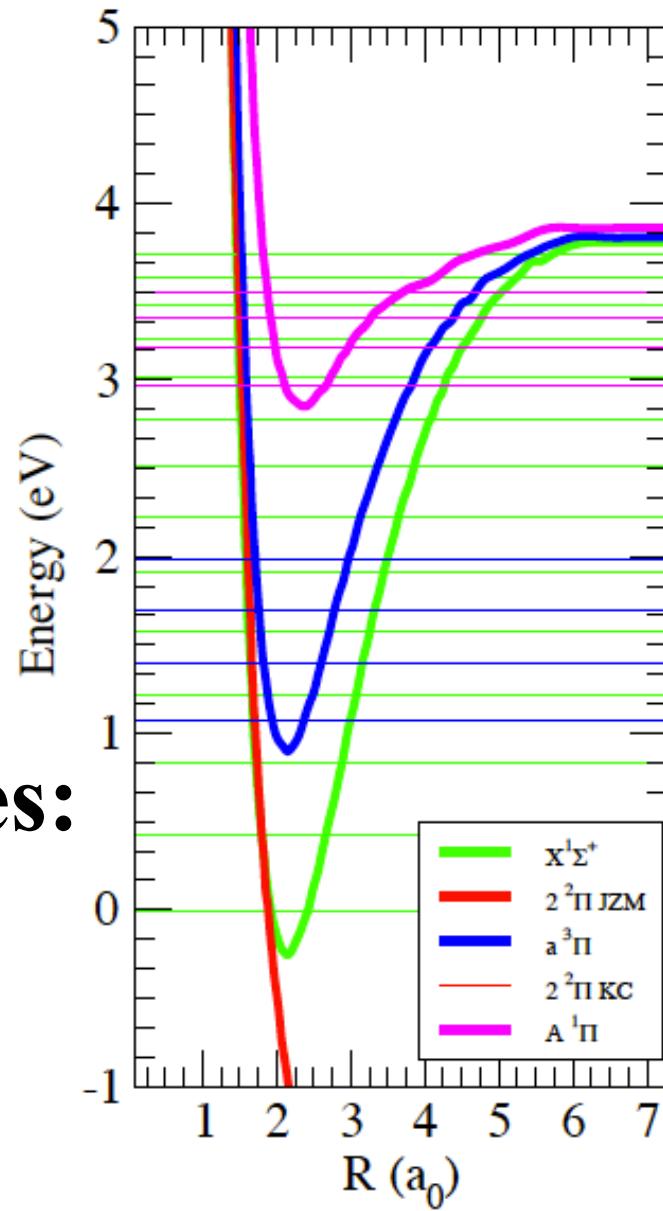


Used in the kinetic model of
Alex FAURE et al
(previous talk before coffee-break !)

