

# Physical and Chemical Evolution of Hydrides in Protoplanetary Disks

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Pontoppidan, Geoff Blake*

The Hydride Toolbox,  
Universite Pierre et Marie Curie  
Paris, France December 15, 2016



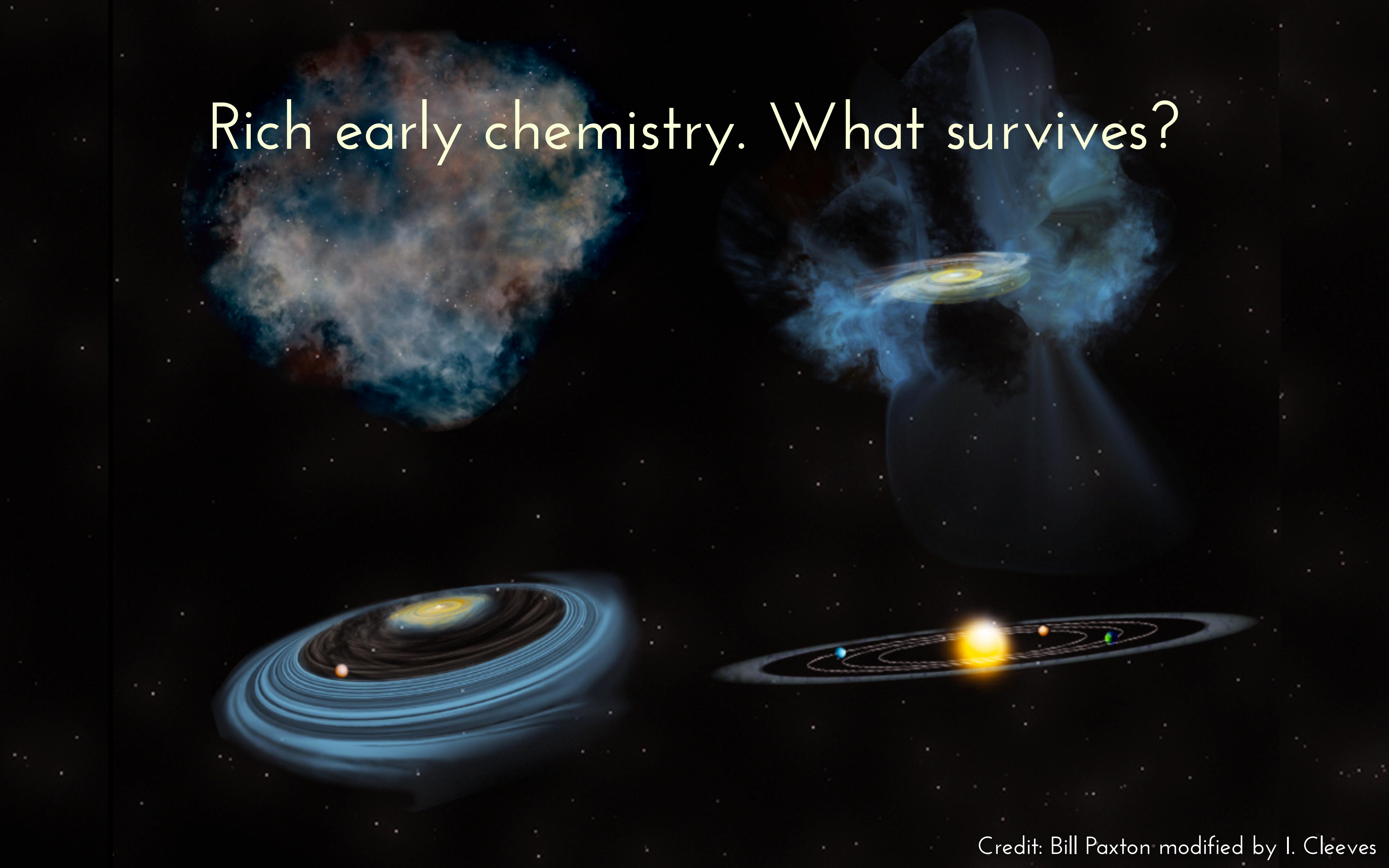
## Hydrides are a starting point of prebiotic chemistry

Laboratory irradiation of “astrophysically large” molecules in a NH<sub>3</sub> and H<sub>2</sub>O mixture has been shown to form biologically interesting molecules including cytosine, urea, and glycine (e.g., Nuevo, Milam et al. 2013).



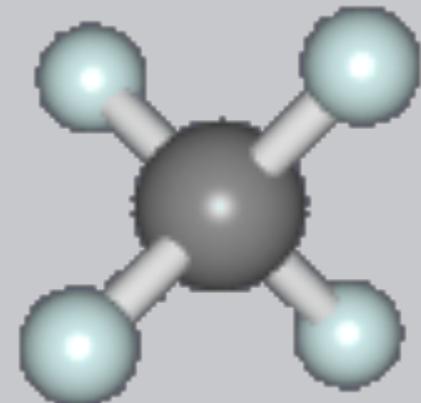
## Hydrides provide insight into key question of inheritance vs. reprocessing during star formation

High binding energies result in key hydrides (NH<sub>3</sub> and H<sub>2</sub>O) being largely in the ice phase under dense ISM conditions. Sensitive to extreme energetic events, like forming a star!

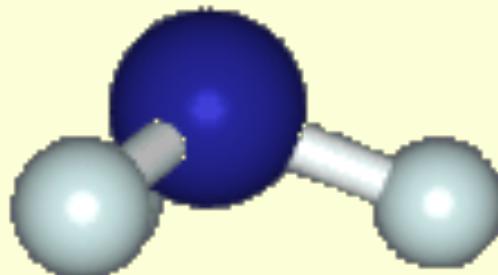
The background of the slide features three distinct astronomical illustrations. In the upper left, a large, colorful nebula with shades of blue, orange, and red is shown against a dark space background with numerous small stars. In the upper right, a black hole is depicted with a bright, glowing accretion disk surrounding it, and a stream of matter falling into its event horizon. In the lower center, a solar system is shown with a yellow sun at the center, five planets (Mercury, Venus, Earth, Mars, and Jupiter) in their respective orbits, and several small, distant stars scattered across the dark void.

Rich early chemistry. What survives?

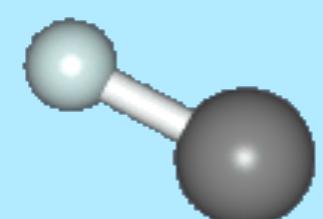
# Detected Disk Hydrides



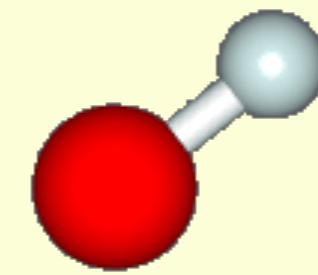
One source, GV  
Tau N (Gibb et al.  
2013)



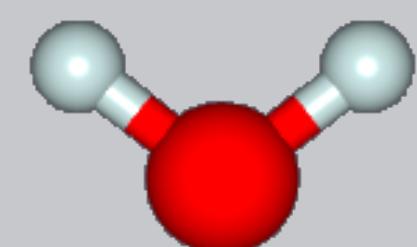
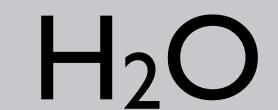
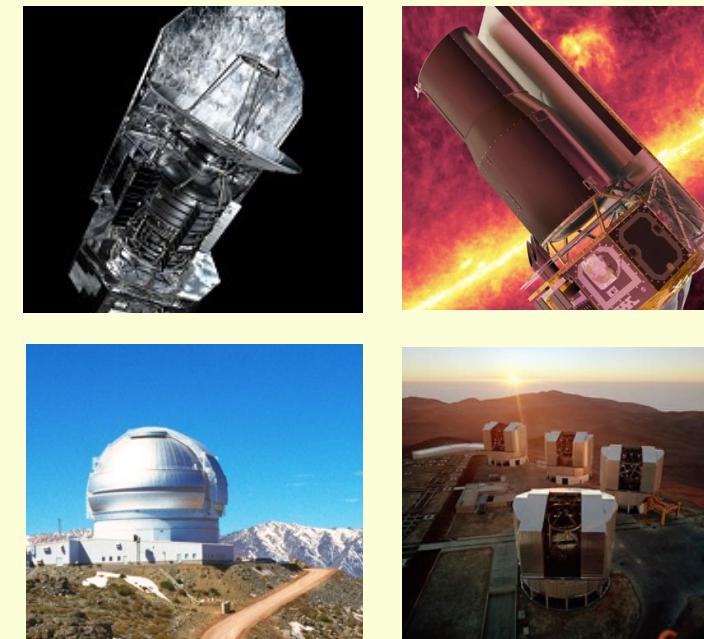
One source, TW  
Hya (Salinas et al.  
2016)



One source, HD  
100546 (Thi et al.  
2011)



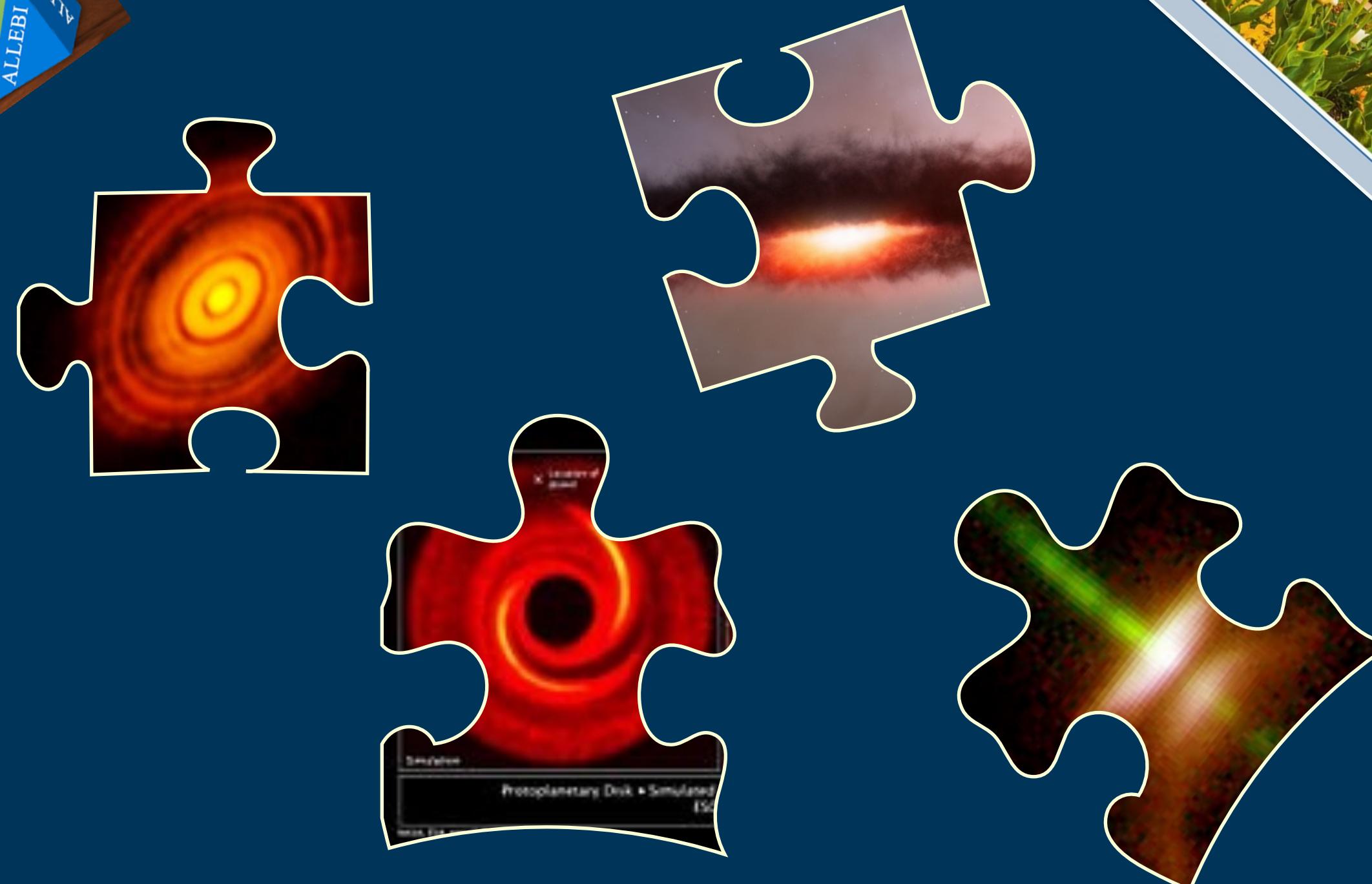
*Herschel:* Sturm et al.  
2010, Fedele et al. 2012,  
Bergin, et al. 2013,  
*Spitzer:* e.g., Pontoppidan  
et al. 2010  
*Ground:* Mandel+08,  
Lisowsky+12



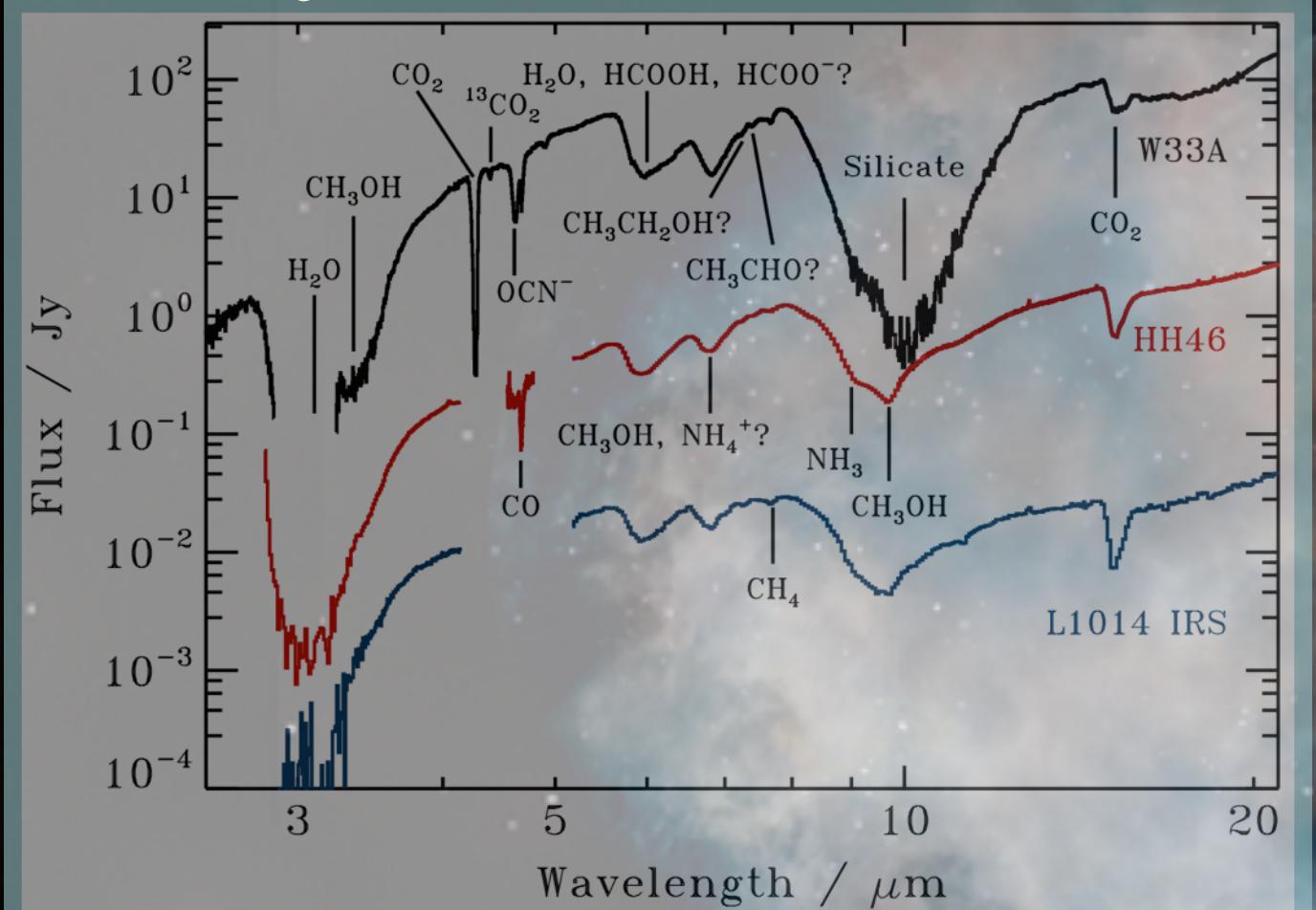
Pontoppidan+10, +11,  
Bergin+10, Sturm+10,  
Hogerheijde+11, Du+16,  
Fedele+12, Banzatti+16  
Hot H<sub>2</sub>O: Sample ~ 120  
disks, 60-80% detection  
Cold (<100 K) H<sub>2</sub>O: ~15%



