

“Molecular physics challenge for exoplanet studies”



**UCL**

# Line lists for hot hydride molecules

Jonathan Tennyson  
Physics and Astronomy  
University College London

**Ex****Mol**

**Calculated molecular line lists**  
for the opacity of exoplanets,  
cool stars and other hot atmospheres

Hydride Toolbox  
Paris  
December 2016

Image credit Shutterstock

“Molecular physics challenge for exoplanet studies”



**UCL**

# Line lists for hot hydride molecules and their friends

Jonathan Tennyson  
Physics and Astronomy  
University College London

**Ex****Mol**

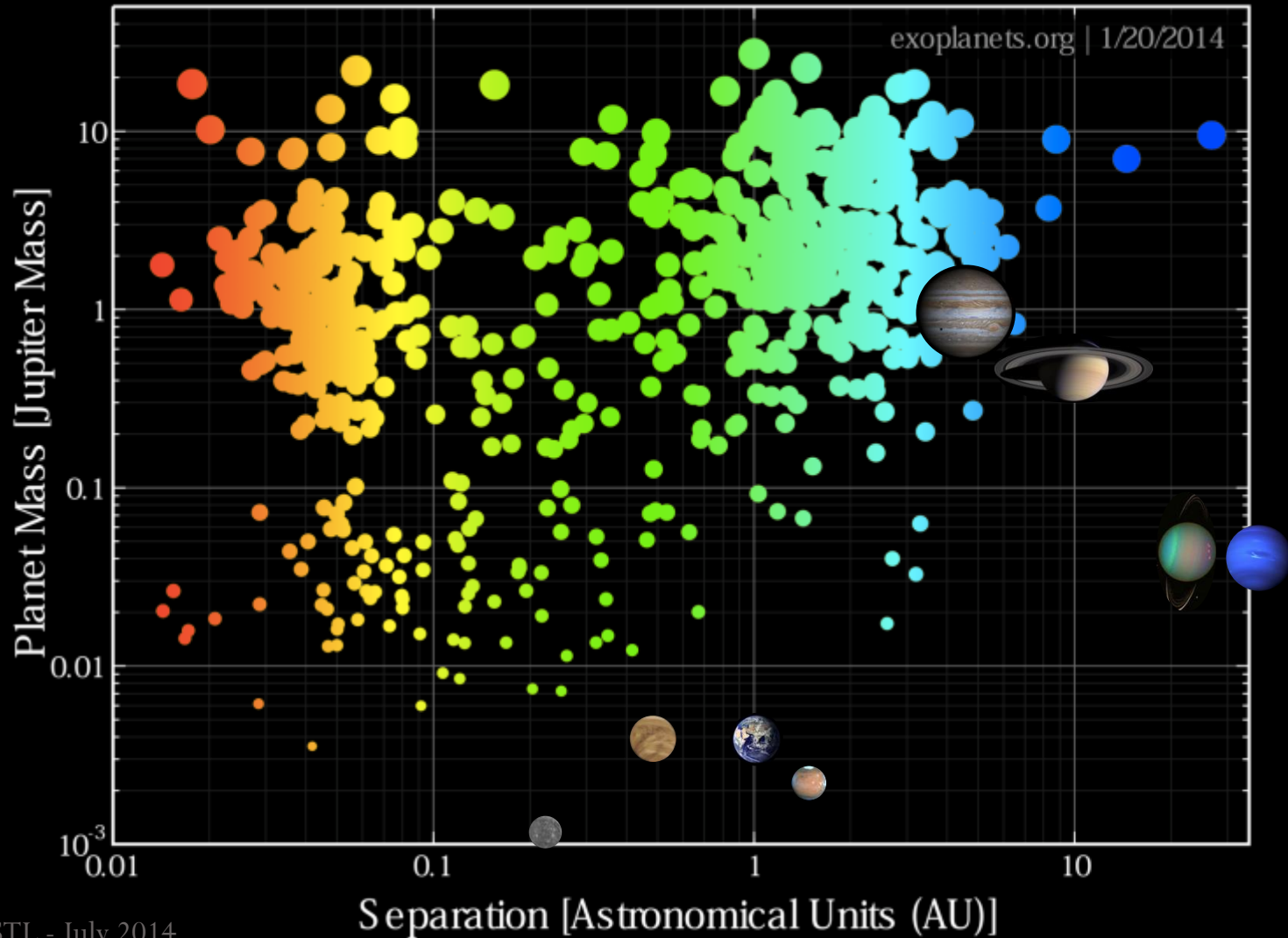
**Calculated molecular line lists**  
for the opacity of exoplanets,  
cool stars and other hot atmospheres