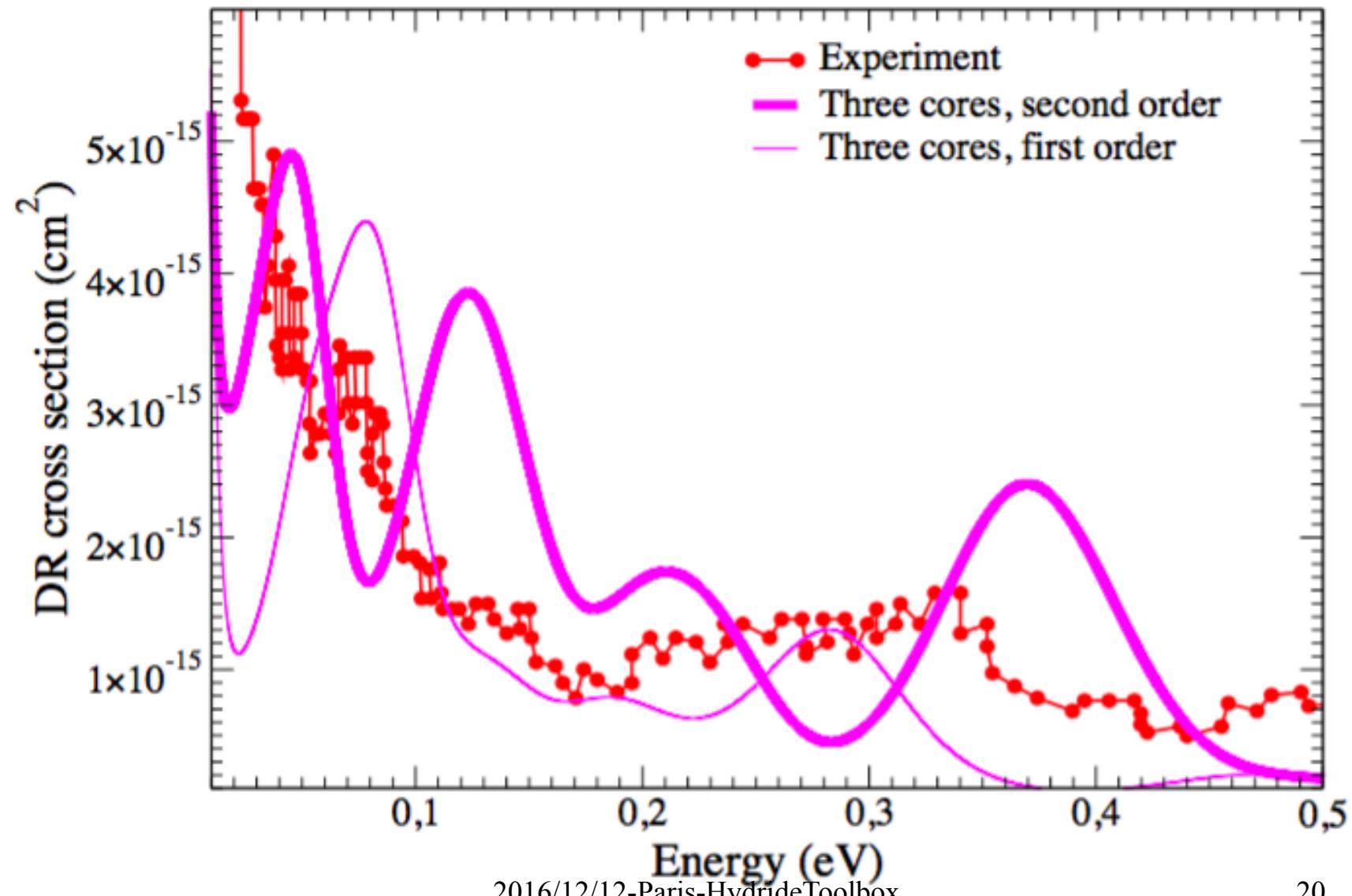
CH⁺ + e⁻ @ high energy

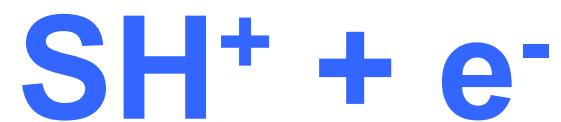


CH⁺: BOUND ionic cores:
 c1, c2, c3
 +
 electron