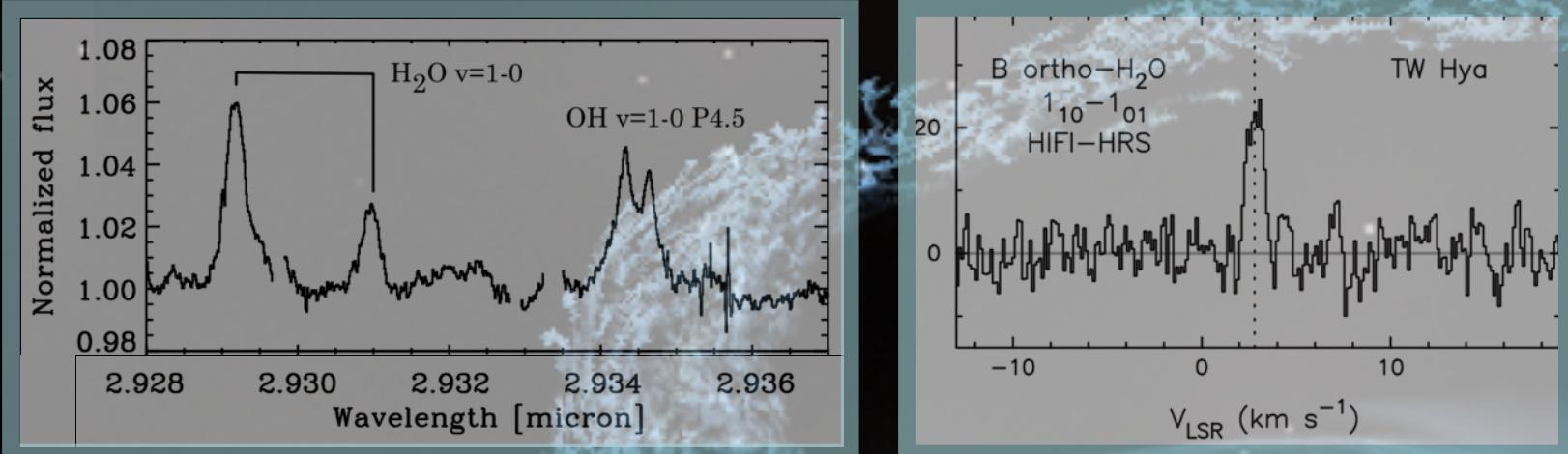
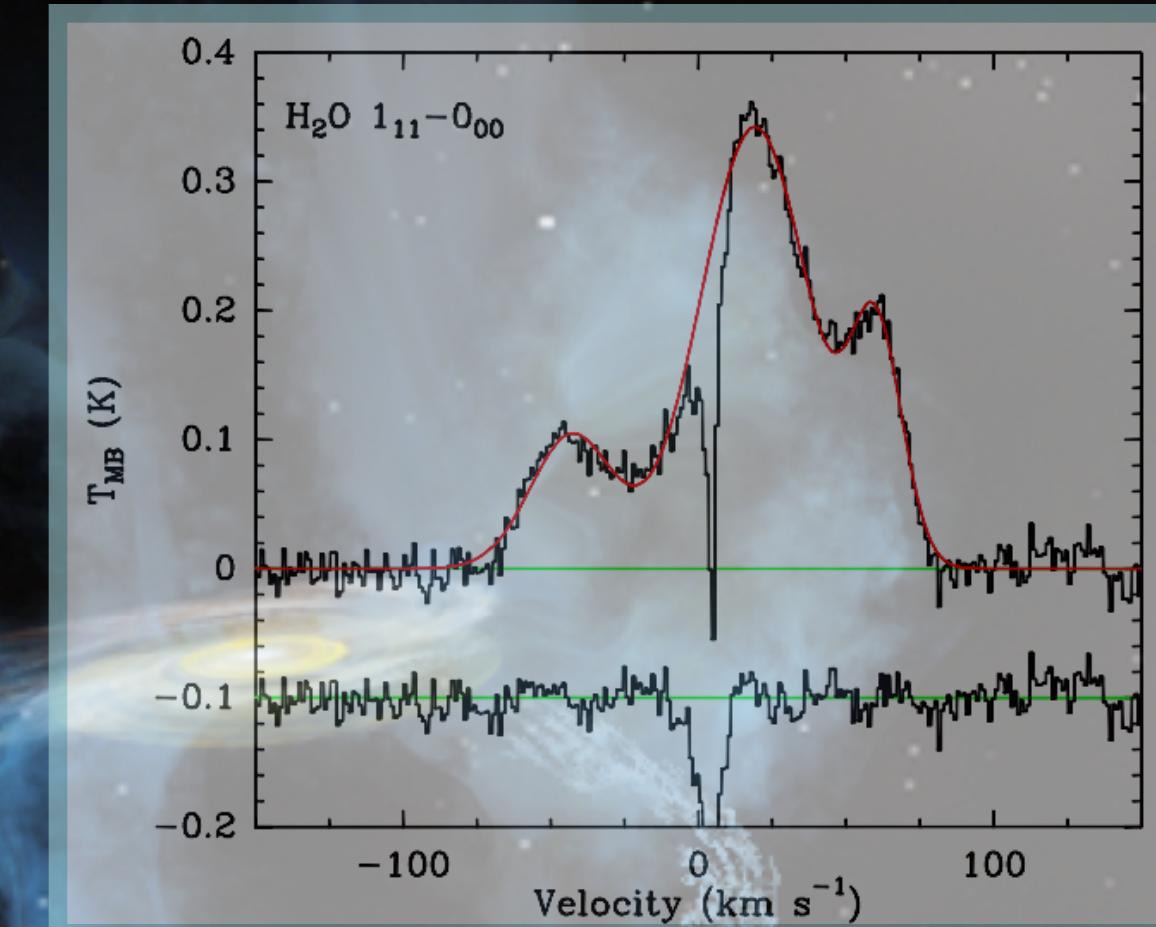
Measuring pieces of the puzzle, but  
from all different boxes.



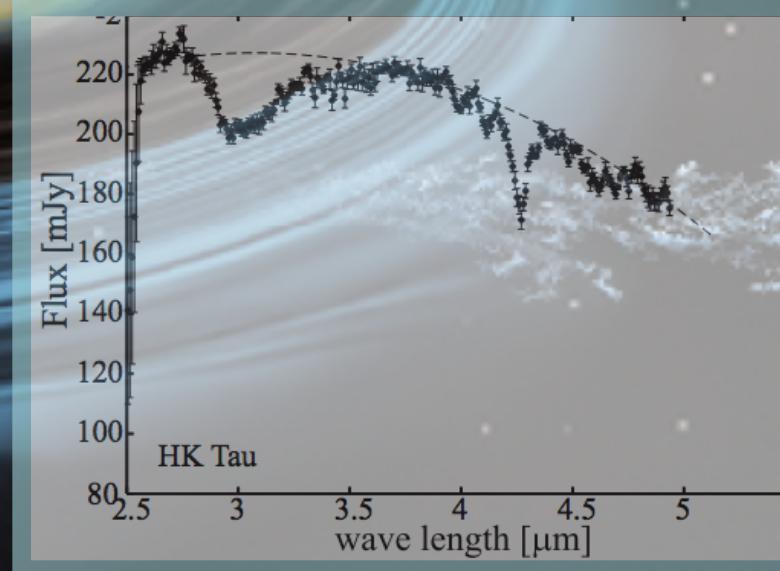
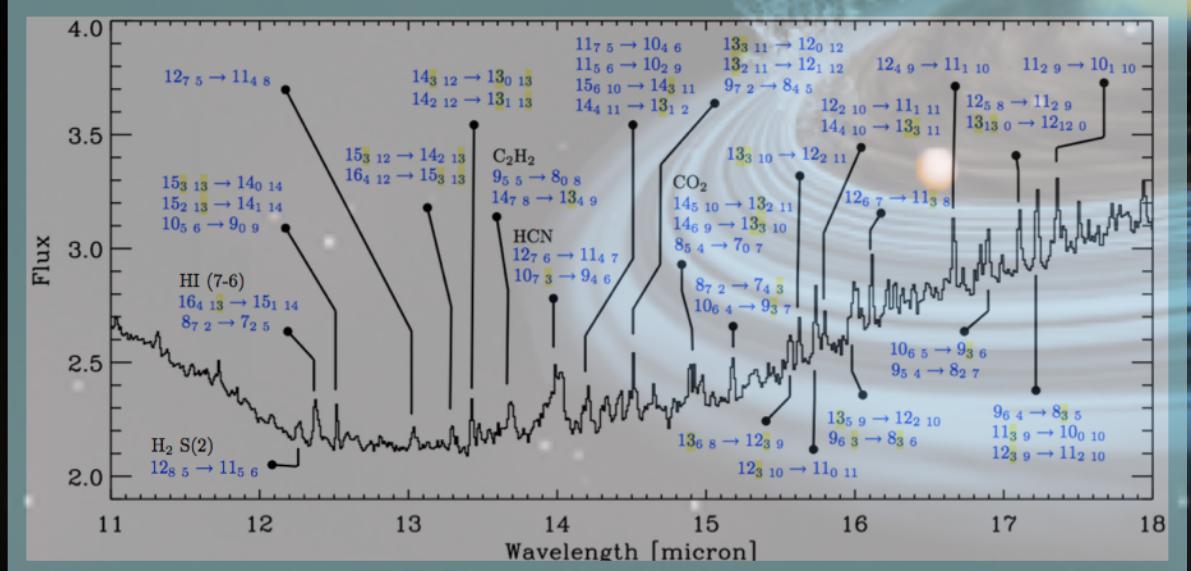
Ice: Öberg et al 2011, Gas: Caselli et al. 2012



Kristensen et al. 2011 (and many others)

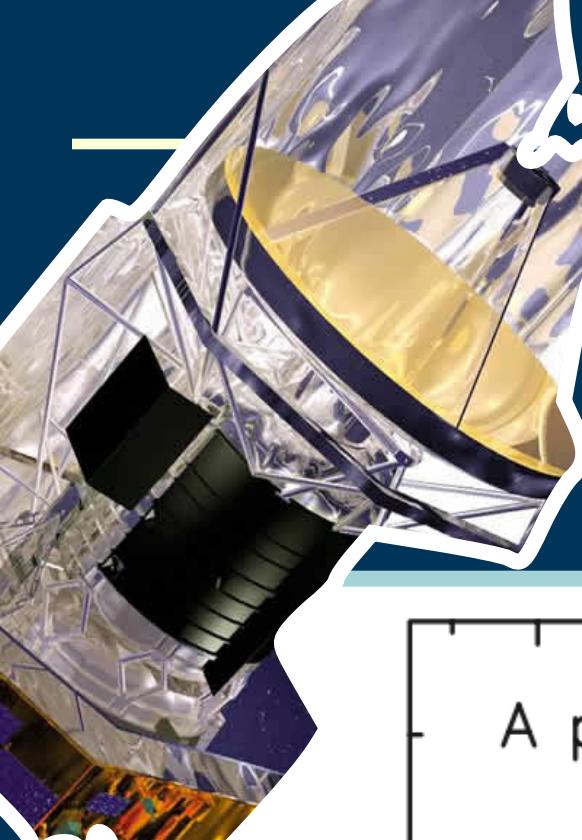


Pontoppidan et al. 10, 11, Hogerheijde et al. 11, Aikawa et al. 12

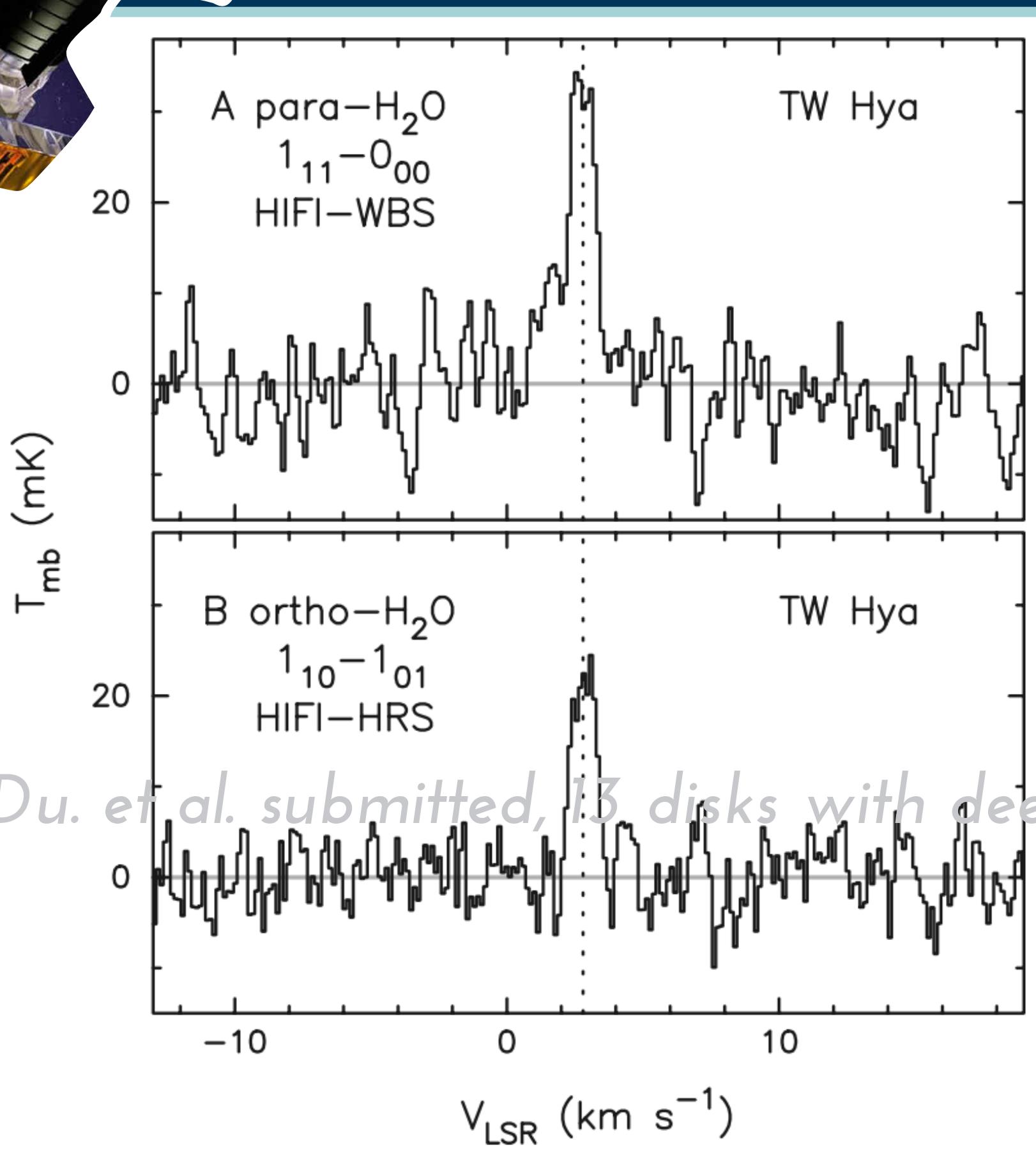


Biver et al. 2015



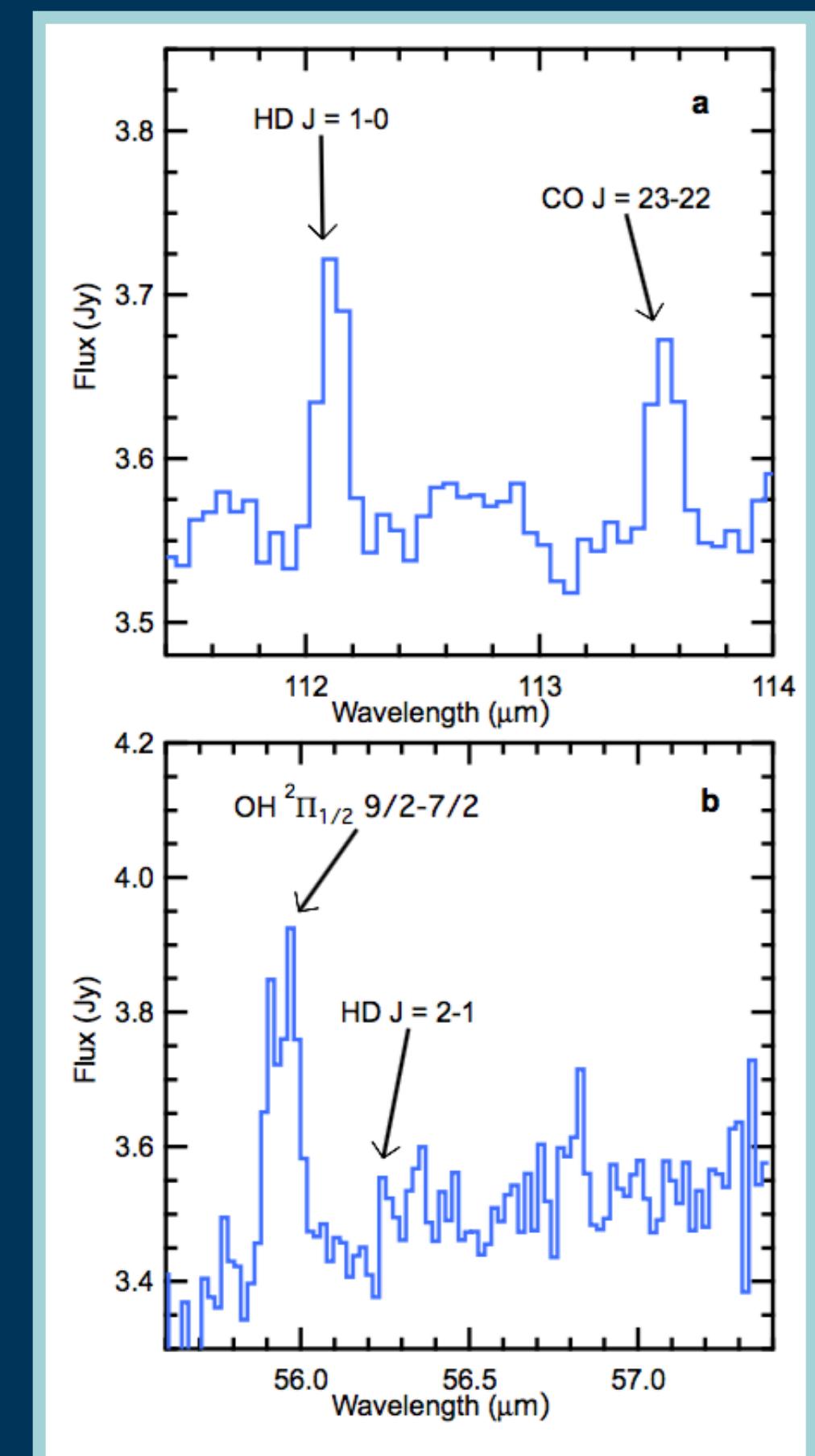


# Cold Water in Protoplanetary Disks



Cold water detected  
with *Herschel* (and  
opr), Hogerheijde et al.  
2011.

Also Du et al. submitted, 13 disks with deep integrations  
Herschel, Bergin et al.  
2013. Mass  
unambiguous, water  
depleted.