Hydride Toolbox  
Paris  
December 2016

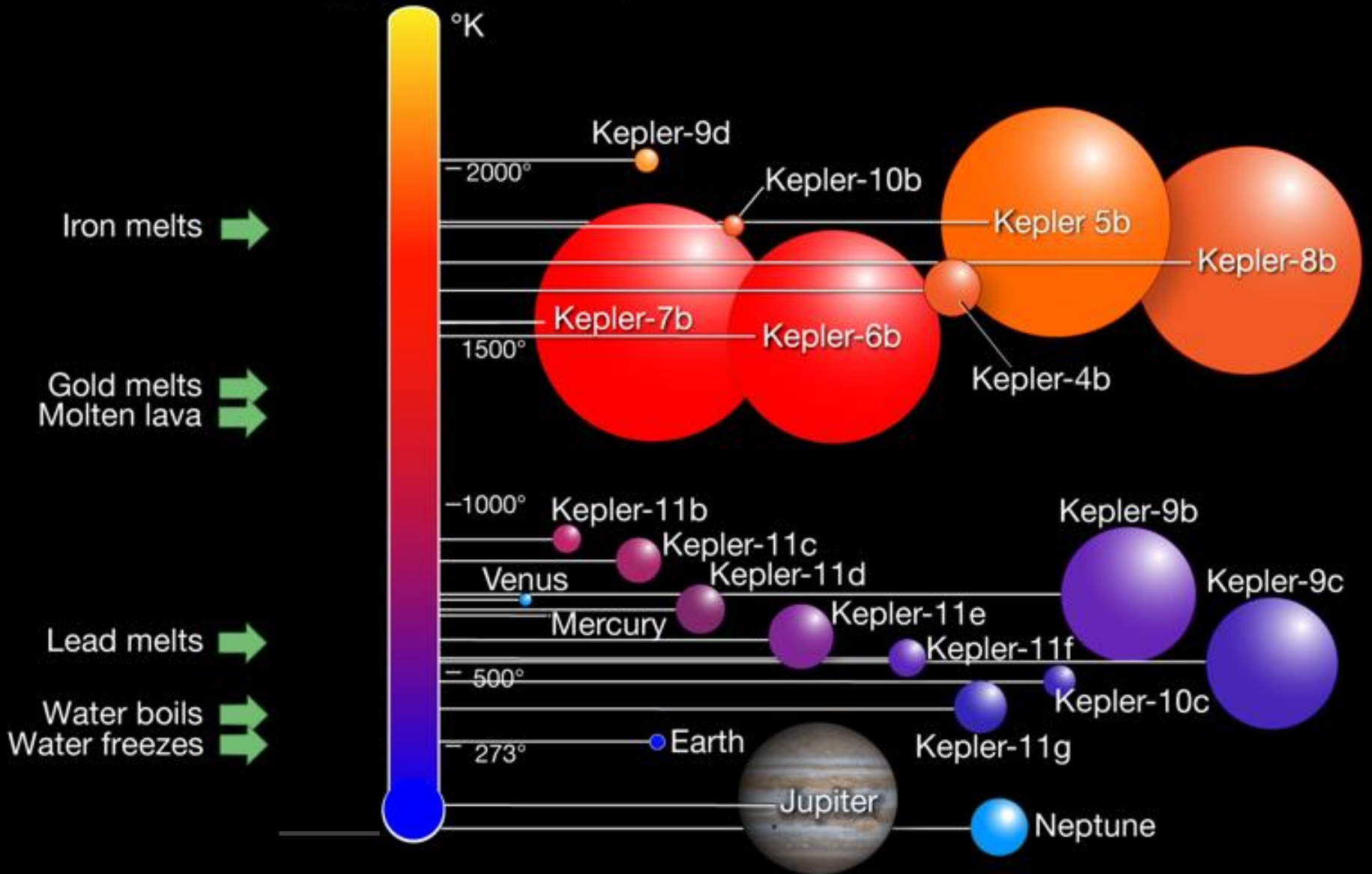
Image credit Shutterstock

# The Exoplanet Revolution

9 to > 3000 in 20 years!