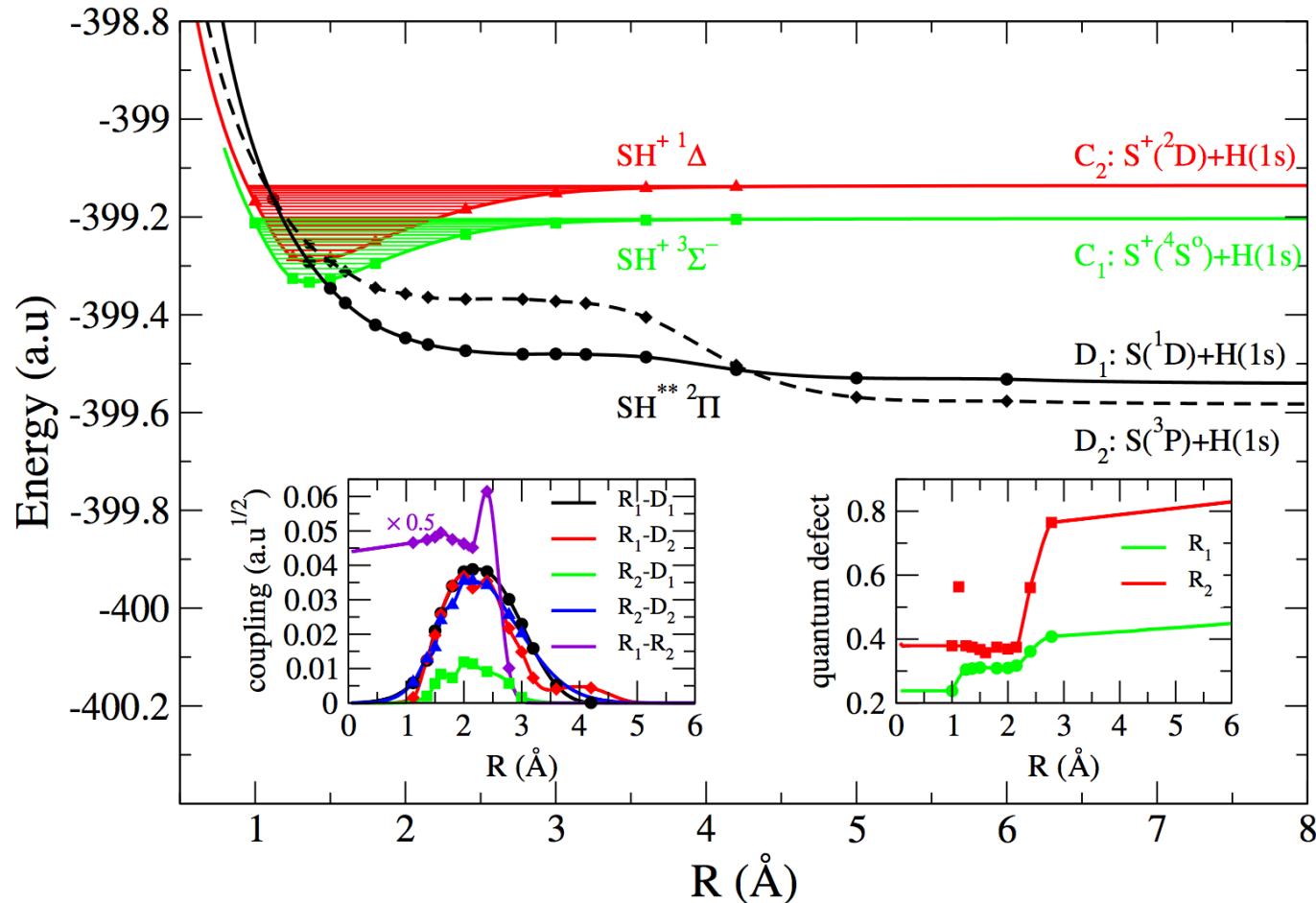
$\text{CH}^+ + \text{e}^-$ @ high energy: rotational effects neglected,
but account of the core-excited Rydberg states:



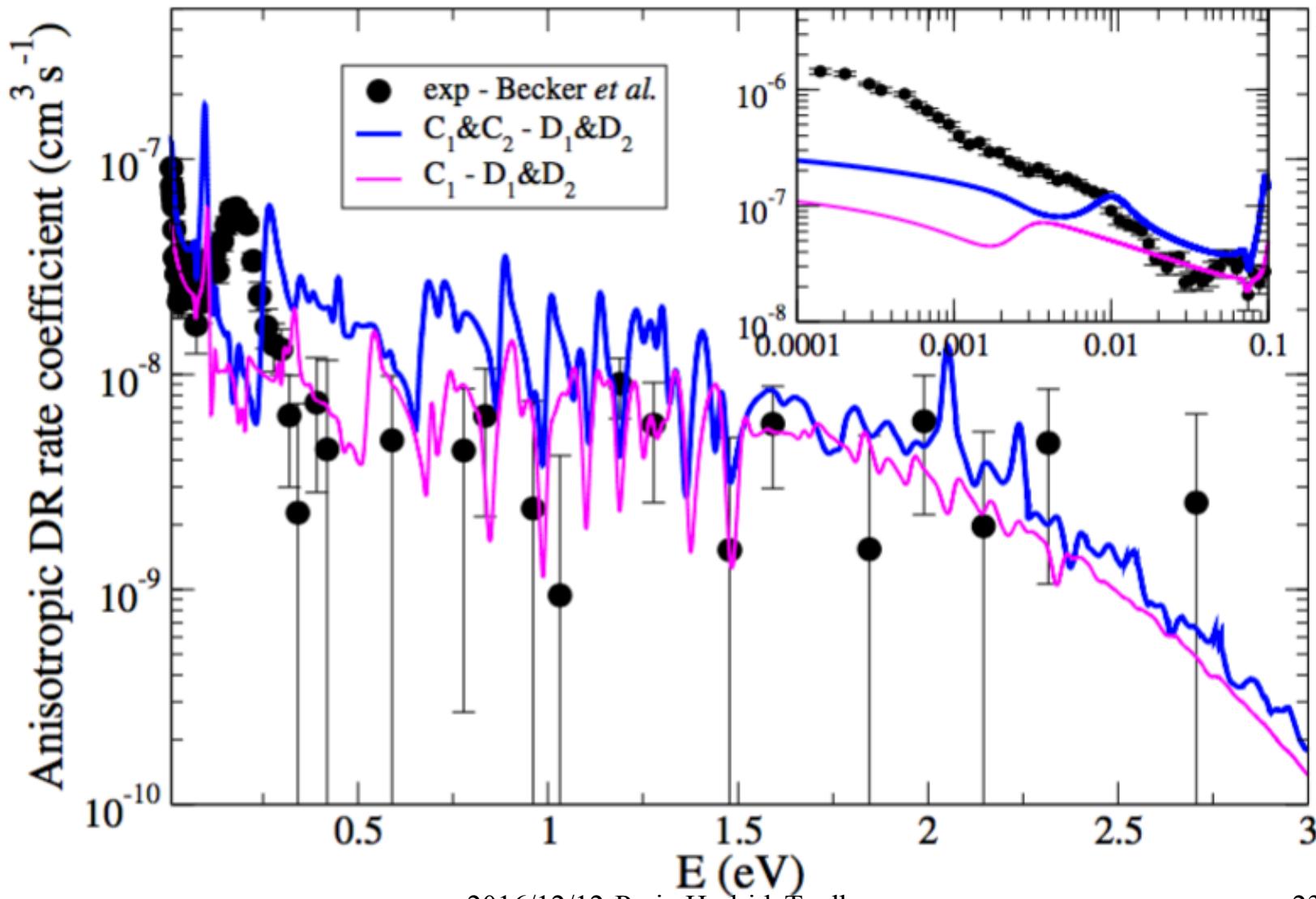


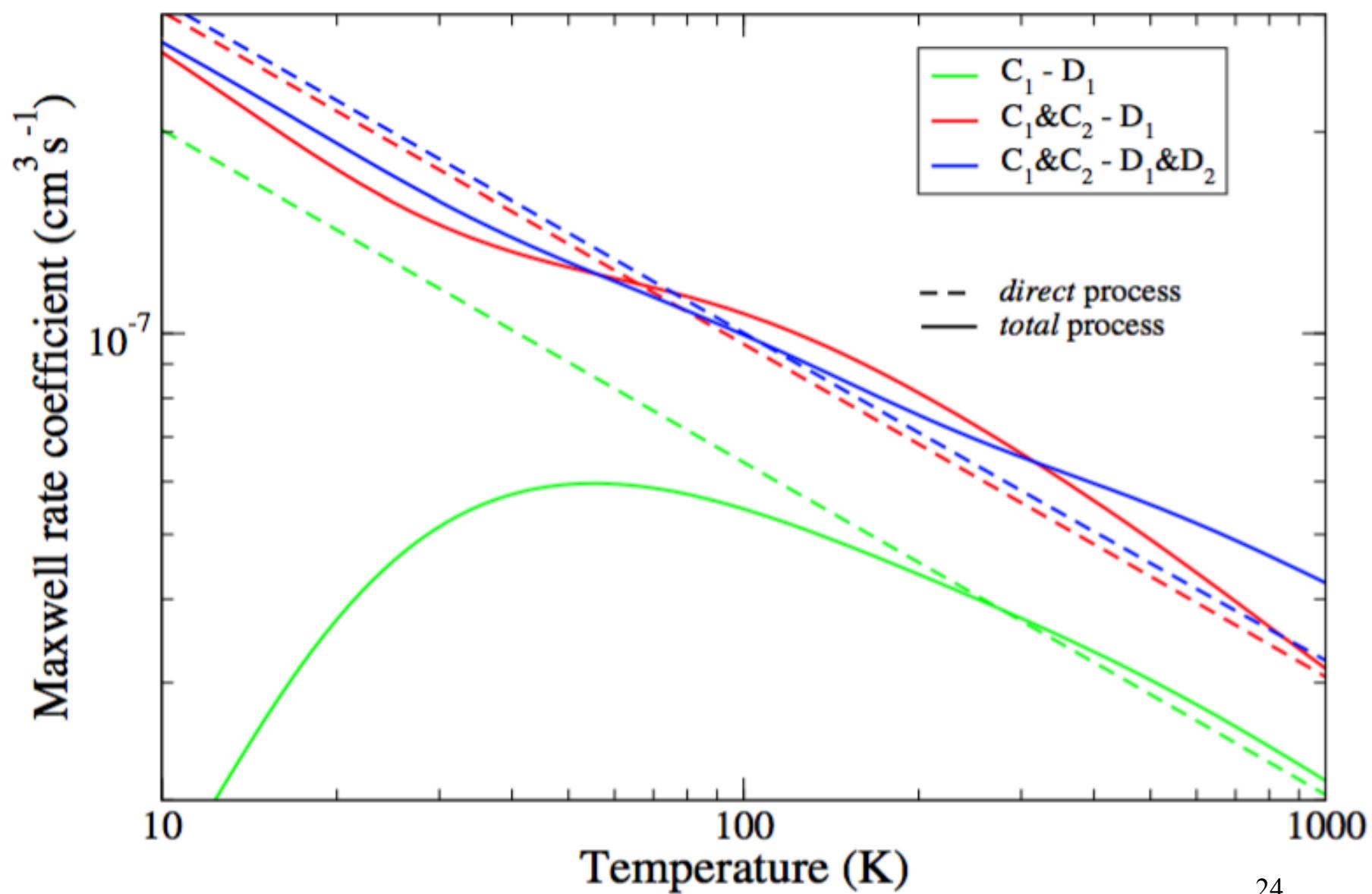
$\text{SH}^+ + \text{e}^-$: rotational effects neglected, account of the core-excited Rydberg states:

Bock-diagonalization method, Talbi et al 2015:



**SH⁺ + e⁻ : rotational effects neglected,
account of the core-excited Rydberg states:**





BeH⁺ + e⁻



Beryllium monohydride (BeH): Where we are now, after 86 years of spectroscopy

Nikesh S. Dattani

Due to its simplicity, BeH is expected to be present in astronomical contexts such as exoplanetary atmospheres, cool stars, and the interstellar medium [27], but in the context of astronomy, has only been found on our Sun, in the two studies described in [28, 29].

ExoMol line lists – I. The rovibrational spectrum of BeH, MgH and CaH in the $X^2\Sigma^+$ state

Benjamin Yadin, Thomas Veness, Pierandrea Conti, Christian Hill,
Sergei N. Yurchenko and Jonathan Tennyson[★]

Department of Physics and Astronomy, University College London, Gower Street, WC1E 6BT London

BeH is one of the simplest heteronuclear diatomic molecules, and hence it is a strong contender for being observed in contexts such as exoplanetary atmospheres, cool stars and the interstellar medium. However, there are only very few astrophysical records of BeH, for example, a detection of $A^2\Sigma^+ \rightarrow X^2\Sigma^+$ emission lines of BeH in the sunspot umbra spectra by Wöhl (1971) and Shanmugavel et al. (2008).

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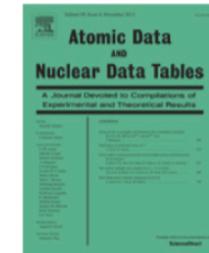
Atomic Data and Nuclear Data Tables ■ (■■■) ■■■-■■■



Contents lists available at [ScienceDirect](#)

Atomic Data and Nuclear Data Tables

journal homepage: www.elsevier.com/locate/adt



Low-energy collisions between electrons and BeH⁺: Cross sections and rate coefficients for all the vibrational states of the ion