# Cold Water in Protoplanetary Disks

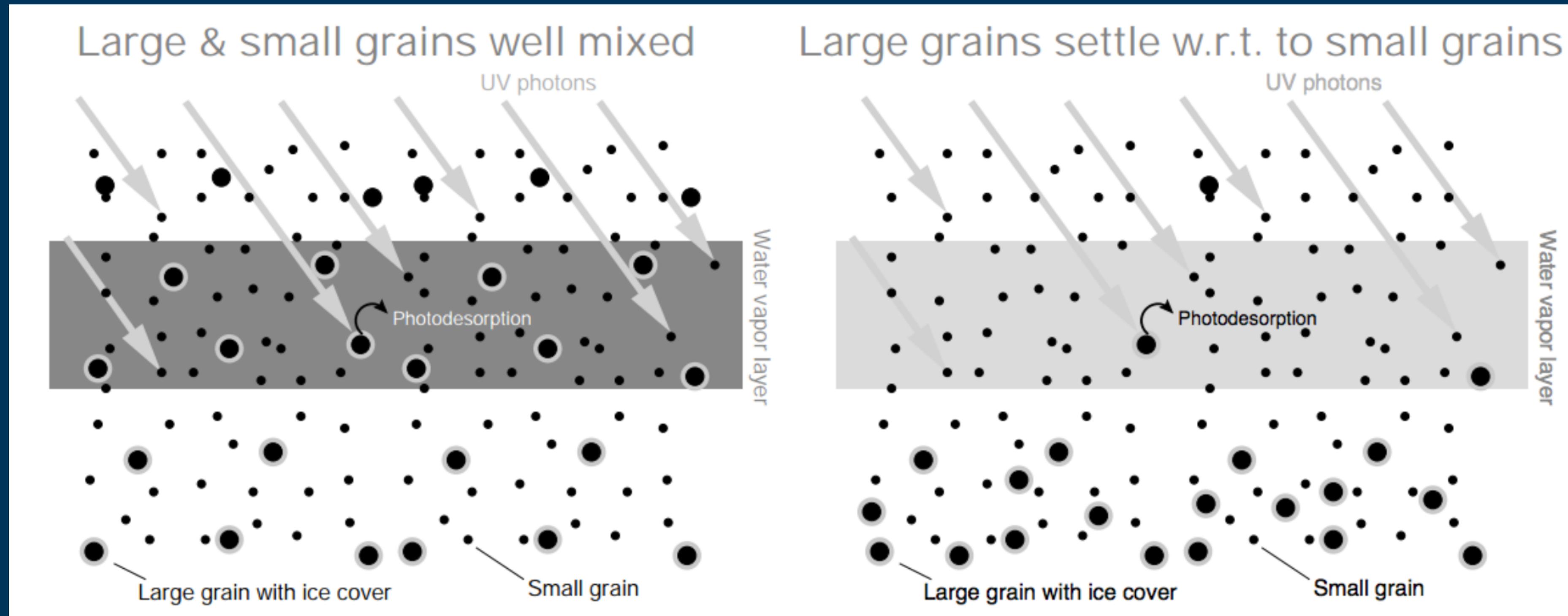


Fig. courtesy M. Hogerheijde

# Outline: Cold Water Evolution In Disks

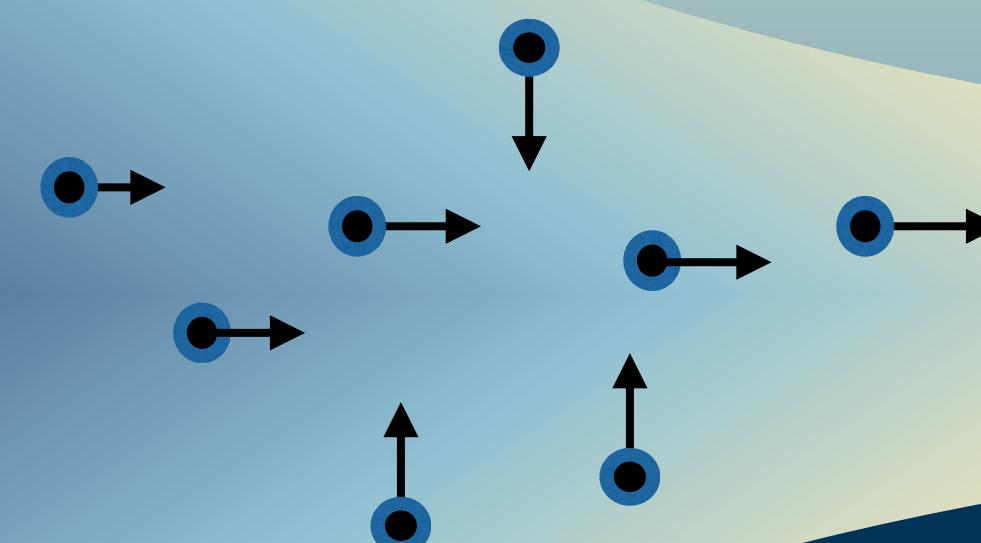
I. Primordial material?



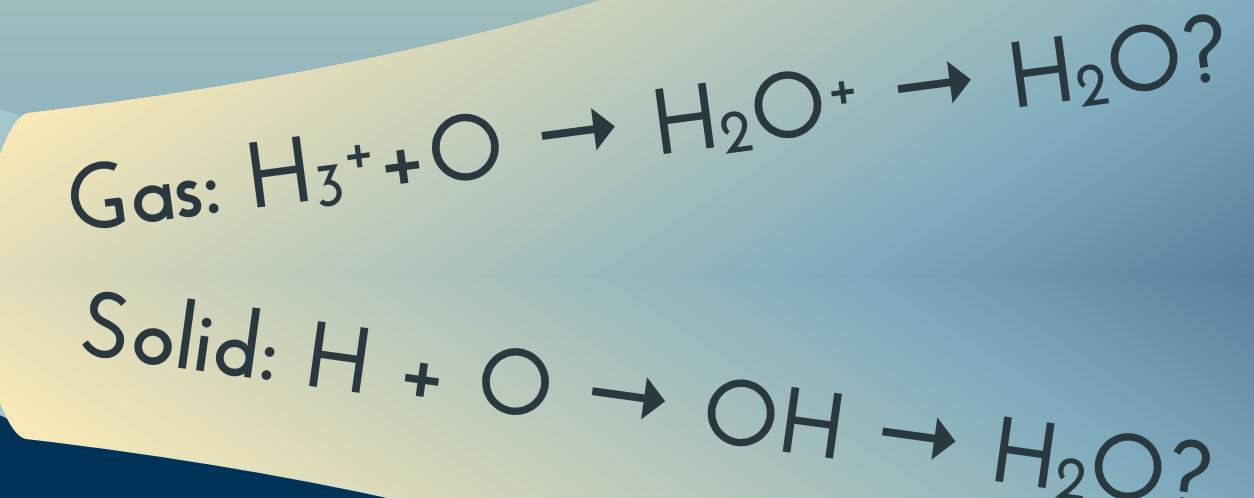
II. Kinematics?



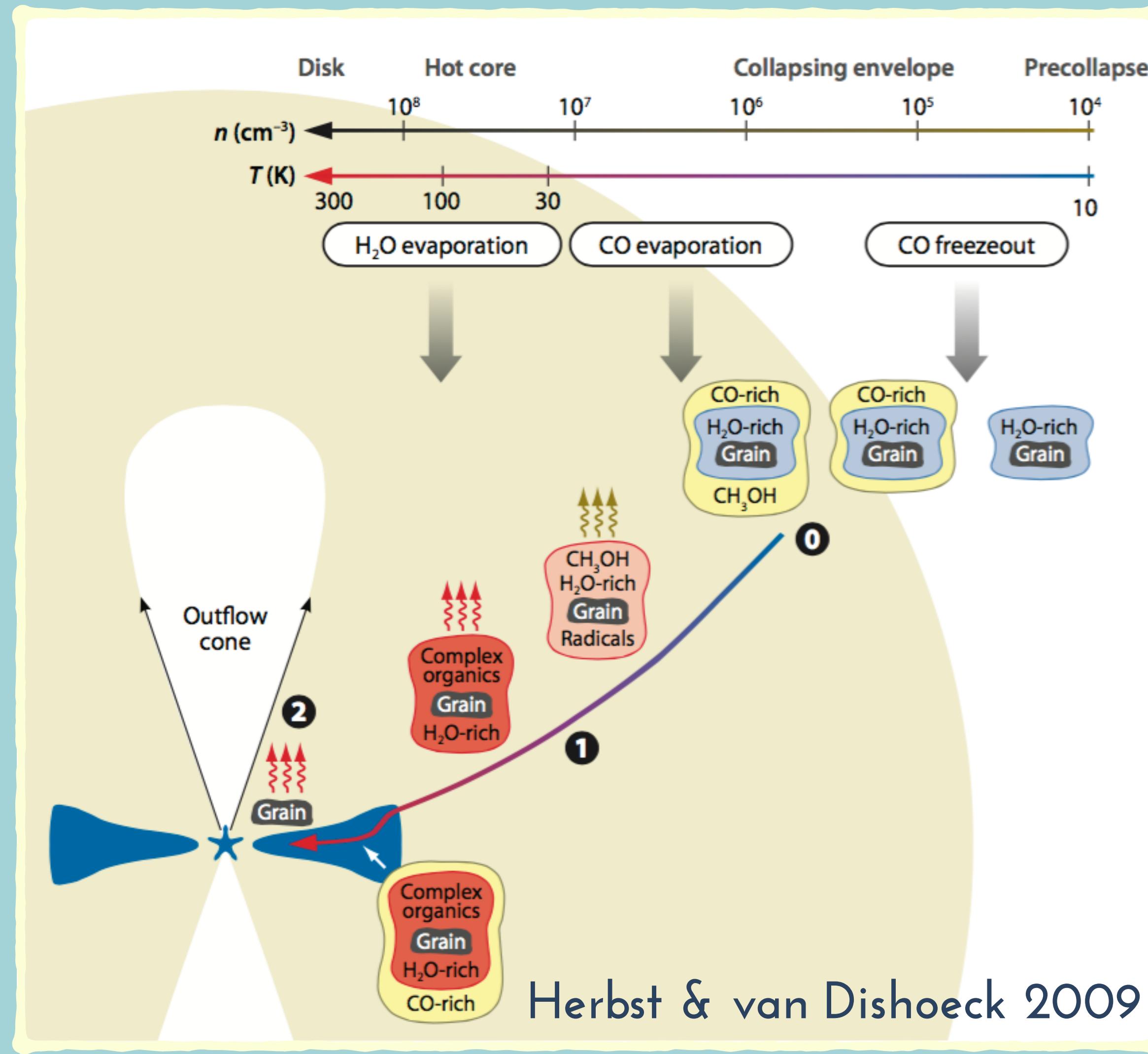
III. Aerodynamics?



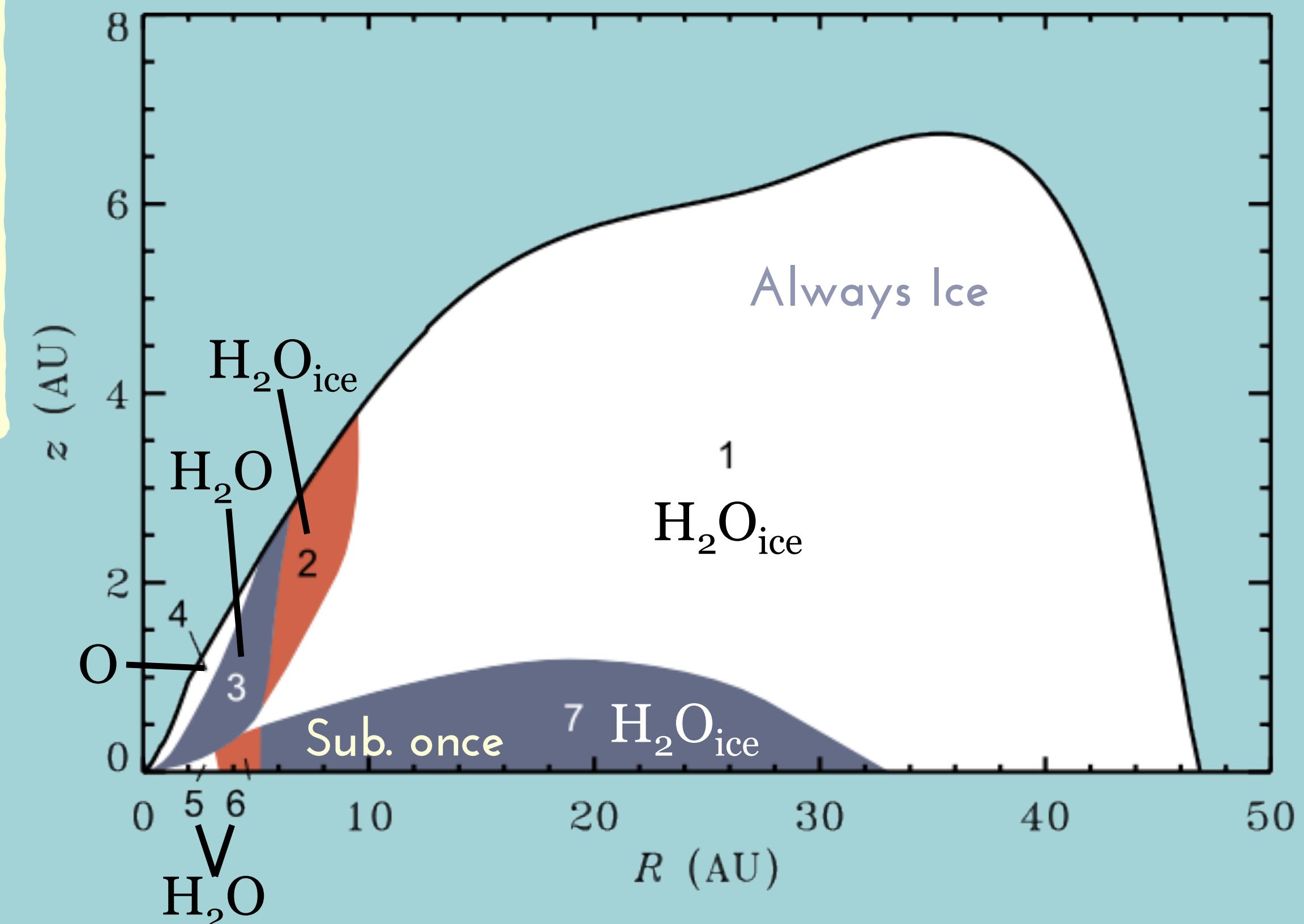
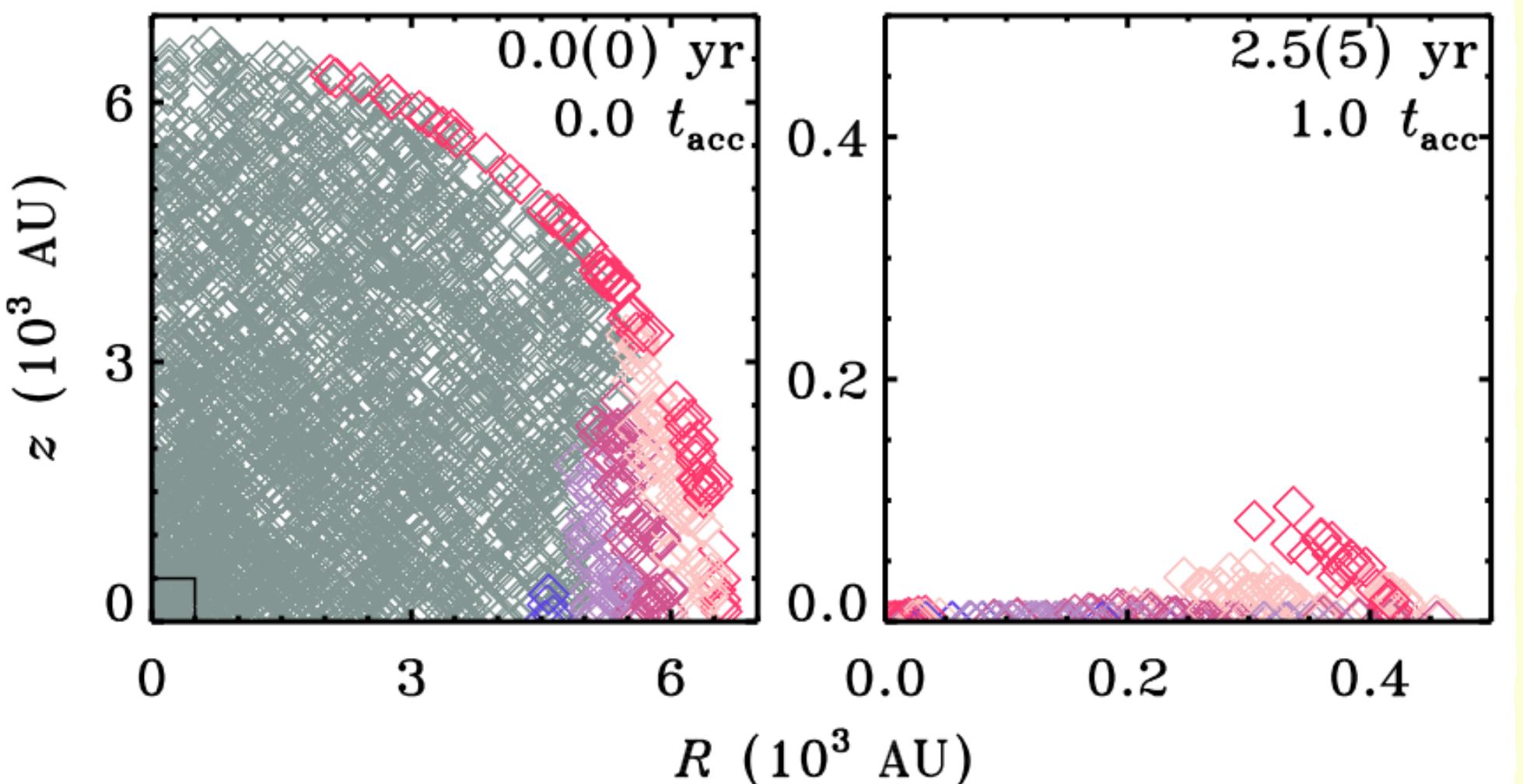
IV. Disk chemistry?