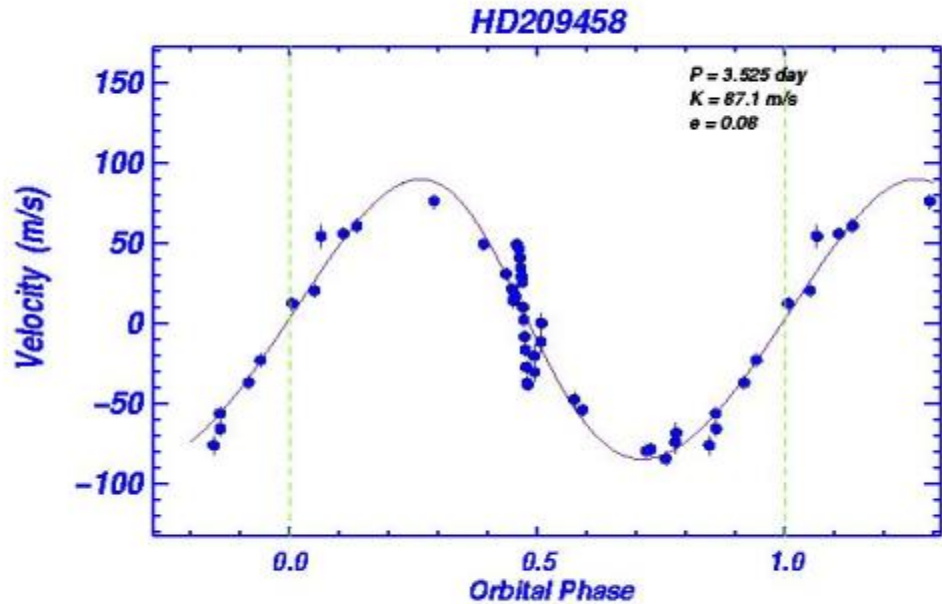
# Planet Temperature & Size



Courtesy of Kepler's team

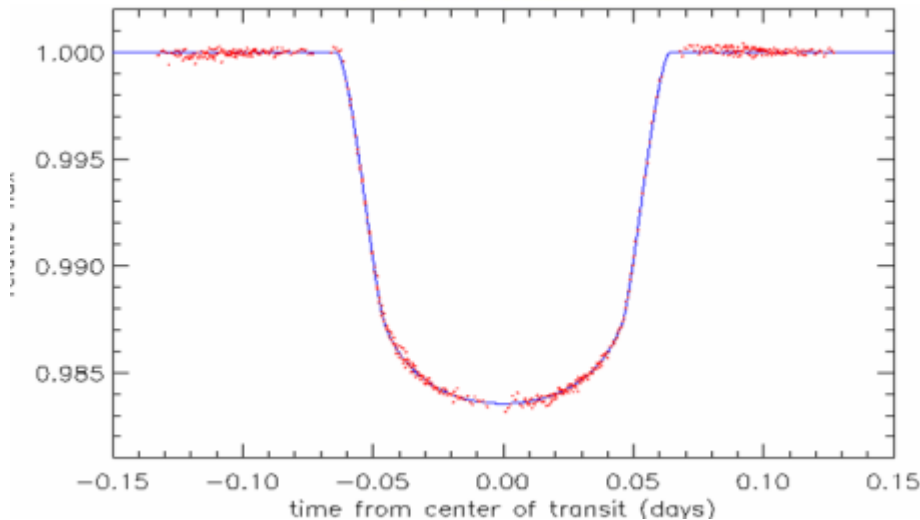


# HD 209458b



Period = 3.52 days

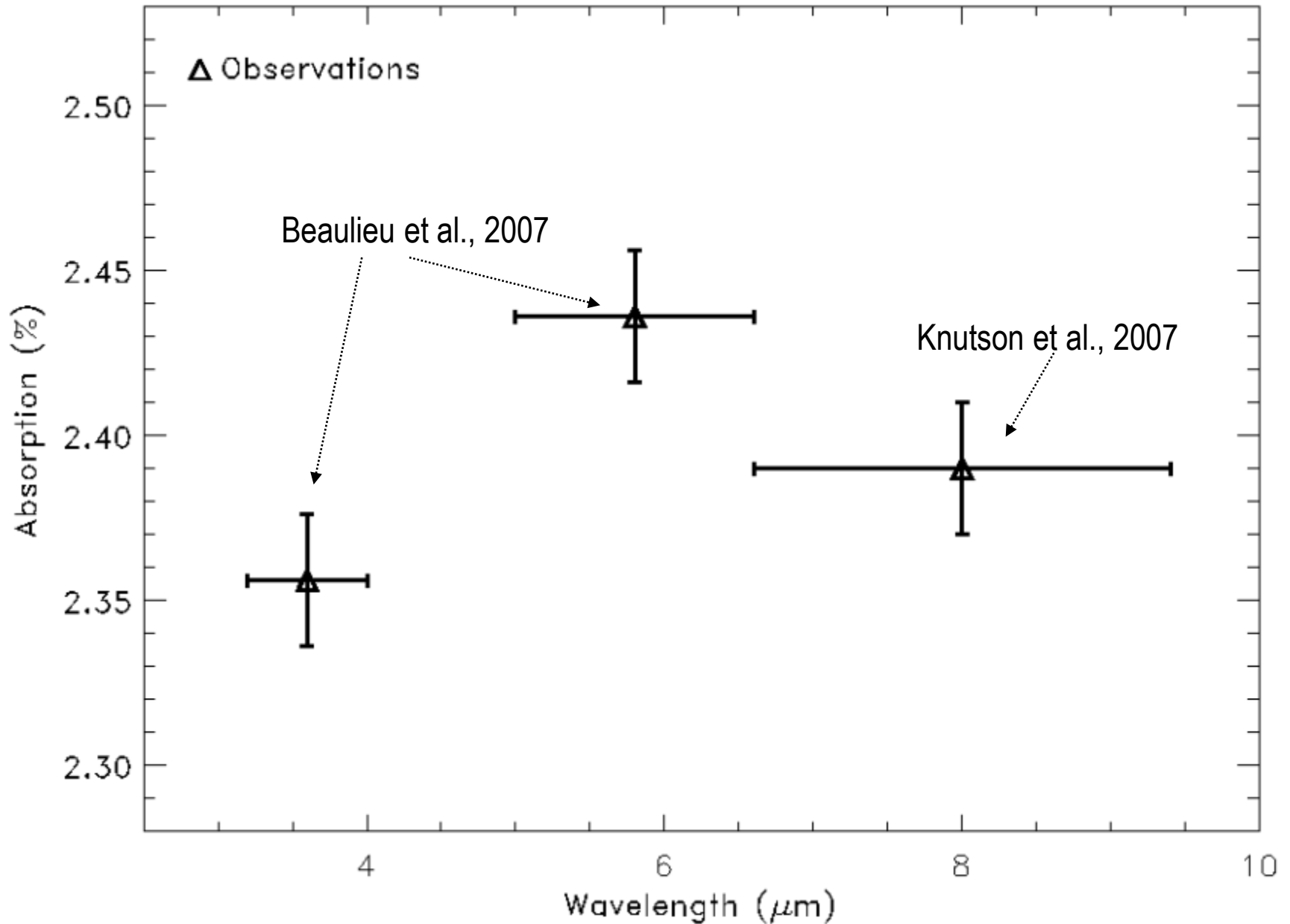
Mass =  $0.69 \pm 0.05 M_{\text{Jupiter}}$