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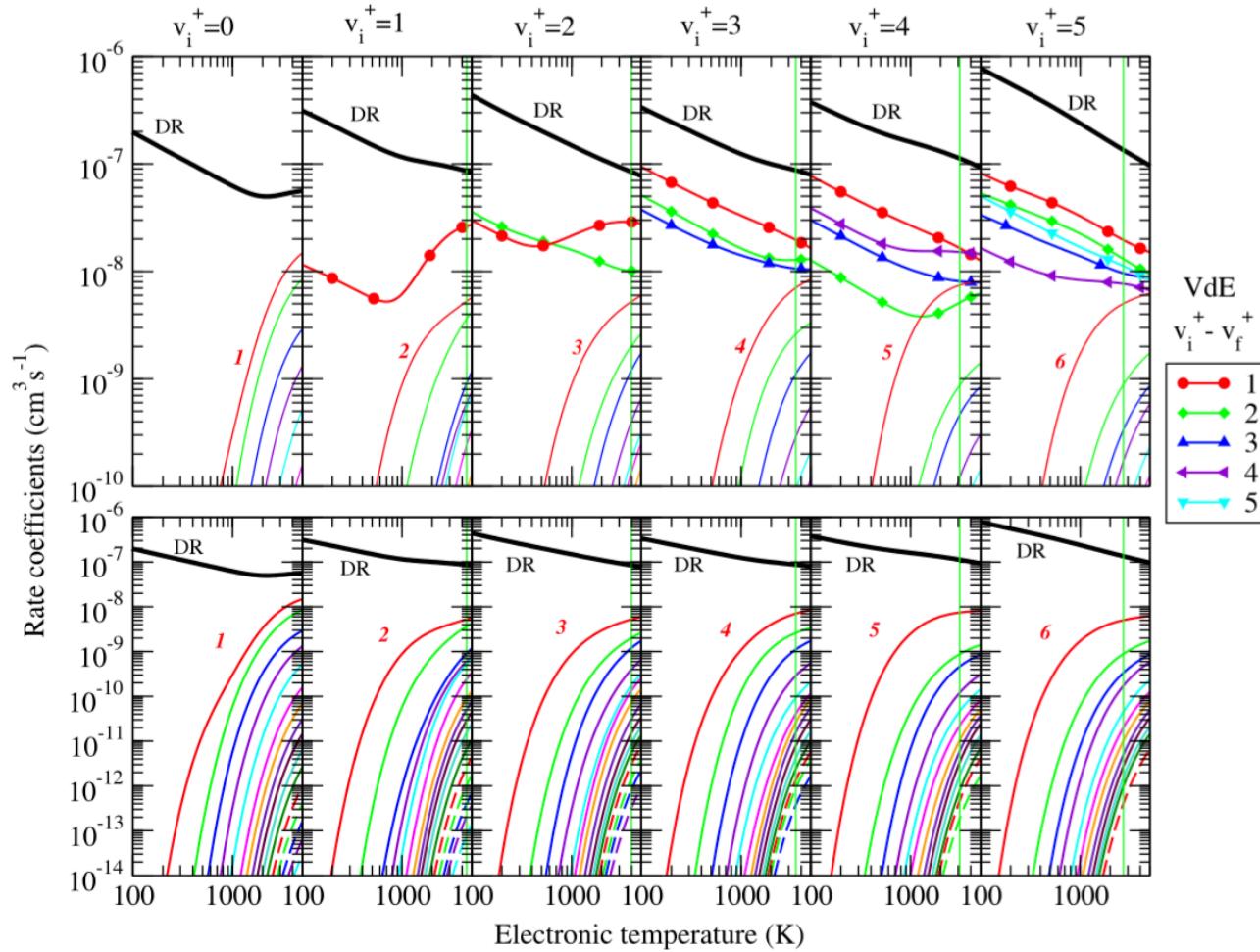
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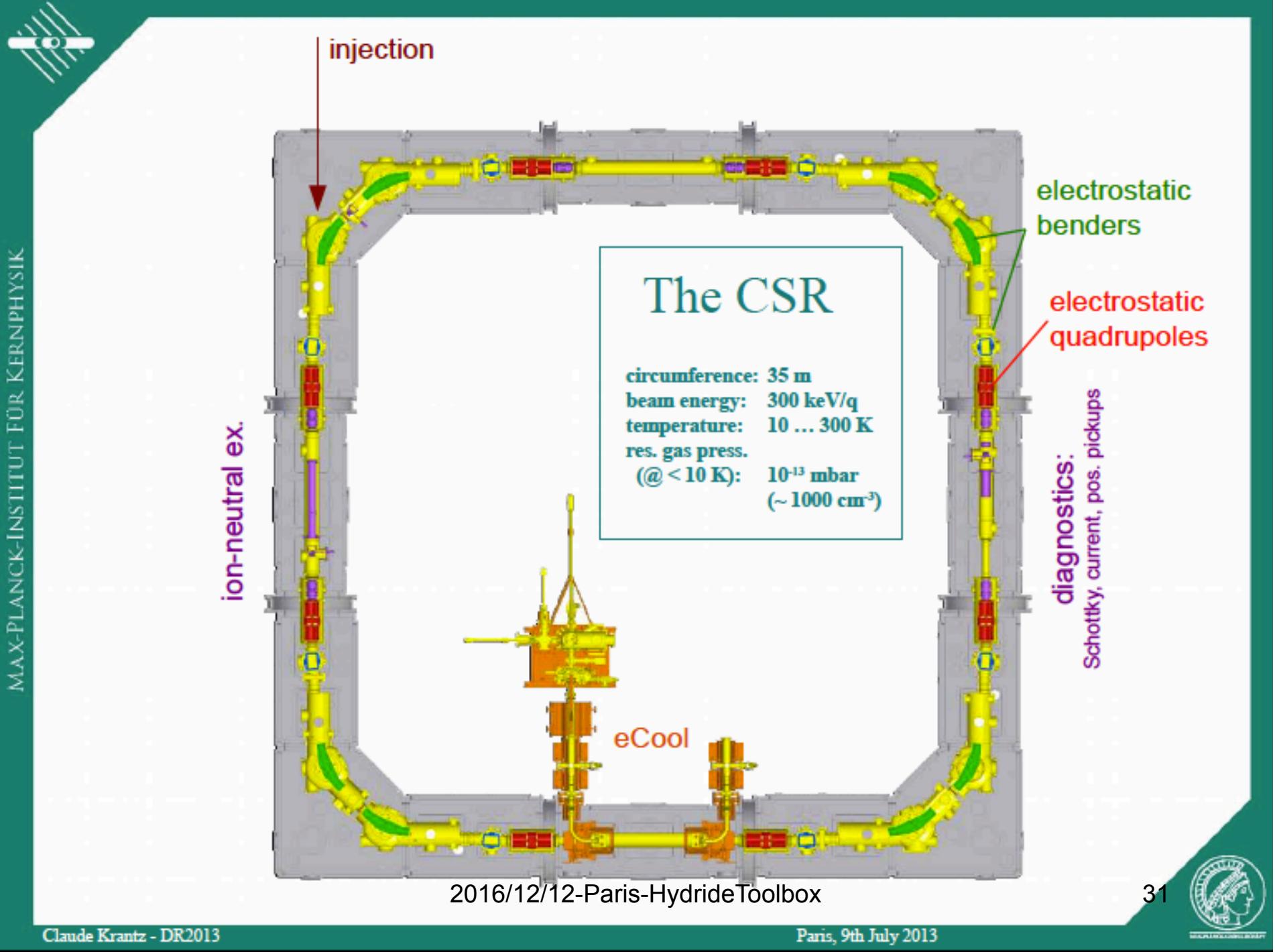
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^l Institute of Energy and Climate Research-Plasma Physics, Forschungszentrum Jülich GmbH Association EURATOM-FZJ, Partner in Trilateral Cluster, 52425 Jülich, Germany