# I) What are the initial disk chemical conditions?



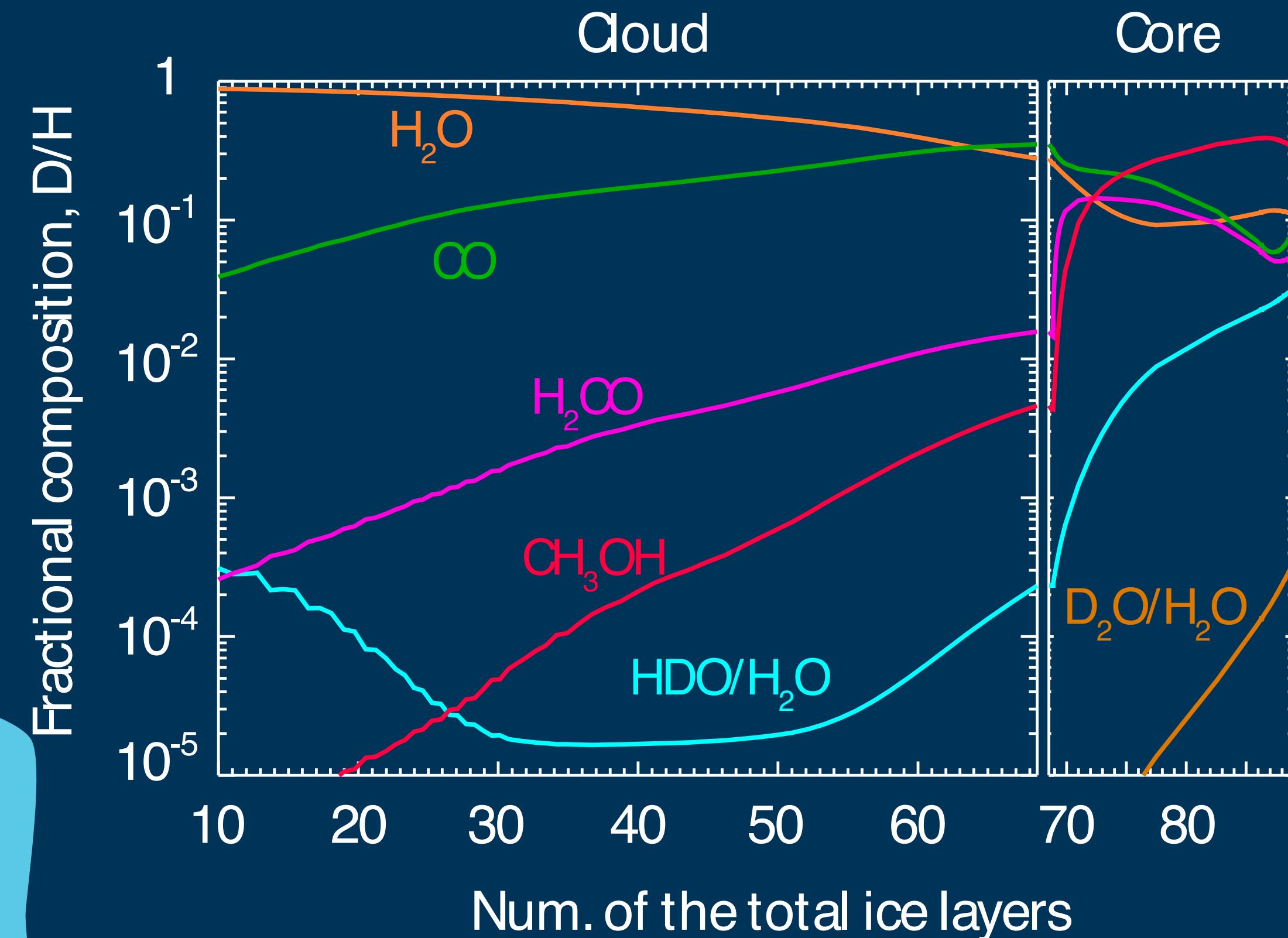
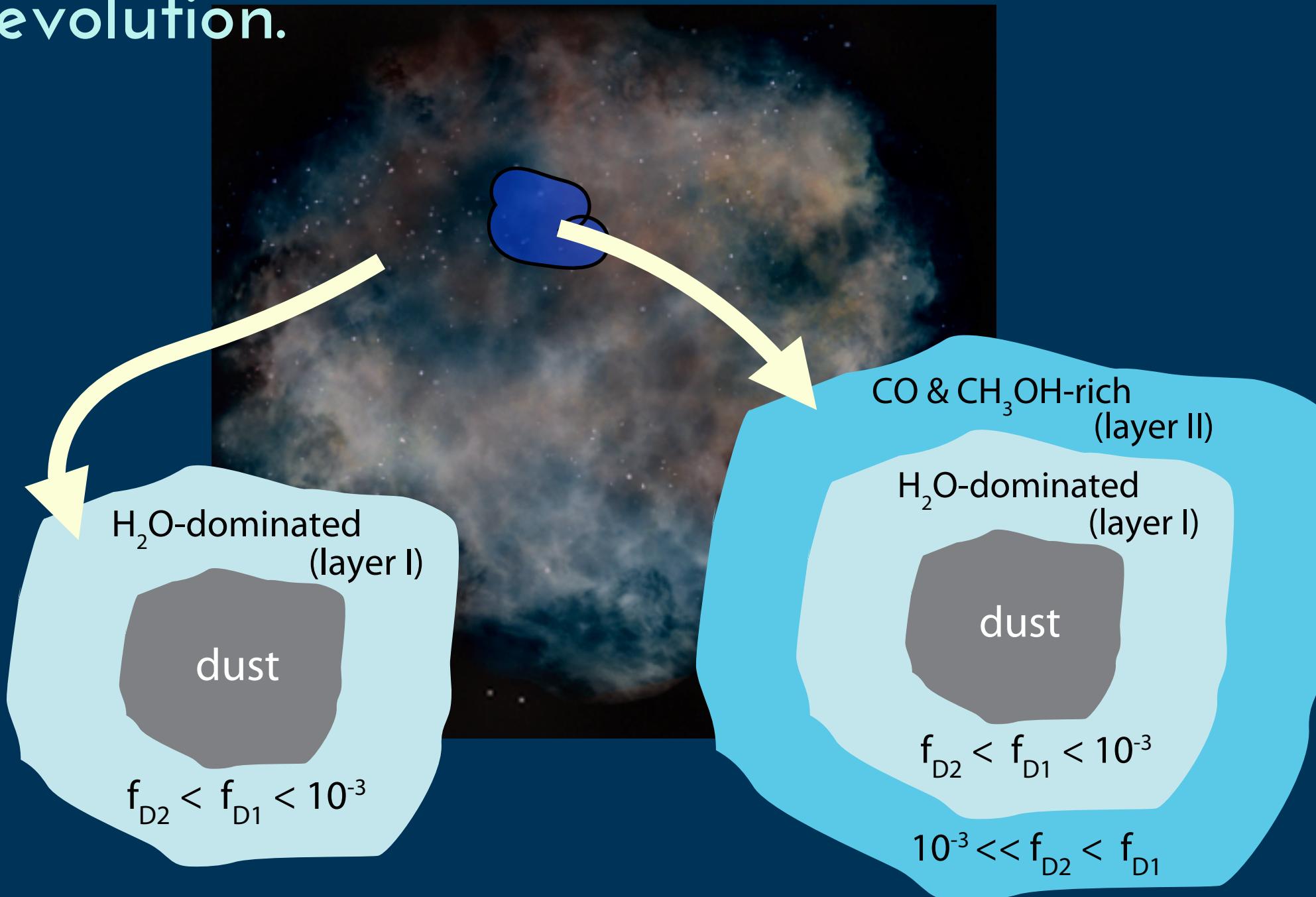
# I) What are the initial disk chemical conditions?



Visser, van Dishoeck, Doty  
& Dullemond 2009

# I) What are the initial disk chemical conditions?

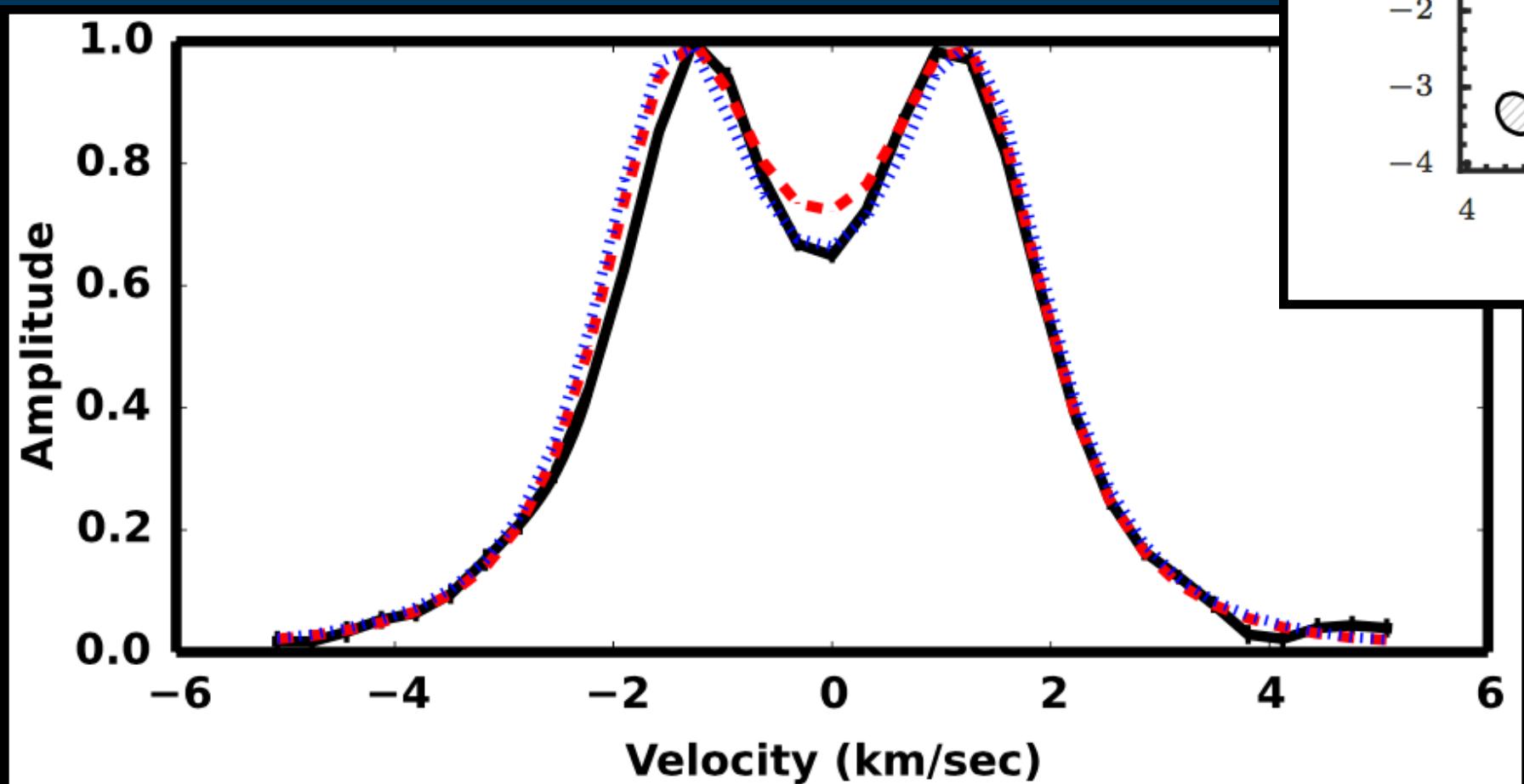
By modeling D<sub>2</sub>O/HDO, HDO/H<sub>2</sub>O and using layered ice model the Furuya +2015 models can explain water's early evolution.



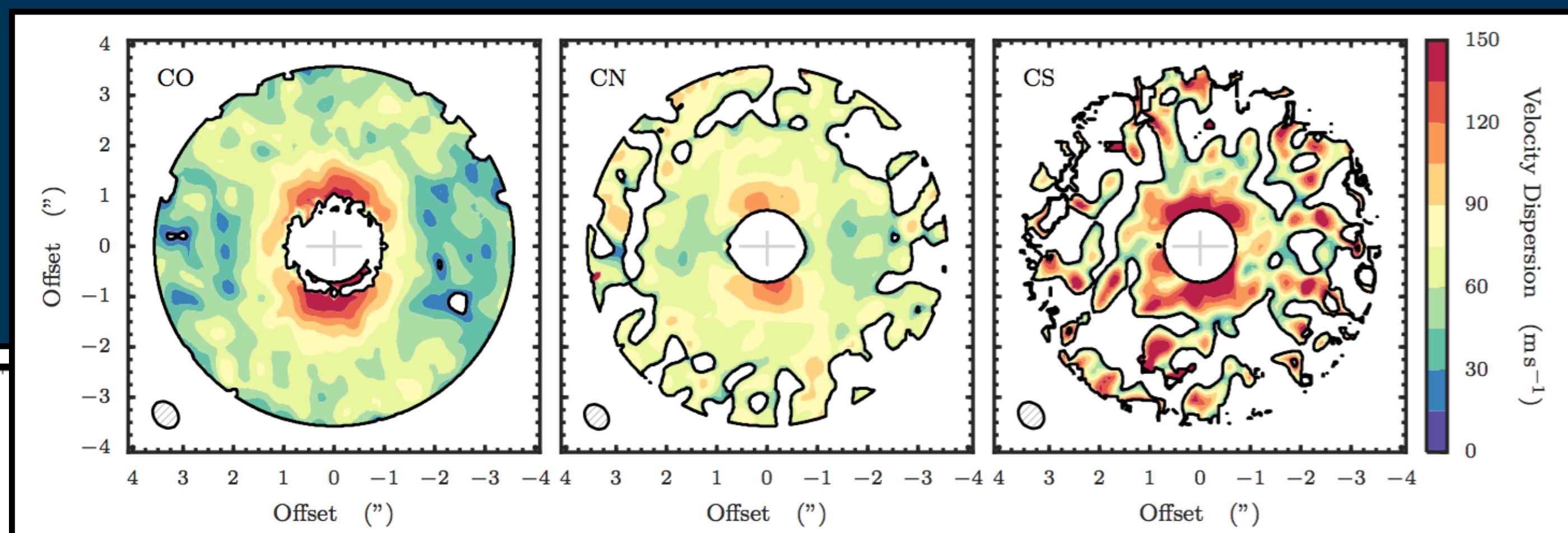
## II) The role of gas kinematics on H<sub>2</sub>O?

Cold water formation via turbulent mixing? e.g., Furuya et al. 2013, Albertsson et al. 2014

Constraints on disk turbulence are low, e.g., Hughes+2011, Guilloteau+2012.



CO 2-1, Flaherty+2015

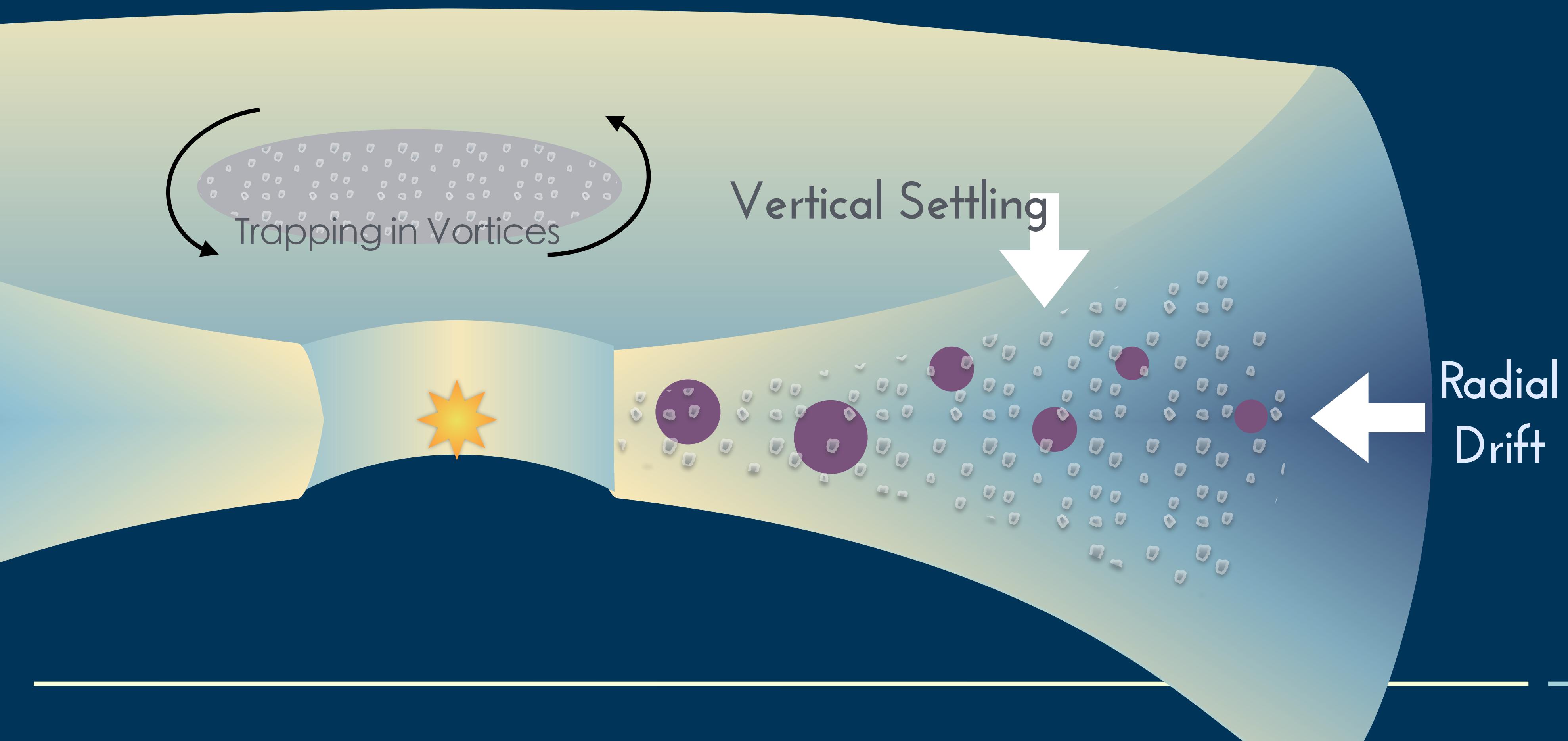


"CO and CN displayed a near constant  $v_{\text{turb}} \sim 0.2$  cs. However, the analysis of the possible sources of errors shows that these numbers should most likely be interpreted as upper limits."