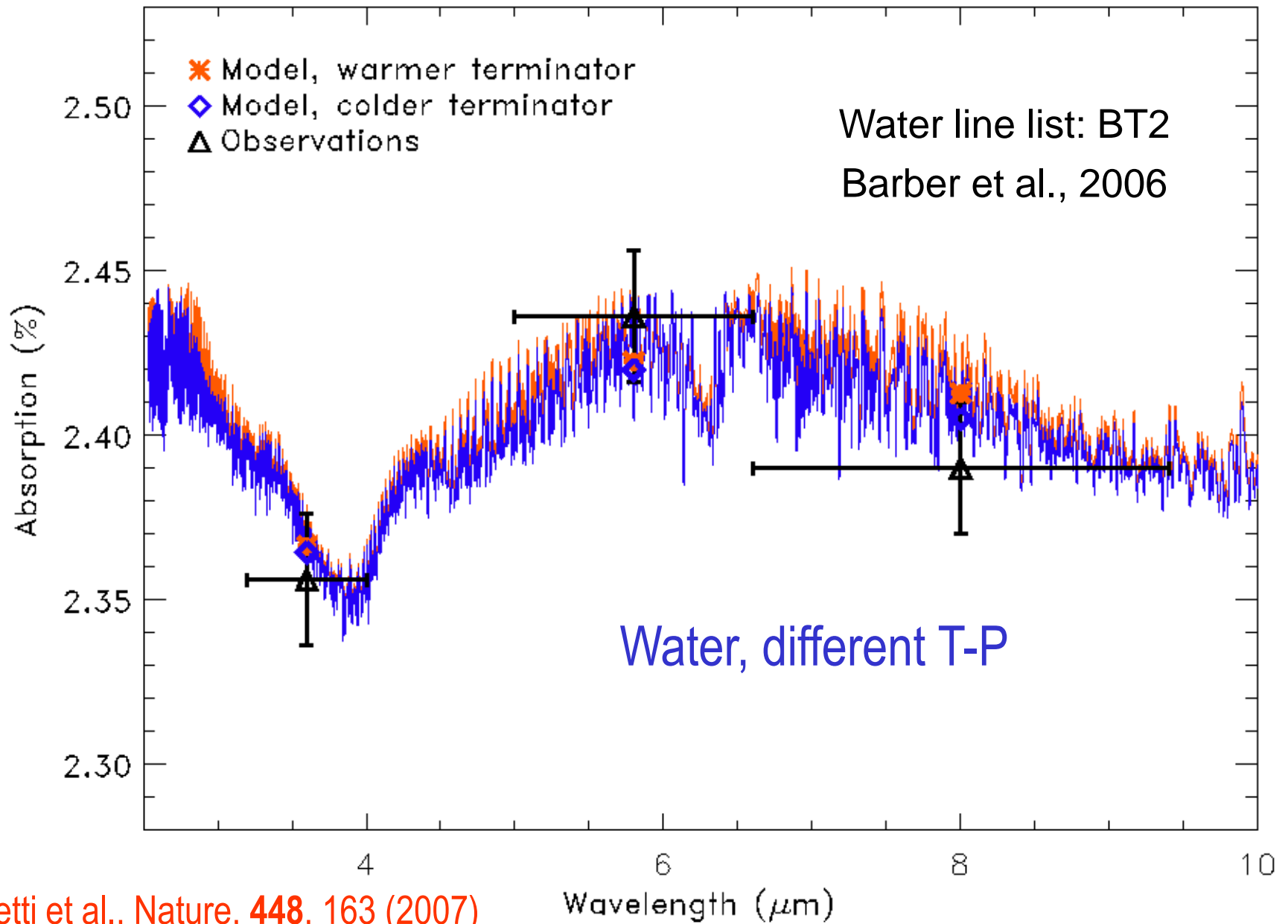


Radius =  $1.35 \pm 0.04 R_{\text{Jupiter}}$