Graph 4. Dissociative recombination (DR, thick line), vibrational excitation (VE, thin lines) and vibrational de-excitation (VdE, symbols and thick lines) Maxwell rate coefficients of ground ($v_i^+ = 0$) and excited ($v_i^+ = 1, \dots, 5$) BeH^+ in its electronic ground state (total mechanism). For VE, since the rate coefficients decrease monotonically with the excitation, the lowest final vibrational quantum number of the target is indicated only, and the lower panels extend the range down to $10^{-14} \text{ cm}^3/\text{s}$.

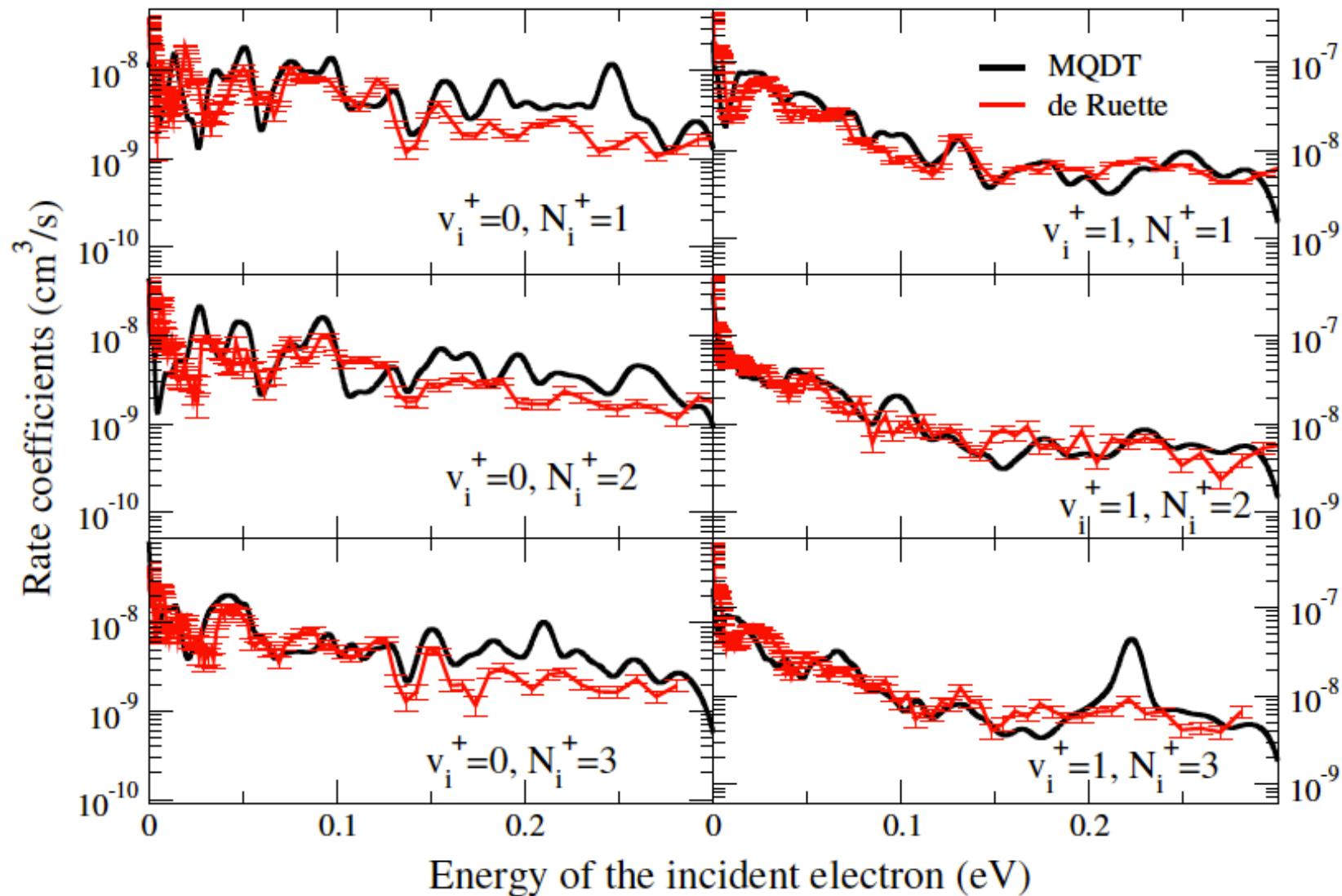
**And ...
the FUTURE**



Are we prepared ?



1st state-to-state comparison experiment/theory ever !



Perspectives:

$\text{OH}^+ + \text{e}^-$

$\text{HCl}^+ + \text{e}^-$

$\text{ArH}^+ + \text{e}^-$

Polyatomic⁺ + e⁻

...



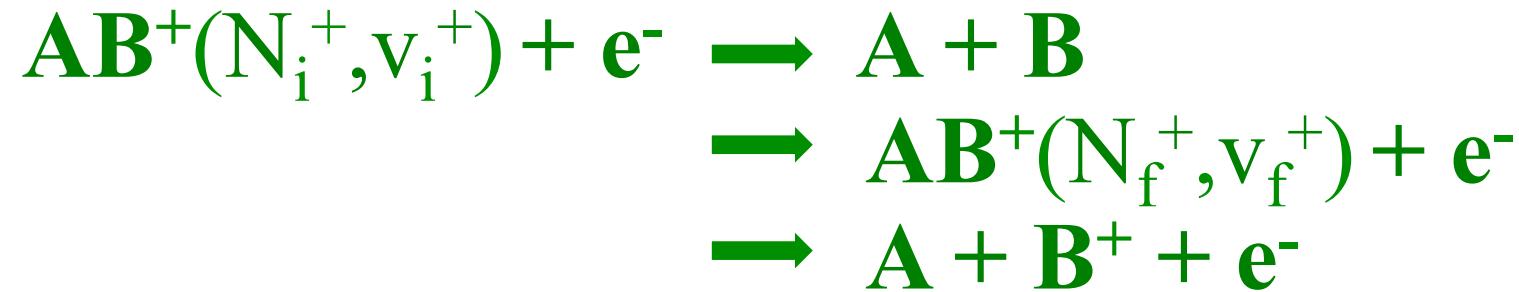
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Thank you for
your attention !