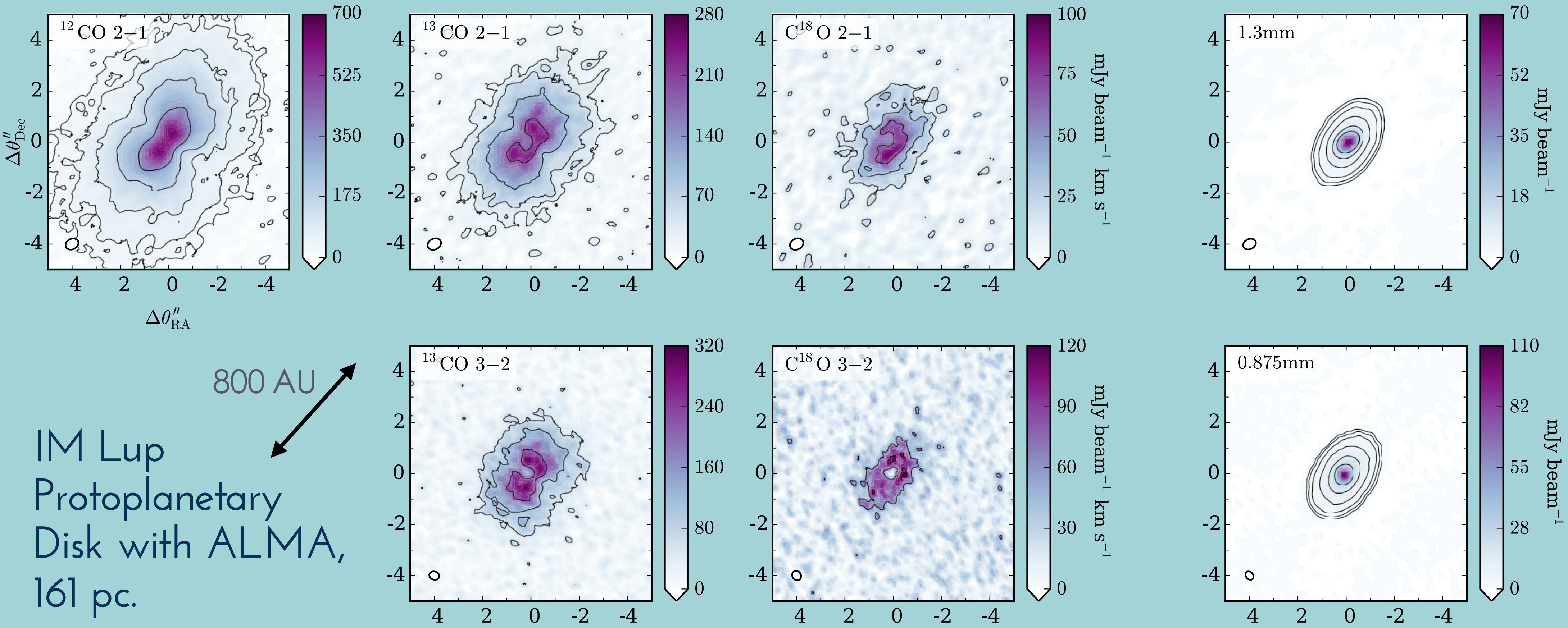
Teague+2016

### III) Aerodynamics: Differential Evolution of Solids

- Redistributions volatiles carried in the ices (Hogerheijde+2010, Bergin +2016, Du+2015, 2016, sub.). Changes the C/O ratio (e.g., Piso+2015).

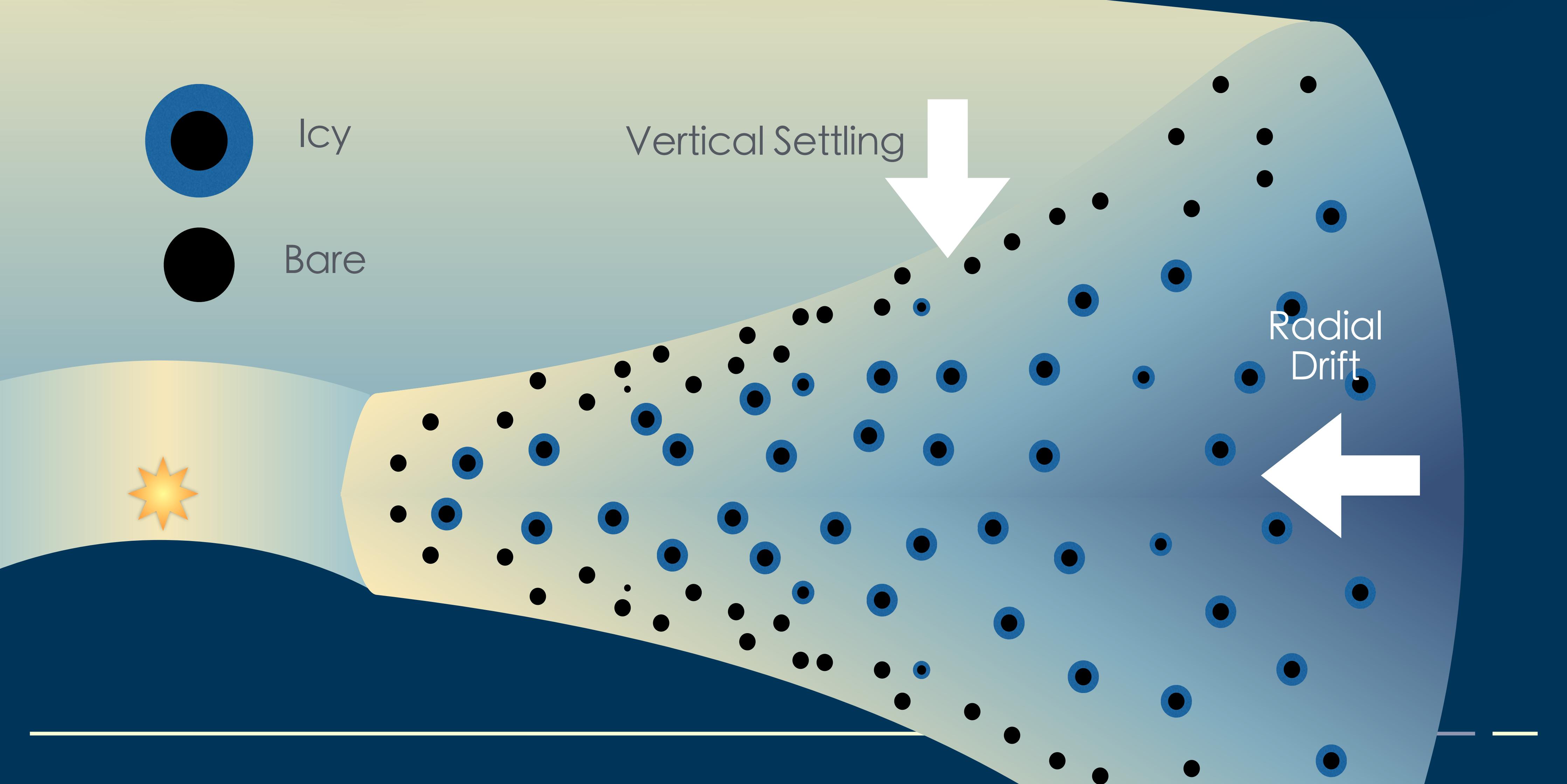


### III) Aerodynamics: Differential Evolution of Solids



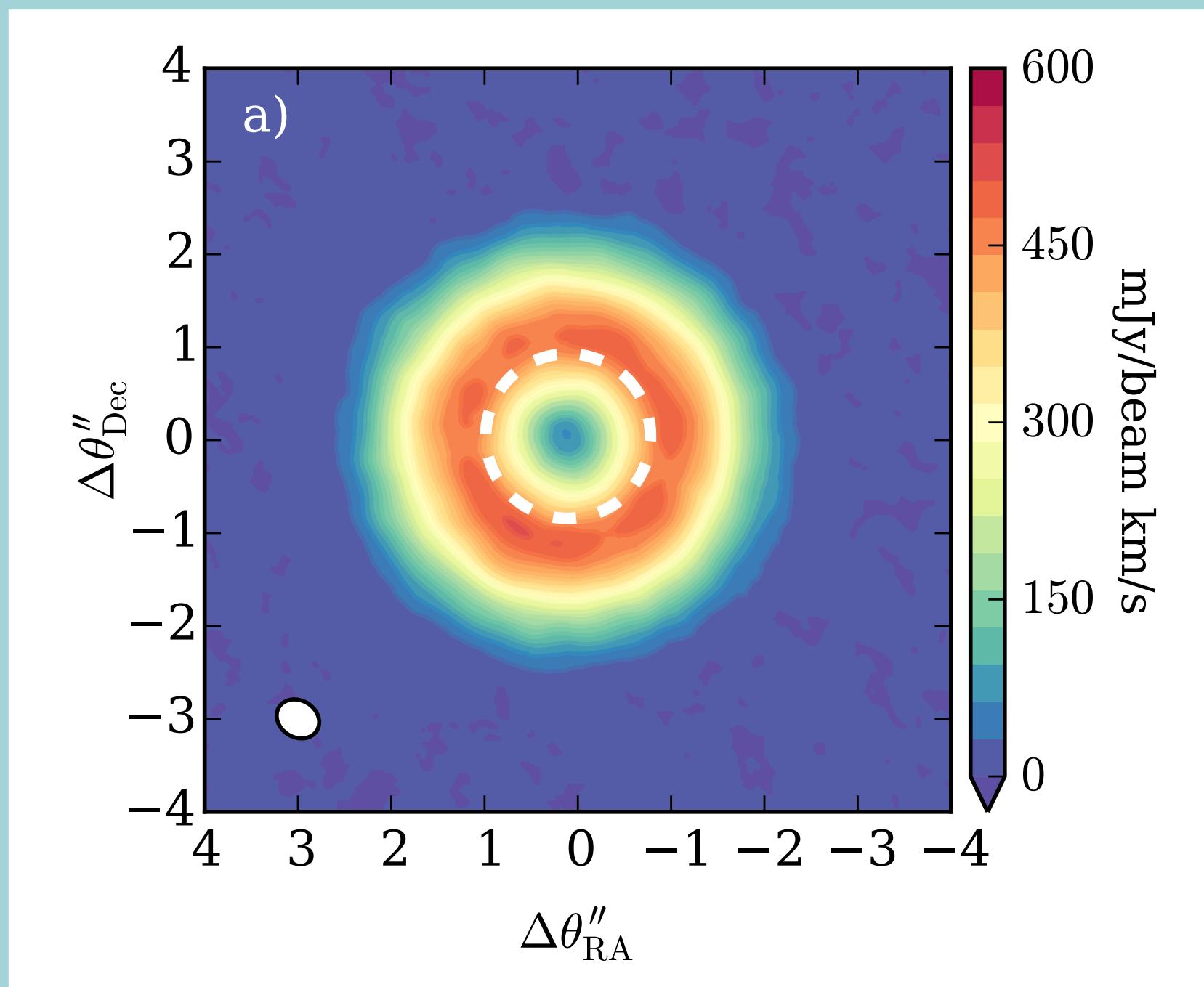
Panic et al. 2009, Cleeves et al. 2016c

### III) Aerodynamics: Differential Evolution of Solids



# III) Aerodynamics: Differential Evolution of Solids

Extremely bright C<sub>2</sub>H in two disks with ALMA! Bergin et al. 2016



Surface and outer disk is UV dominated.

C, N (little O) in H<sub>2</sub> rich gas – is a hydrocarbon factory (Du, Bergin, & Hogerheijde 2015)

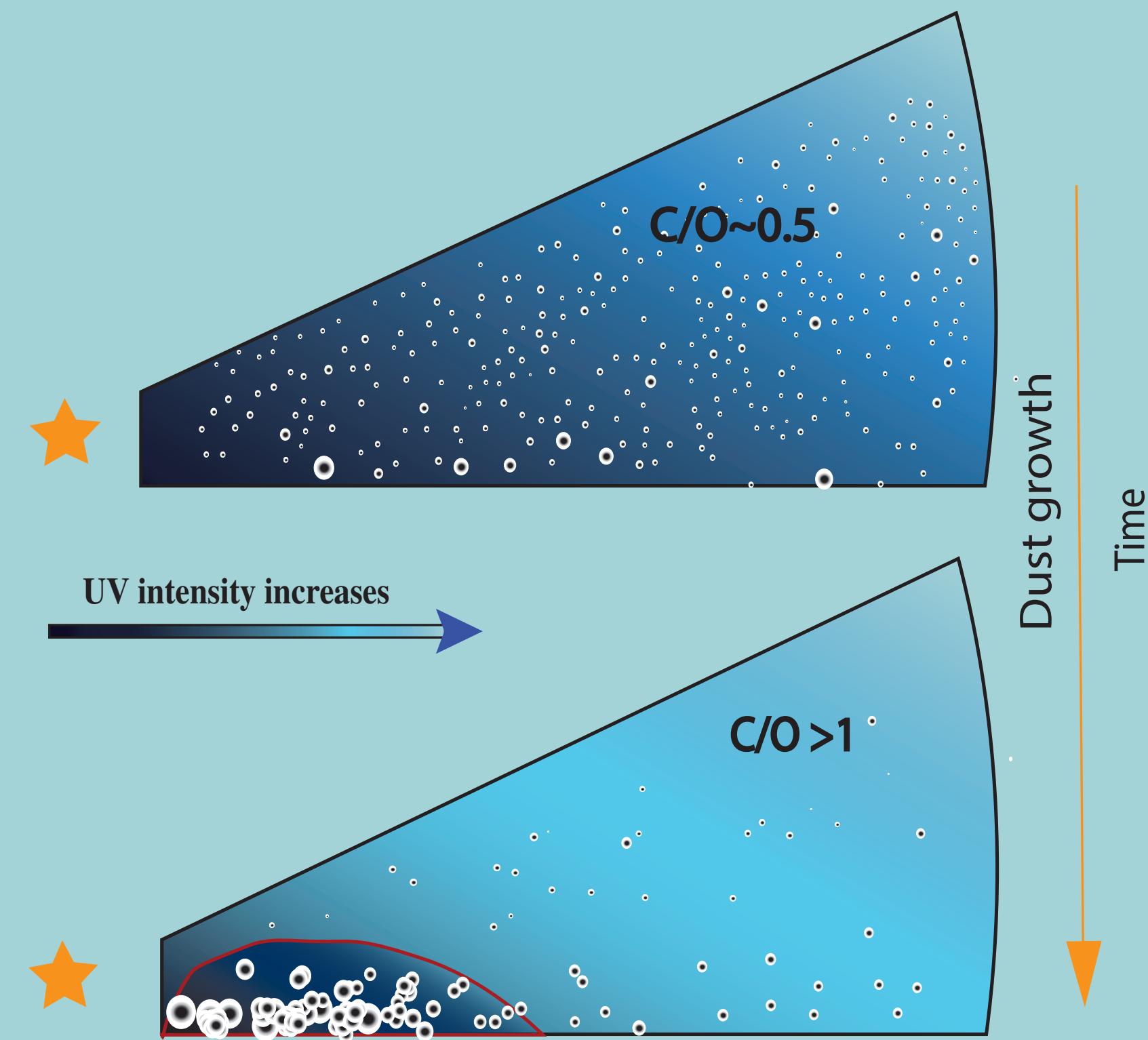
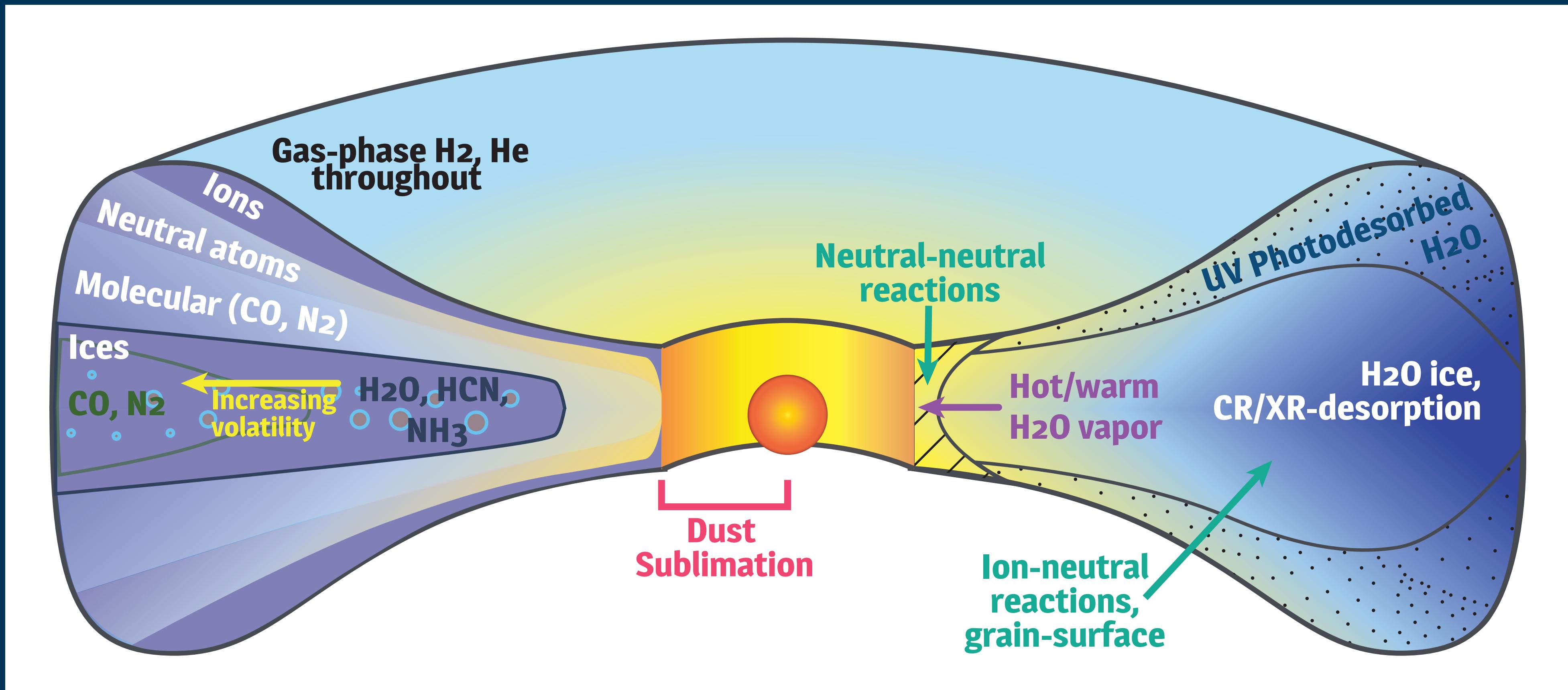