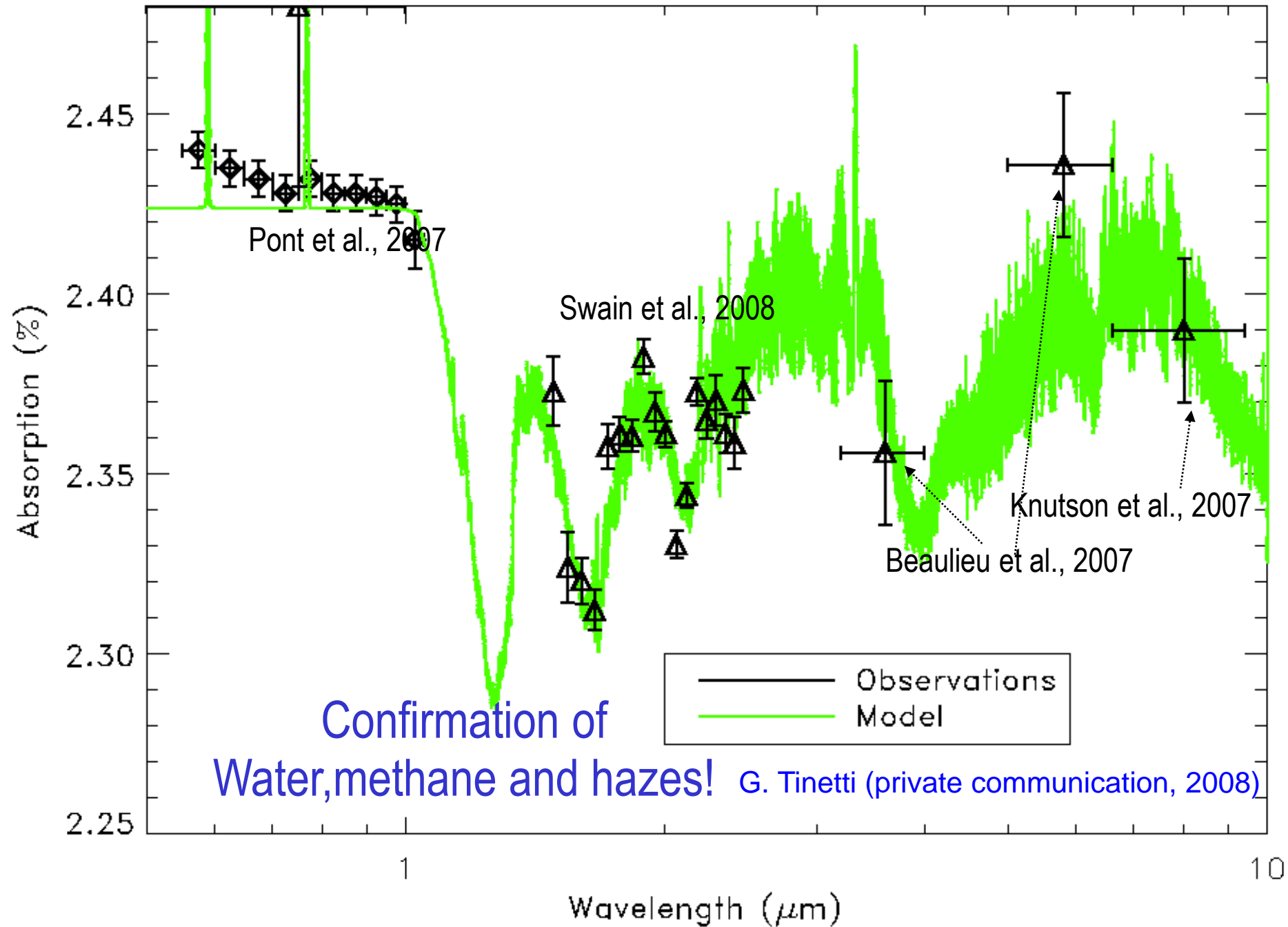
Density =  $0.35 \pm 0.05 \text{ g/cm}^3$