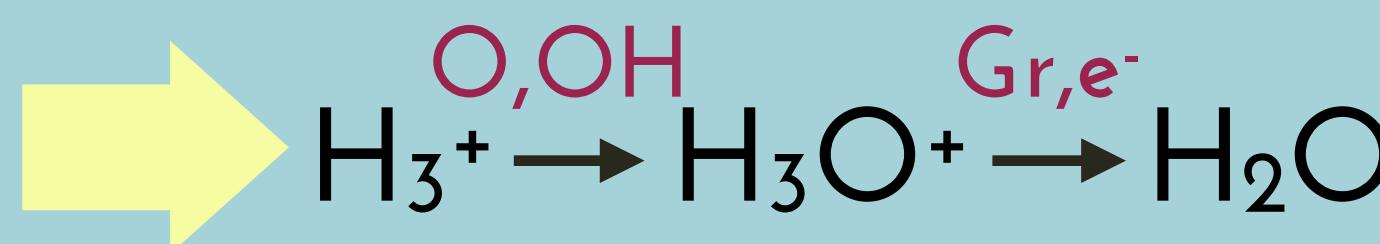
Figure courtesy of Ted Bergin

# IV. In situ disk chemistry?



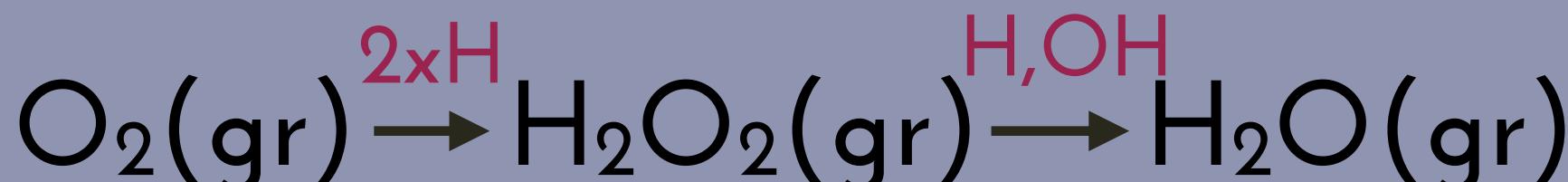
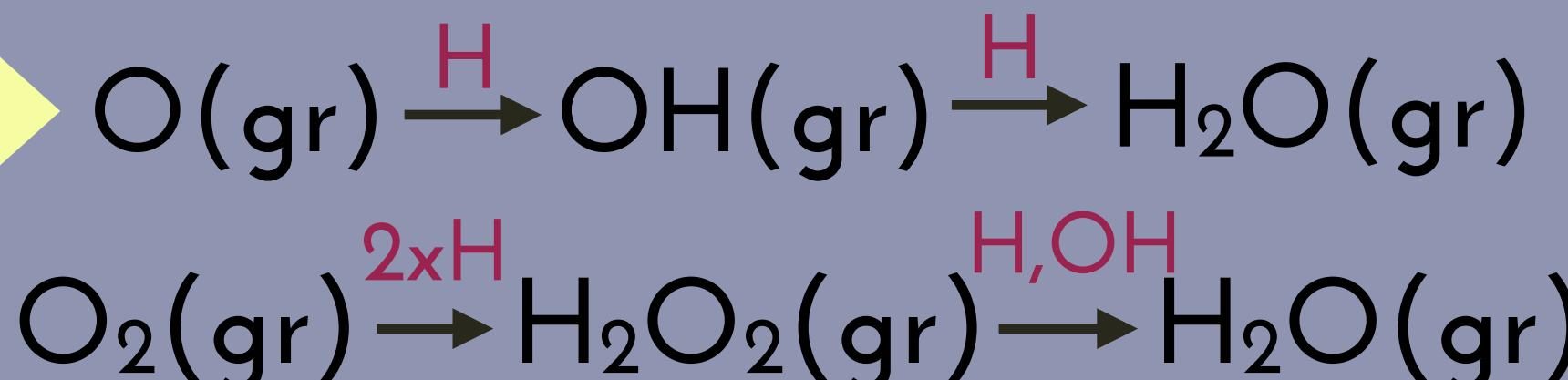
# IV. In situ disk chemistry?

## I. In the Gas Phase



Neutral-Neutral like  $\text{H}_2 + \text{OH}$   
only at high ( $T > 200$  K)  
temperatures.

## II. On Grain Surfaces



## IV. In situ disk chemistry?



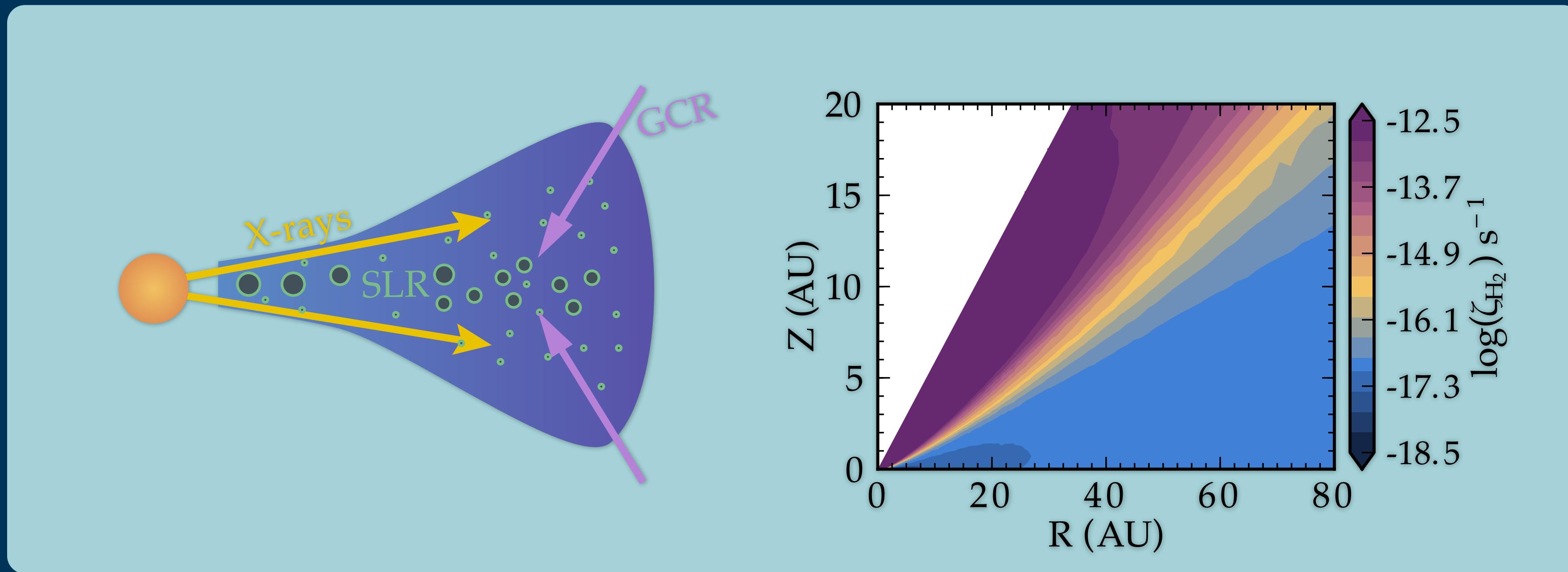
Formation of water at low temperatures requires a source of  
*molecular hydrogen ionization.*

Possible sources: 1. cosmic rays, 2. X-rays, 3. radionuclide decay.

# Modeling Tools: High Energy Processes

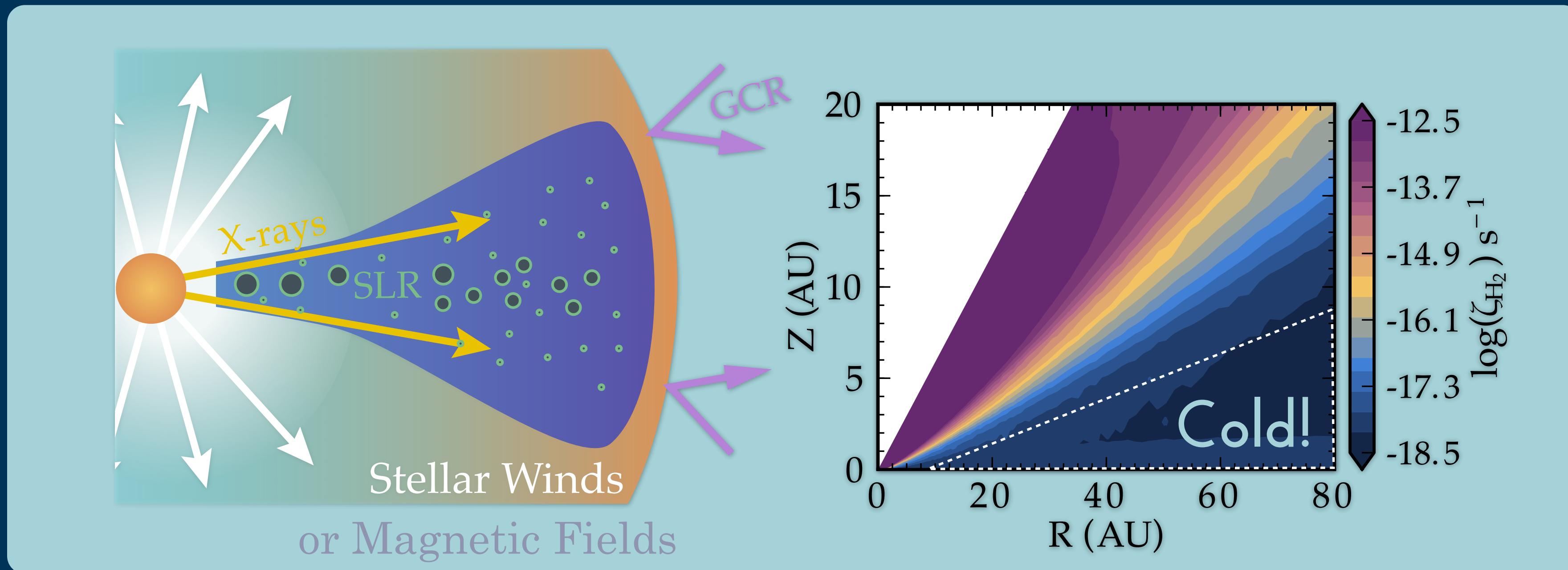
Ionization source:	How we treat:	How this is unique/ new:
X-rays (Bethell)	Monte Carlo	Treat gas and dust opacity separately
UV (Bethell)	MC Continuum, Line	Ly- $\alpha$
Cosmic Rays (Cleaves)	Multiple input spectra with consistent vertical transfer.	Flexibility in CR rates beyond normalization
Radionuclides (Cleaves)	Infinite Slab with Loss	Loss terms with ability to include settling.
Interstellar External Radiation (Cleaves)	Numerically integrate outside in	3D Treatment

# The Classical Picture of Disk Ionization

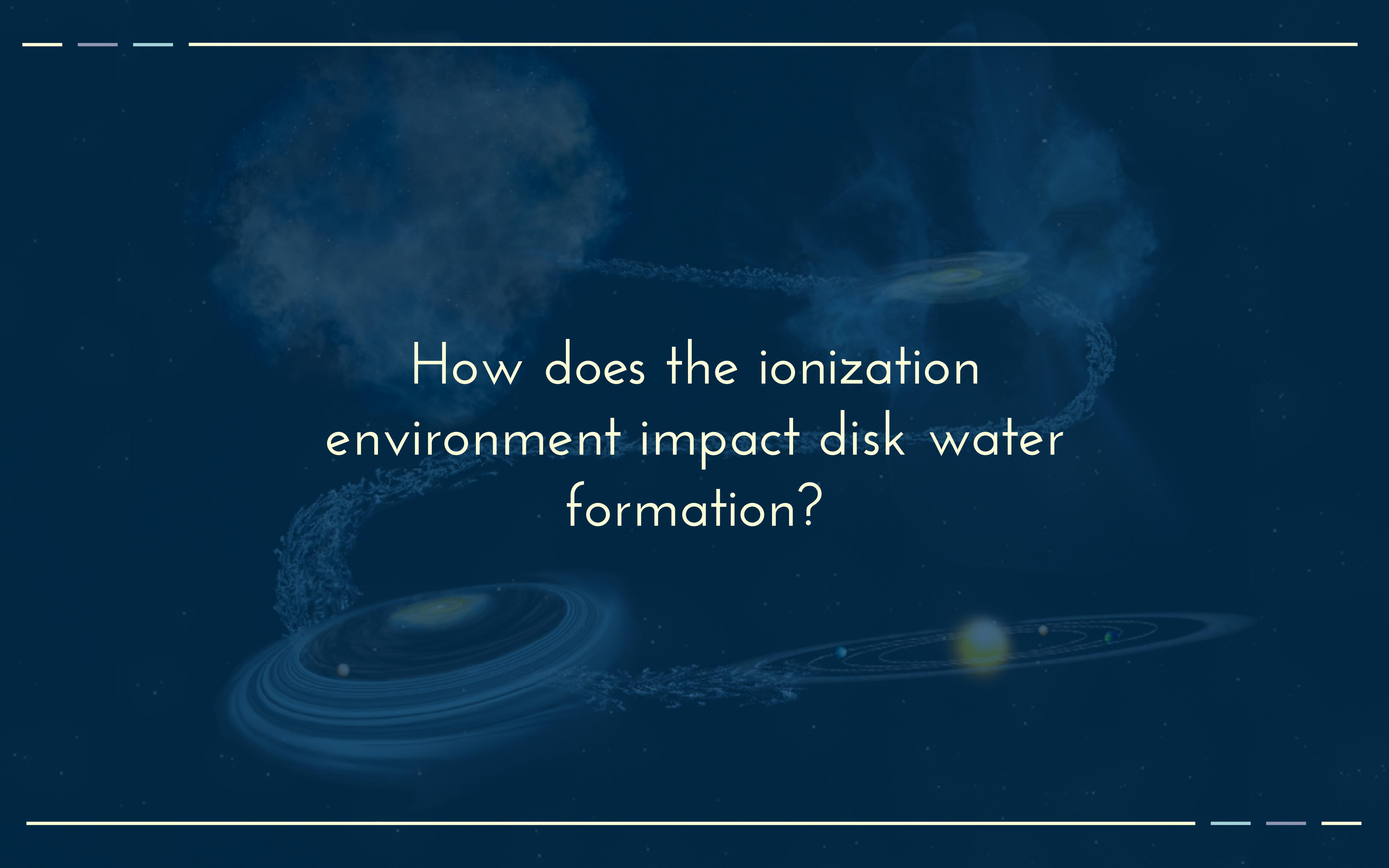


Glassgold 1997, 2000, 2001 (and more),  
Igea & Glassgold 1999, Umebayashi+1989,  
2009, Ilgner & Nelson 2006a/b, 2008.

# An Updated Picture of Disk Ionization



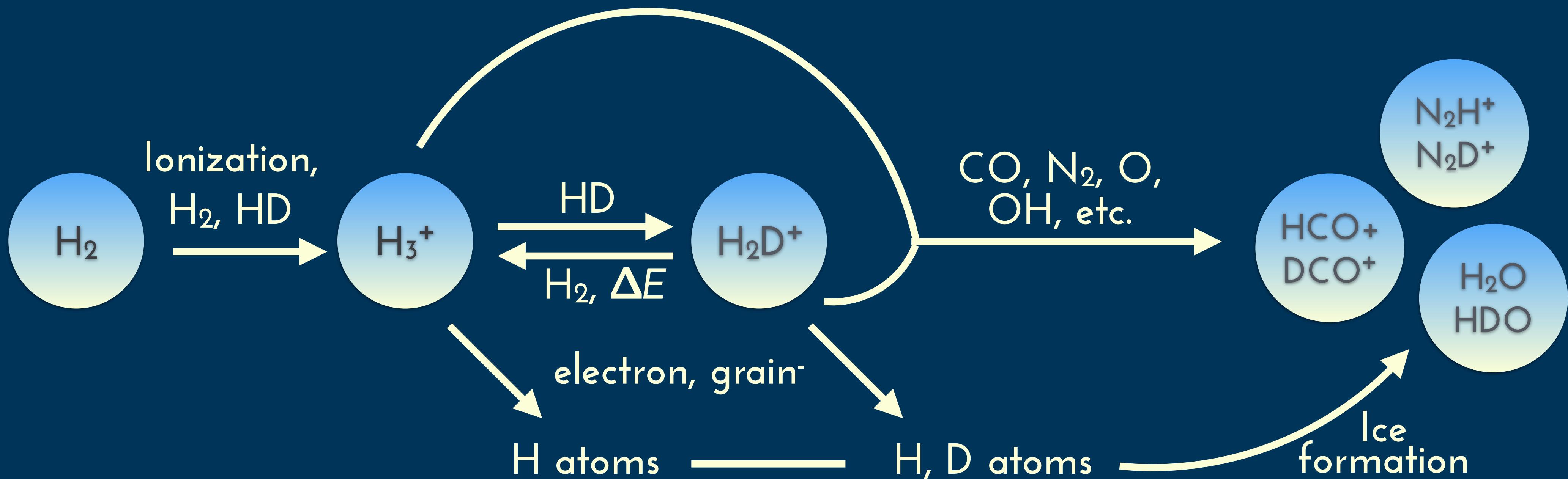
Cleeves et al. 2015 measured a significantly subinterstellar CR ionization rate in TW Hya ( $> 100\times$  reduced).



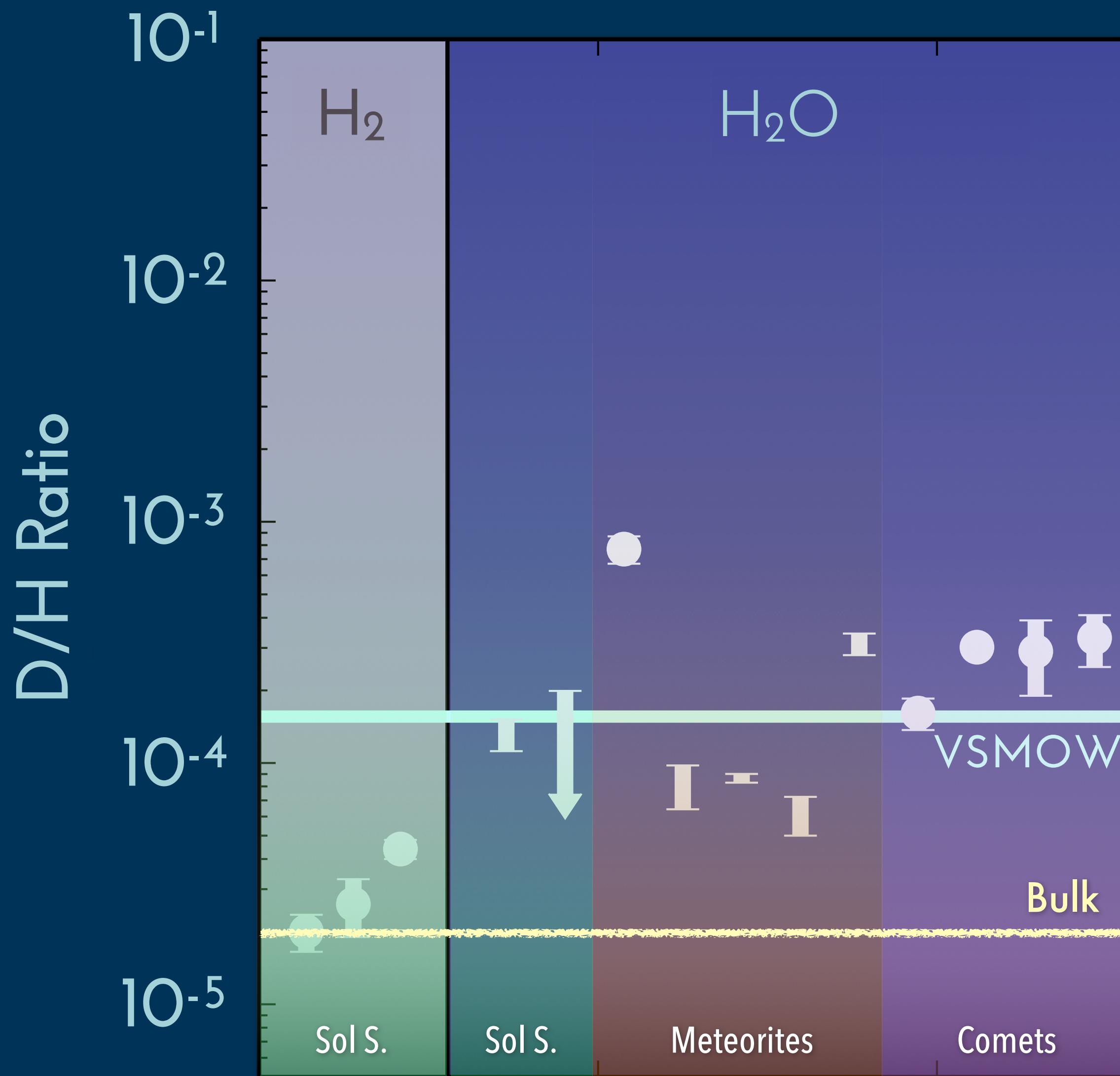
How does the ionization  
environment impact disk water  
formation?