# HD189733b: Primary transit with Spitzer

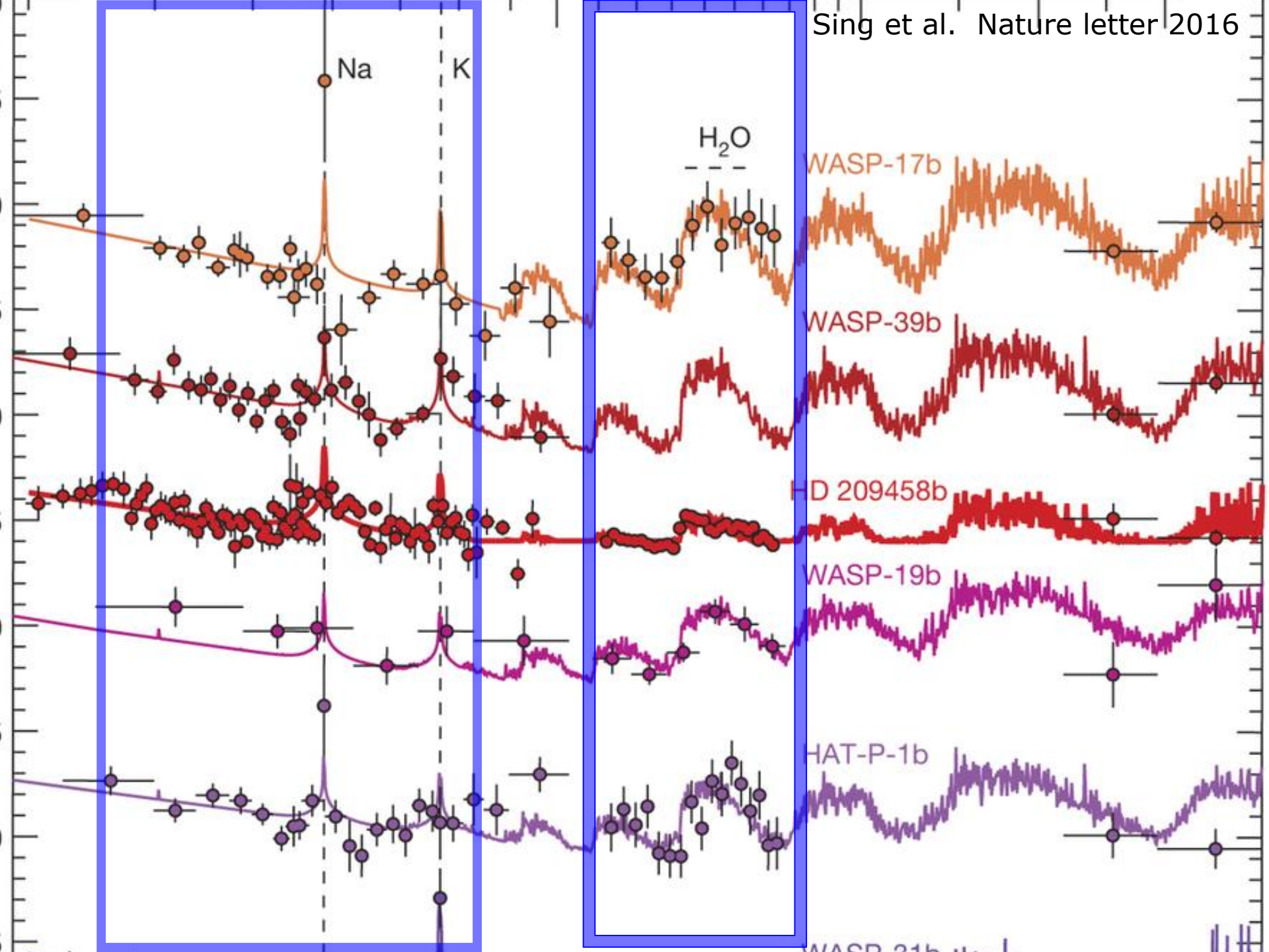