# Additional Clues

*Cold water “chemically tagged” with high D/H.*

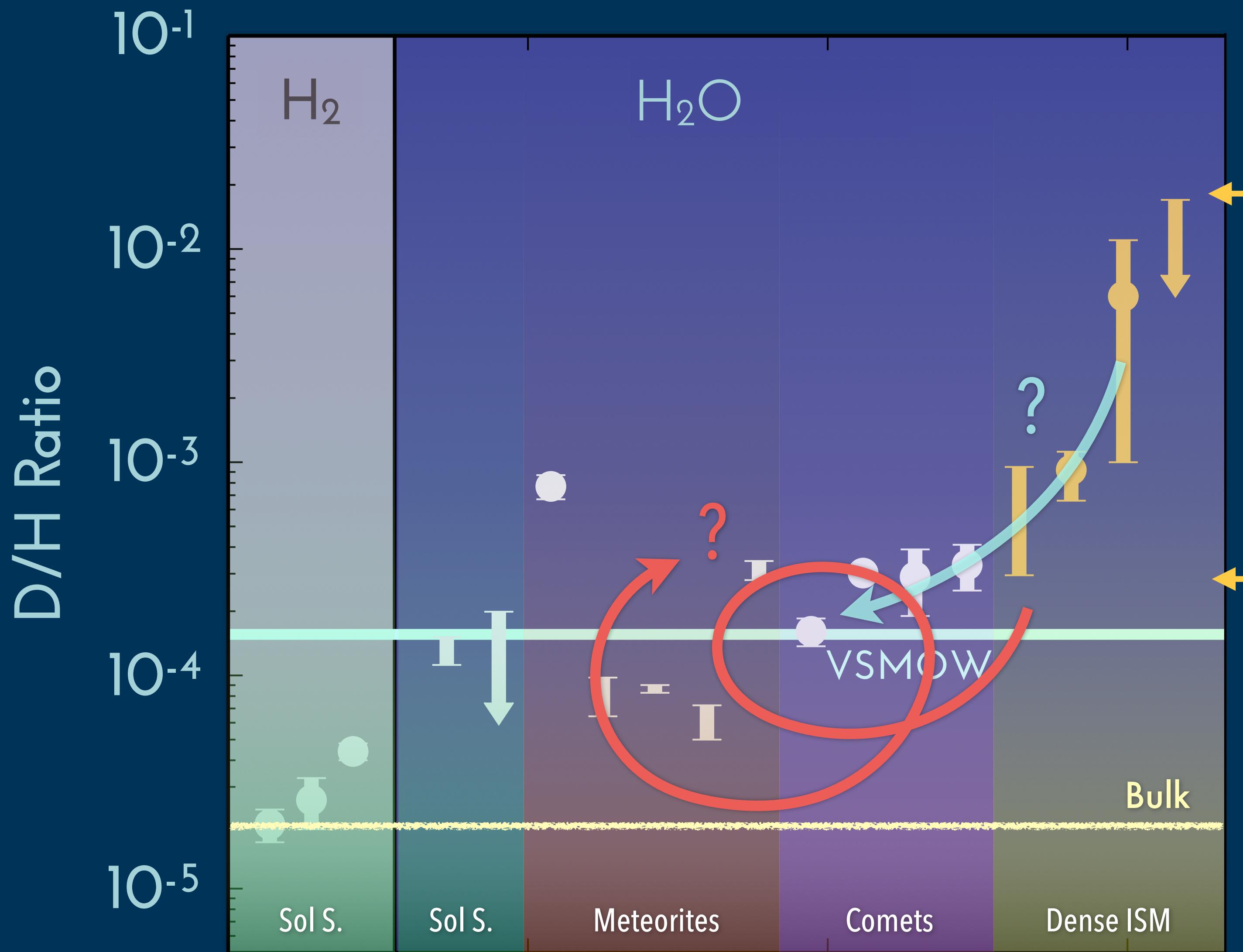


Supports fractionation  
up to  $T < 50 \text{ K}$



Water throughout a diversity of solar system bodies has characteristically high HDO/H<sub>2</sub>O.

We know water formed in a relatively cold environment.



ISM: Persson+2014, 2012, Coutens+2012, Parise 2003.

Primordial ices in the envelopes of protostars exhibit a high level of D/H.

Are these early stages (the primordial ISM ices) chemically linked?

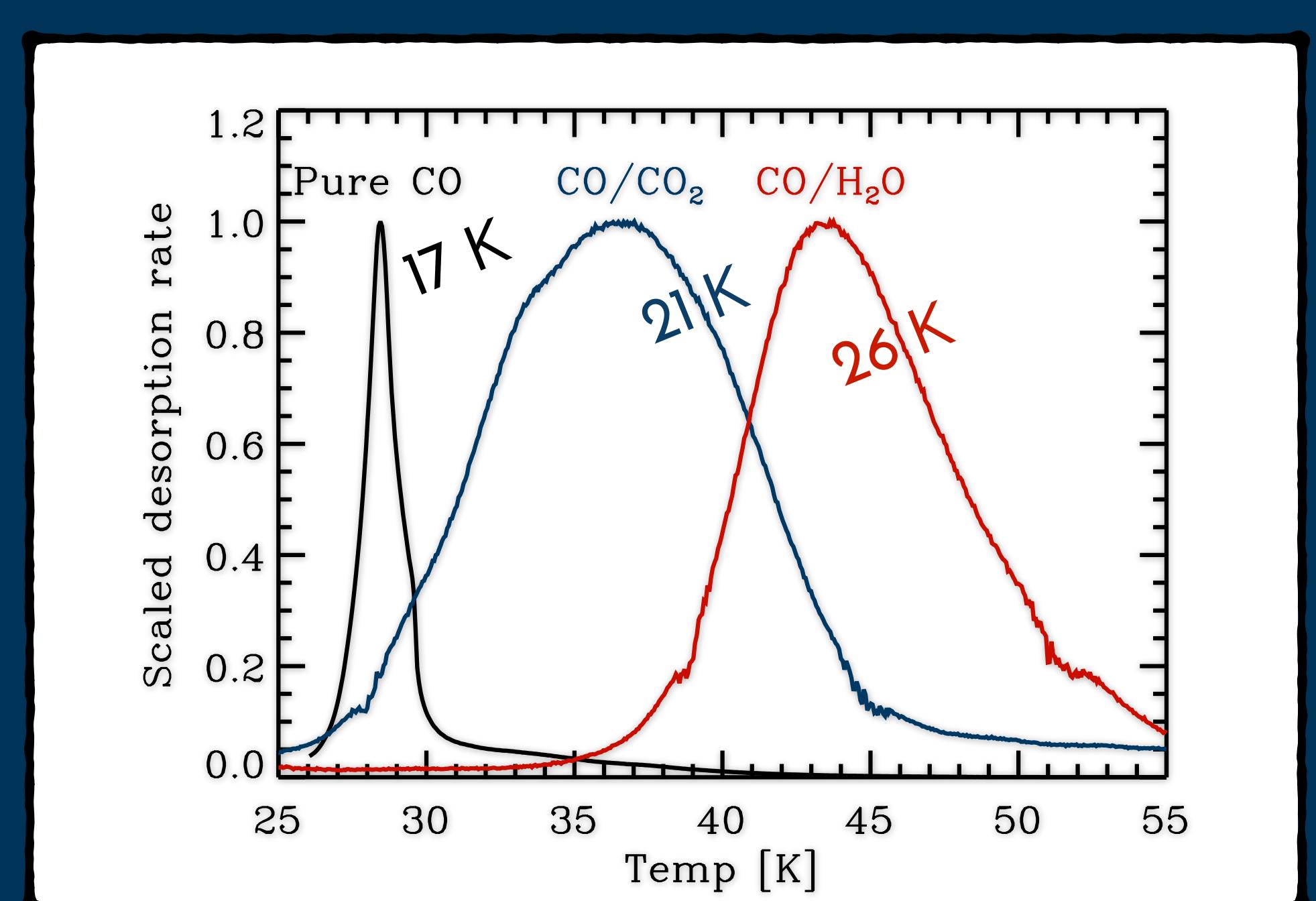
What is the role of disk chemistry?

## Hitting reset:

Starting out with  $\text{HDO}/\text{H}_2\text{O} = \text{HD}/\text{H}_2$ , how much does cold water formation in the disk elevate  $\text{HDO}/\text{H}_2\text{O}$ ?

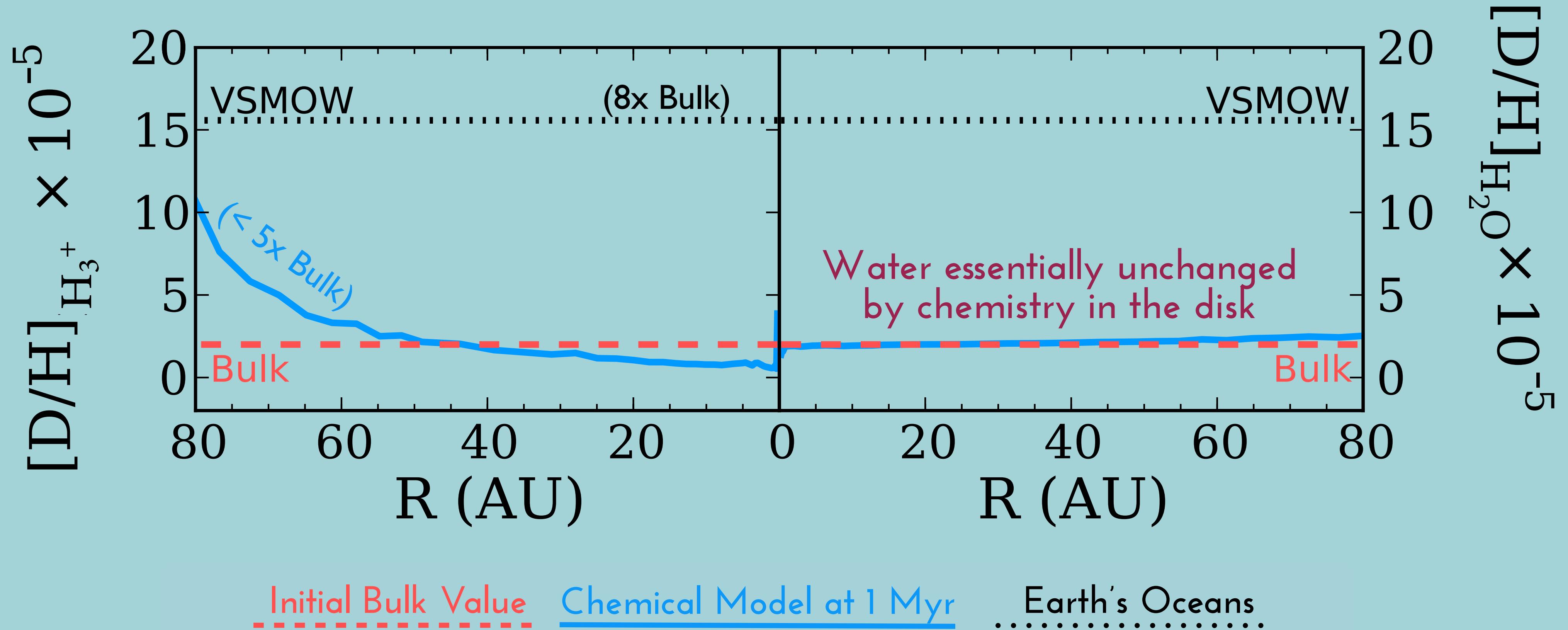
# Chemical Model

- \* Mini-deuterium chemical network.
  - \* 6268 reactions, 600 species.
- \* H<sub>2</sub>/HD/D<sub>2</sub> self-shielding  
(Wolcott-Green+ 2011)
- \* Simple grain-surface chemistry  
(Hasegawa, Herbst, Leung 1992)
- \* Thermal o/p ratios for H<sub>2</sub> and H<sub>2</sub>D<sup>+</sup>  
(Lee & Bergin 2015)
- \* Warm fractionation reactions  
(Thi+ 2010)



And updated lab data on CO binding energies for oxygen-regulation (Graninger).

# HDO/H<sub>2</sub>O Results (1 Myr)



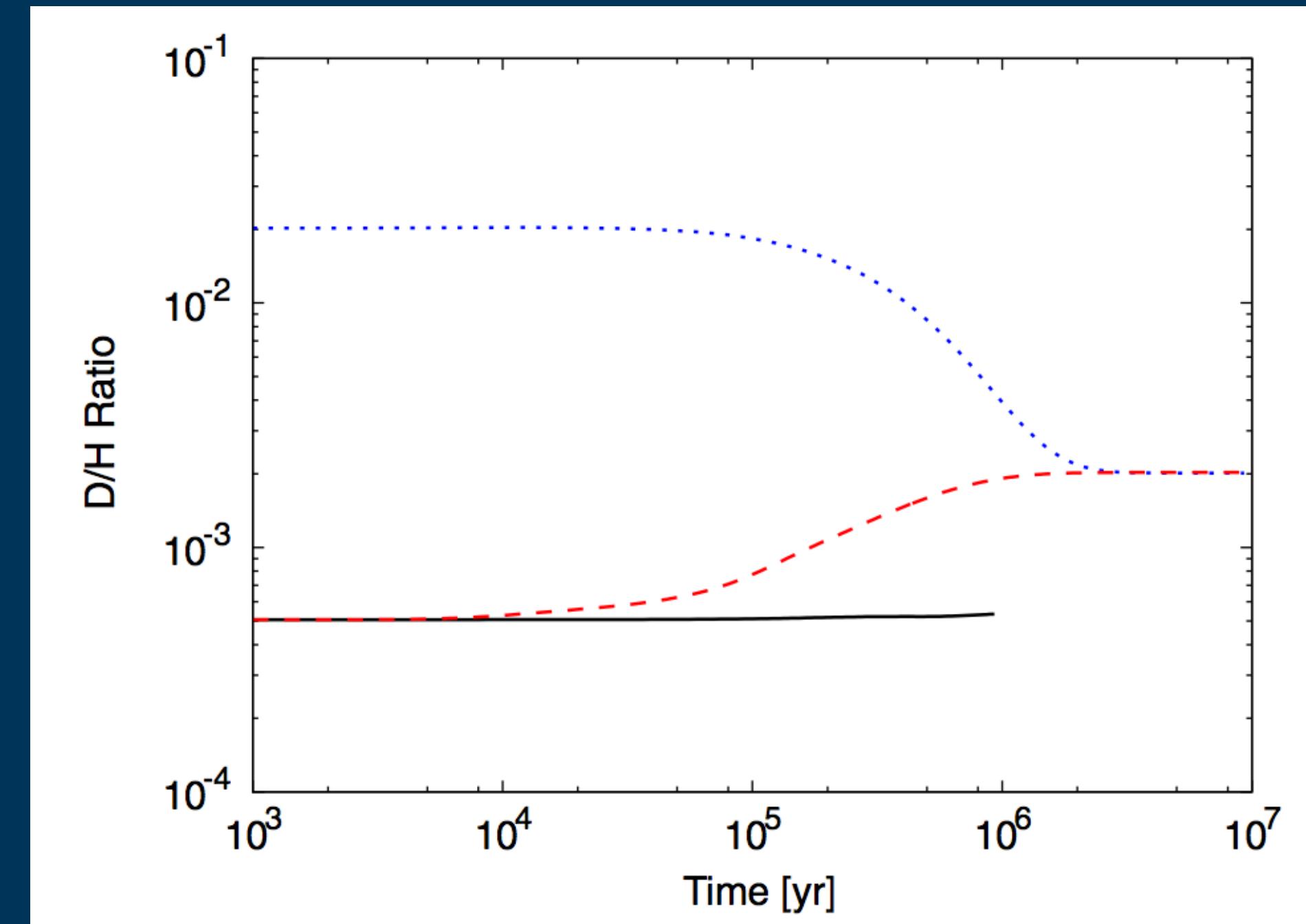
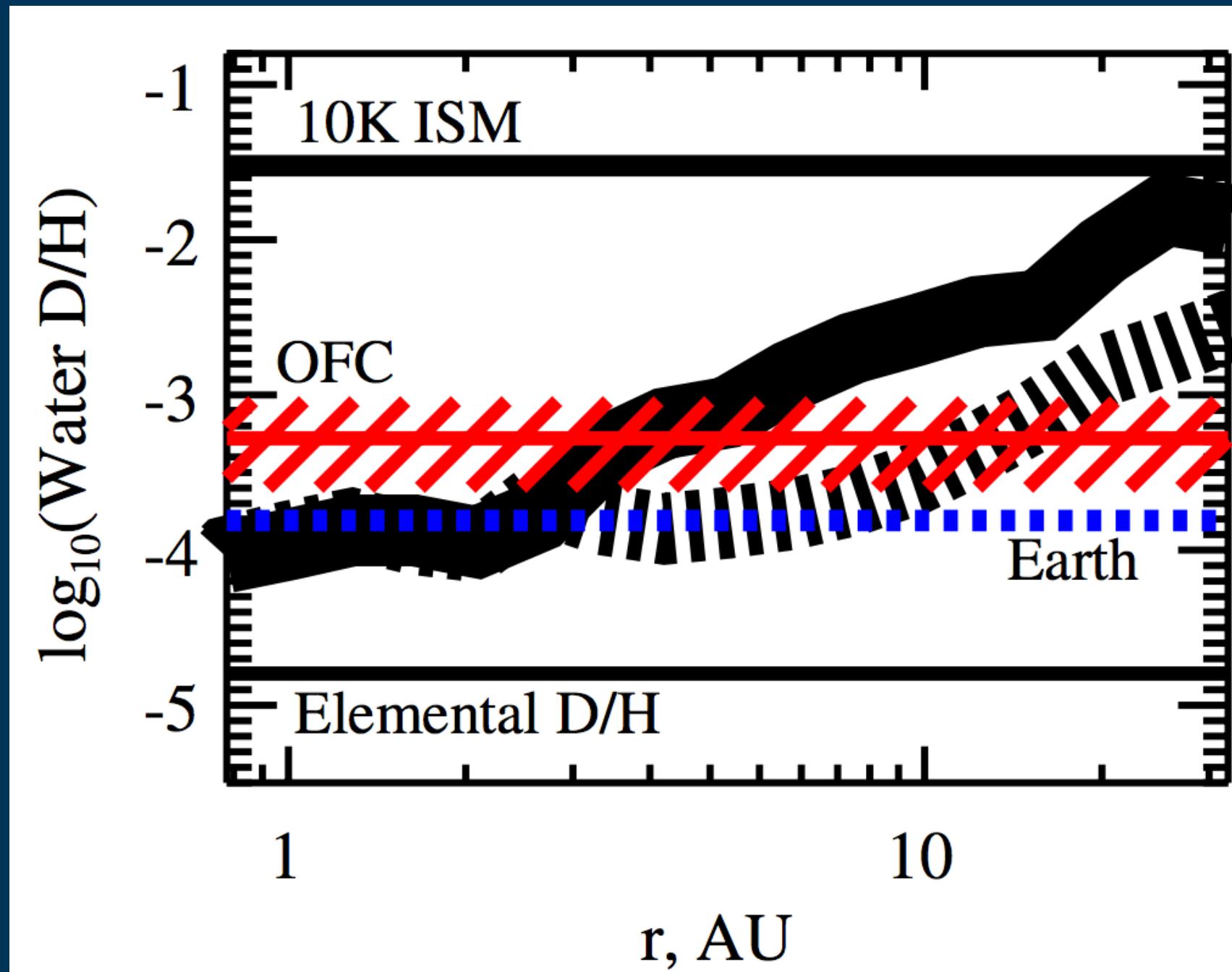
# Disk-Sourced Cold Water: Results