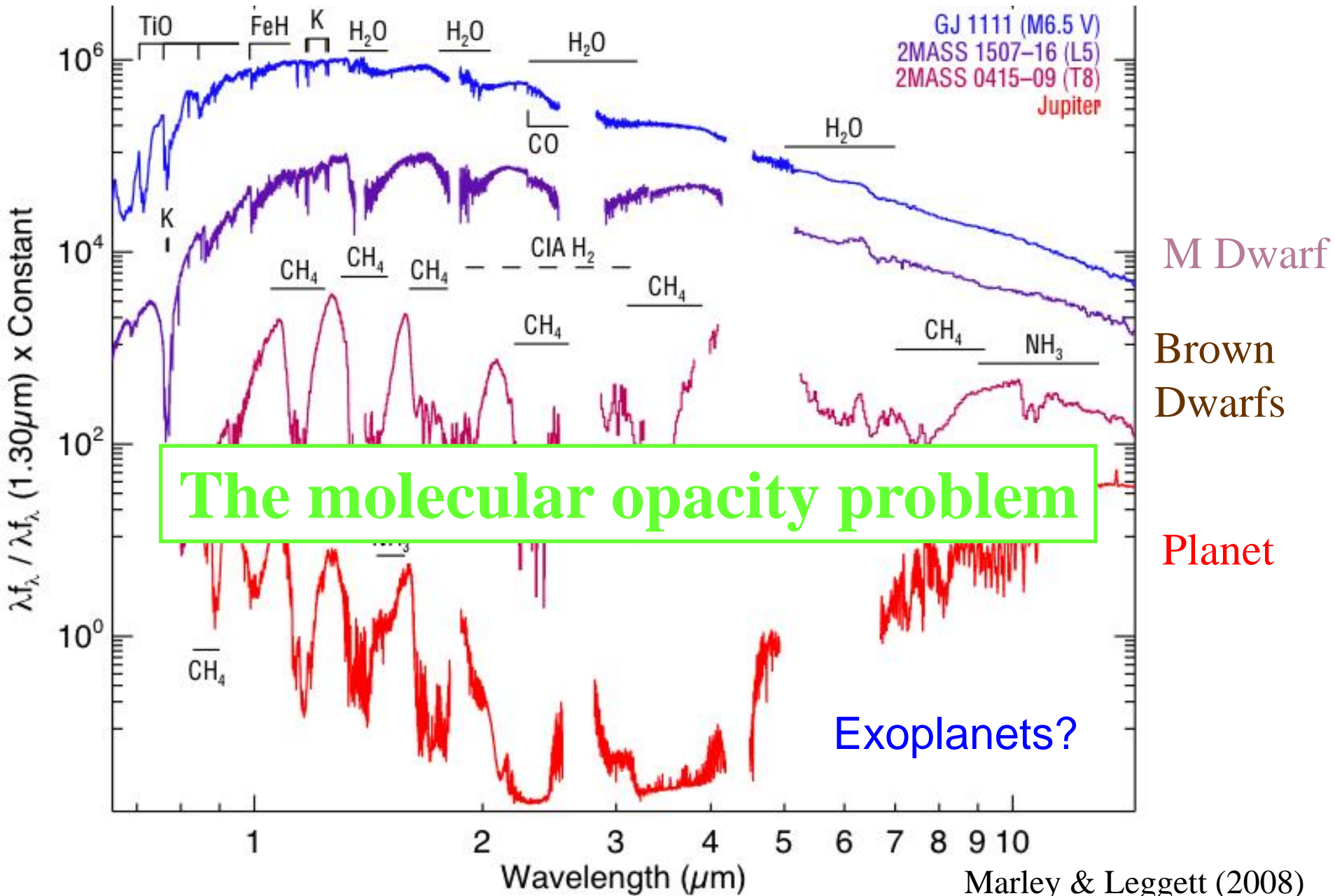
Tinetti et al., Nature, **448**, 163 (2007)