- Chemistry in a laminar disk is not a viable source origin for cold (deuterated) water Solar System.
- These conditions require ISM heritage such that interstellar ices would be incorporated into comets, meteorites, and Earth's oceans, 30-40%.

# Disk-Sourced Cold Water: Results

- ➡ Chemistry in a laminar disk is not a viable source origin for cold (deuterated) water Solar System.
- ➡ These conditions require ISM heritage such that interstellar ices would be incorporated into comets, meteorites, and Earth's oceans, 30-40%.

# But what about mixing?



"Turbulent mixing slowly transports some of the water ice into warmer or irradiated regions where it desorbs and is quickly defractionated..." - Albertsson et al. 2014

"...atomic oxygen is transported from the surface to the deeper region and (re)forms H<sub>2</sub>O and HDO ices." - Furuya et al 2013

But what about mixing?

Scenario 1: Mixing  
high D/H water up,  
reducing D/H  
( $\text{H}_2\text{O}$ )

High temperatures

HDO  $\text{H}_2\text{O}$

Low  
temperatures,  
H-atoms?

But what about mixing?

Scenario 2: Mixing  
oxygen to the cold  
midplane, enhancing  
 $D/H$  ( $H_2O$ )

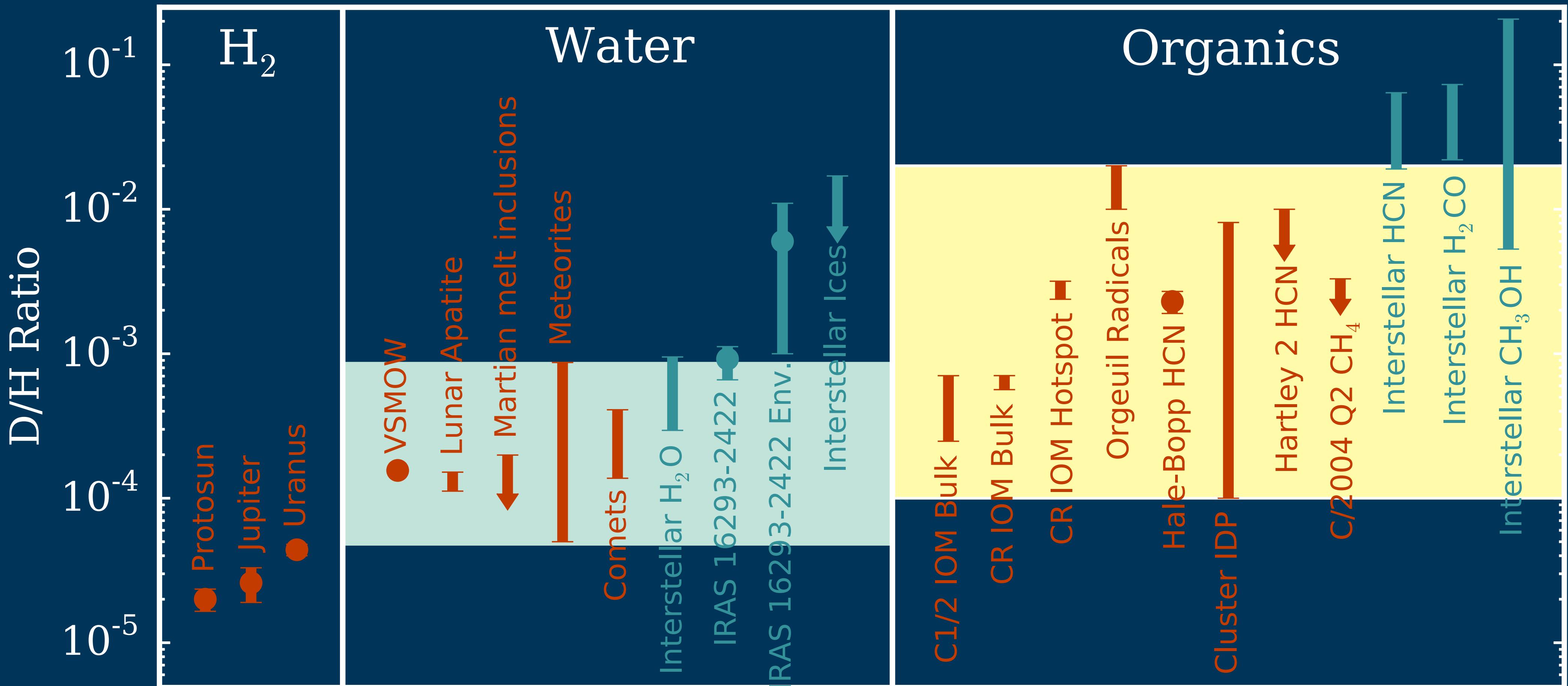
High temperatures

O, OH, H, D

HDO       $H_2O$

Low  
temperatures,  
H-atoms?

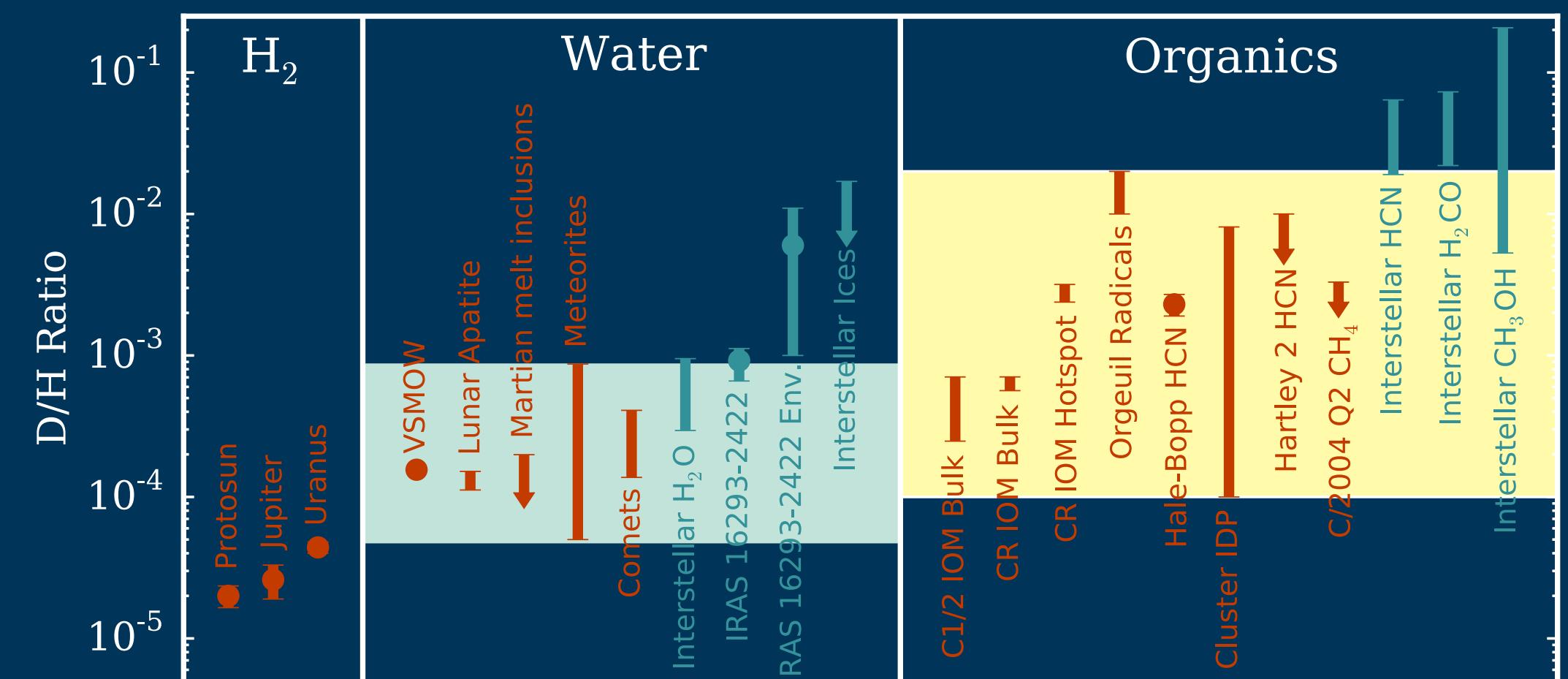
# D/H in Water vs. Organics



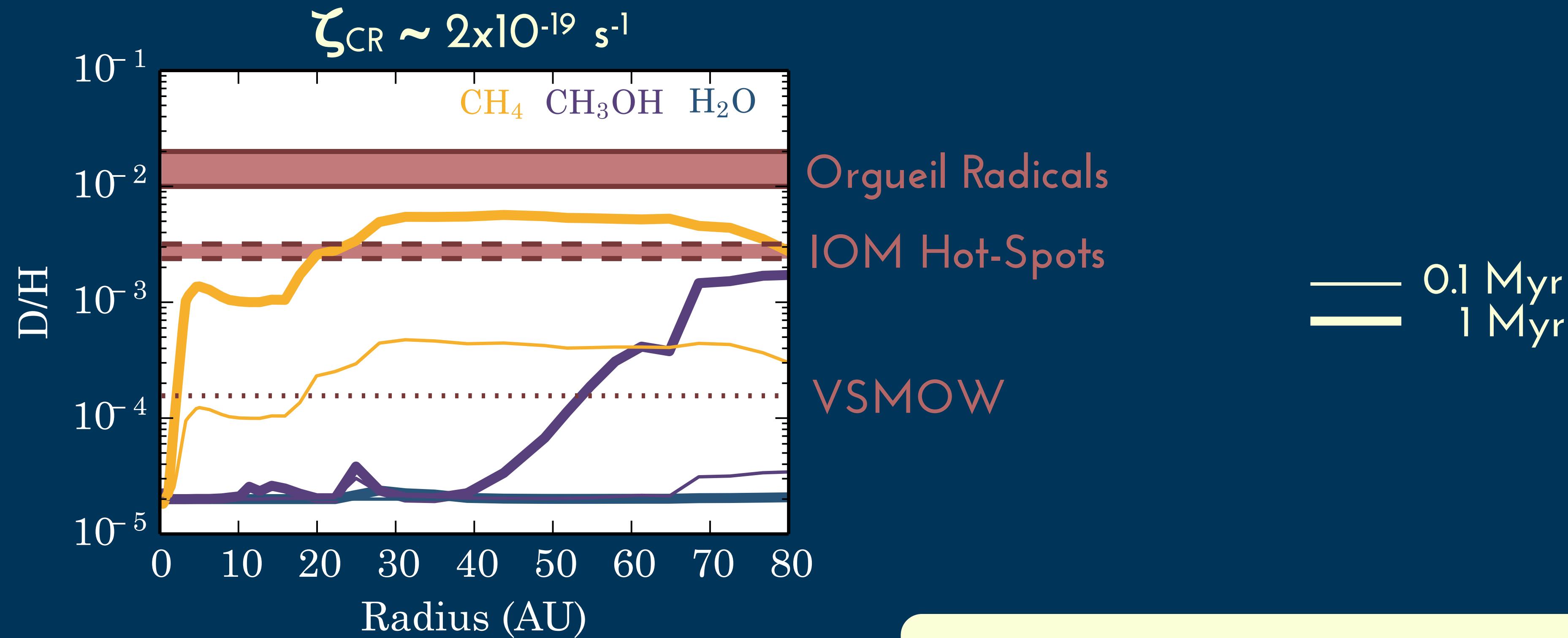
Cleeves, Bergin, Alexander, Du, Graninger, Öberg, Harries, 2016

# D/H in Water vs. Organics

- \* Globally higher organic D/H than water. Perhaps due to:
- \*  $\text{CH}_3^+ + \text{HD} \rightleftharpoons \text{CH}_2\text{D}^+ + \text{H}_2 + \Delta E$  (Roueff+2013).
- \* Larger range in organic D/H = many reaction pathways?

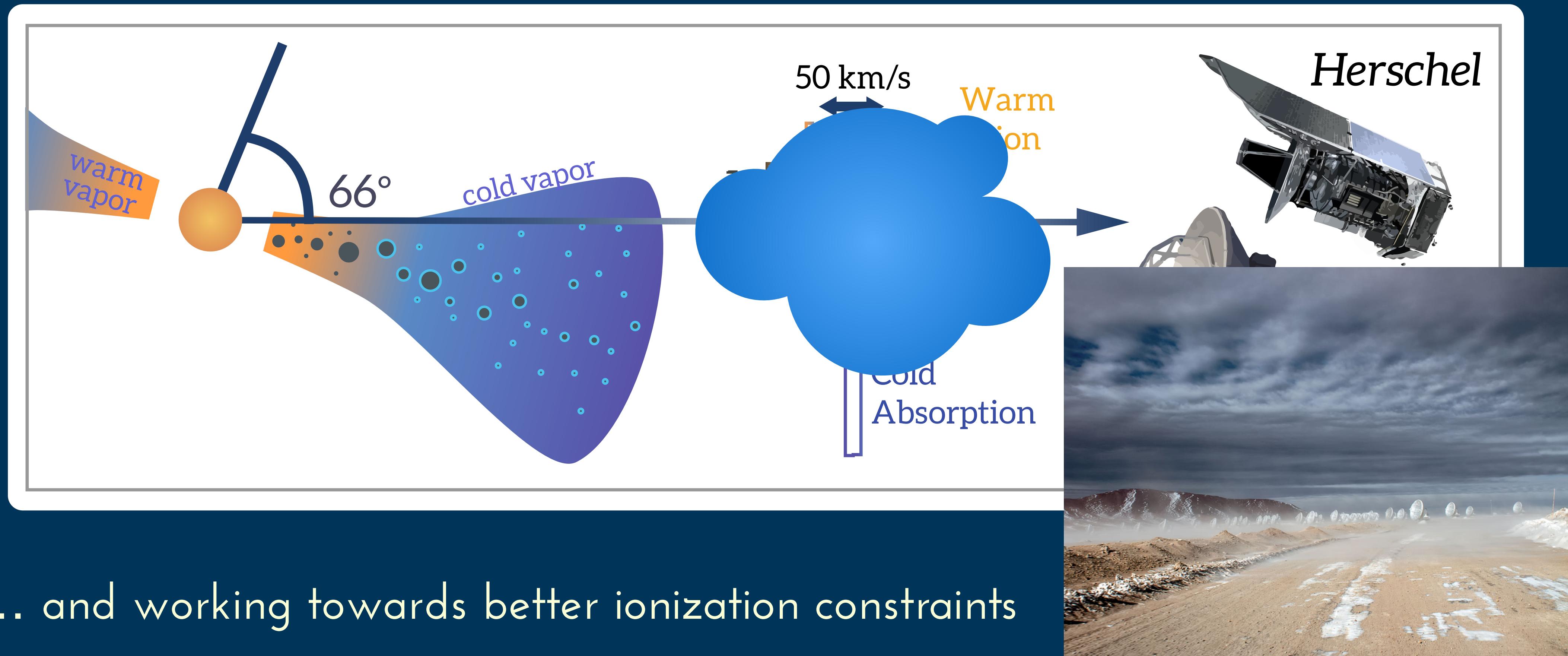


# Midplane Deuterium Fractionation in Hydrides



Interesting predictions for cometary  
D/H in  $\text{CH}_3\text{OH}$ ?

# Future: HDO/H<sub>2</sub>O in a Protoplanetary Disk



# Summary: Cold Water Evolution In Disks

I. Primordial material: Models show bulk water survives disk formation  
(Visser+09, Furuya+16)

II. Kinematics: Observations show mixing weak, and differing model predictions  
(Albertsson+14, Furuya+13, Teague+16)

III. Aerodynamics: Observations show surprisingly efficient and may redistribute volatile ices  
(Du+15, Bergin+16)

IV. Disk chemistry: Surprisingly inefficient with realistic models of disk ionization physics (Cleeves +15,16)