# Cool atmospheres: dominated by molecular absorption



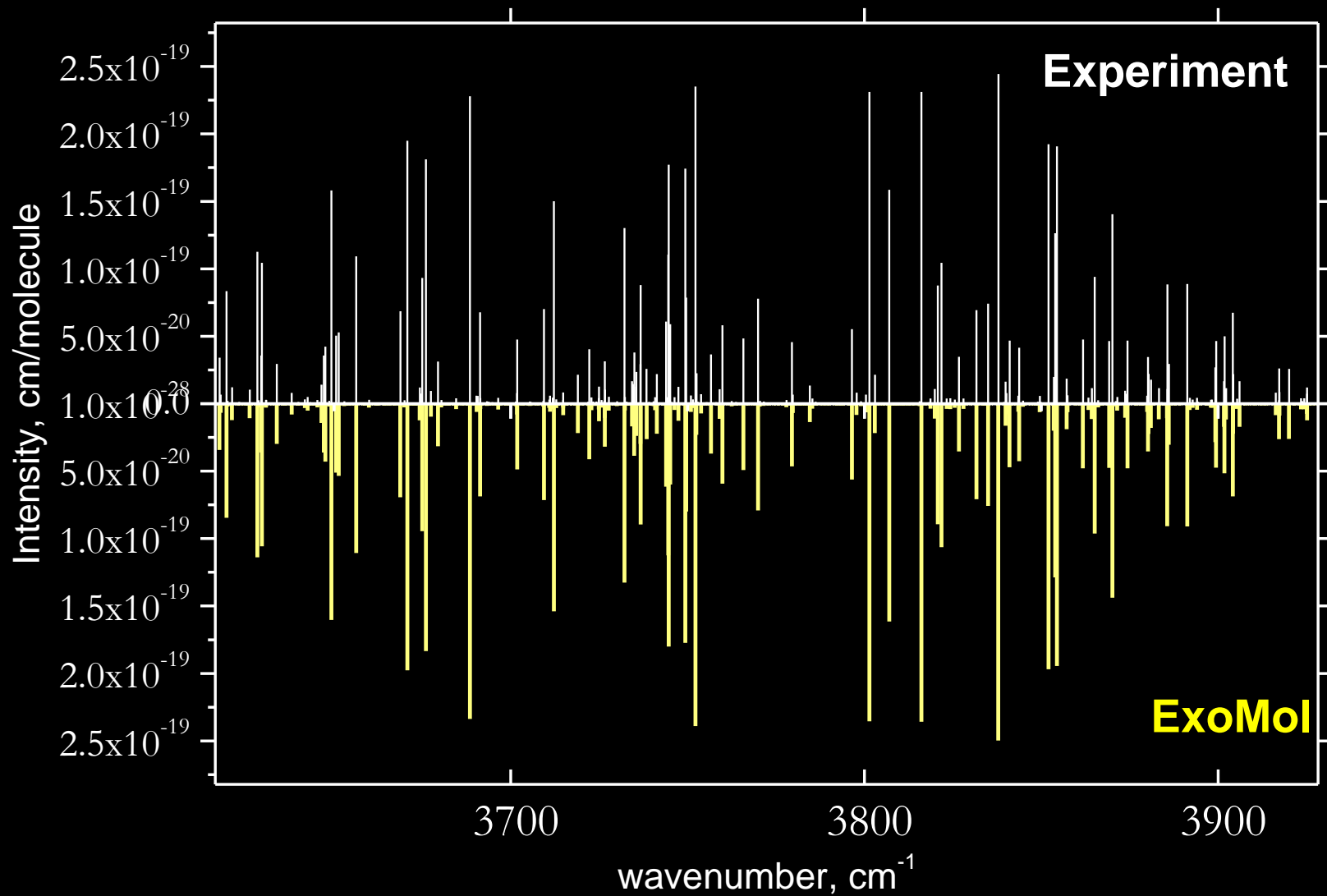
# ExoMol

- 5 year project: 2011-16
- Provide data for all molecular transitions important for exoplanet atmospheres
- Methodology: first principles quantum mechanical calculations, informed by experiment

J Tennyson and S.N. Yurchenko, MNRAS, 425, 21 (2012).



Line list = line positions + line intensities



This is our 2016-line list for water

# Frontier Problems in Exoplanet Characterization

- Non-equilibrium processes in exoplanet atmospheres

(Stevenson et al. 2010; Madhusudhan & Seager 2011; Moses et al. 2013)  $\text{CH}_4, \text{CO}, \text{NH}_3$

- Constraints on thermal inversions in hot Jupiters

(Fortney et al. 2008; Spiegel et al. 2009)  $\text{TiO}, \text{VO}, \text{H}_2\text{S}$

- C/O ratios and Carbon-rich atmospheres

(Fortney et al. 2008; Spiegel et al. 2009)  $\text{H}_2\text{O}, \text{CO}, \text{HCN}, \text{CH}_4,$   
 $\text{C}_2\text{H}_2, \text{TiH}, \text{FeH}$

- Constraints on exoplanet formation conditions

(Madhusudhan et al. 2011; Oberg et al. 2011)  $\text{H}_2\text{O}, \text{CO}, \text{CH}_4$

- Atmospheres and interiors of super-Earths

(Bean et al. 2011; Desert et al. 2011; Miller-Ricci Kempton et al. 2011)  $\text{H}_2\text{O}, \text{CO}_2$

# ExoMol list of molecules

## Molecular line lists for exoplanet & other atmospheres

	Primordial (Metal-poor)	Terrestrial Planets (Oxidising)	Giant-Planets & Cool Stars (Reducing atmospheres)
Already available	H <sub>2</sub> , LiH <b>HeH<sup>+</sup>, H<sub>3</sub><sup>+</sup></b> <b>H<sub>2</sub>D<sup>+</sup></b>	<b>OH, CO<sub>2</sub>, O<sub>3</sub>, NO</b> <b>H<sub>2</sub>O, HDO, NH<sub>3</sub></b>	<b>H<sub>2</sub>, CN, CH, CO, CO<sub>2</sub>, TiO</b> <b>HCN/HNC, H<sub>2</sub>O, NH<sub>3</sub>,</b>
ExoMol		<b>O<sub>2</sub>, CH<sub>4</sub>, SO<sub>2</sub>, SO<sub>3</sub></b> <b>HOOH, H<sub>2</sub>CO,</b> <b>HNO<sub>3</sub></b>	<b>CH<sub>4</sub>, PH<sub>3</sub>, C<sub>2</sub>, C<sub>3</sub>, HCCH, H<sub>2</sub>S,</b> <b>C<sub>2</sub>H<sub>6</sub>, C<sub>3</sub>H<sub>8</sub>, VO, O<sub>2</sub>, AlO, MgO,</b>  <b>CrH, MgH, FeH, CaH, AlH, SiH,</b> <b>TiH, NiH, BeH, YO</b>
Available from elsewhere <b>Already calculated at UCL</b> Will be calculated during the ExoMol project			

Full details:

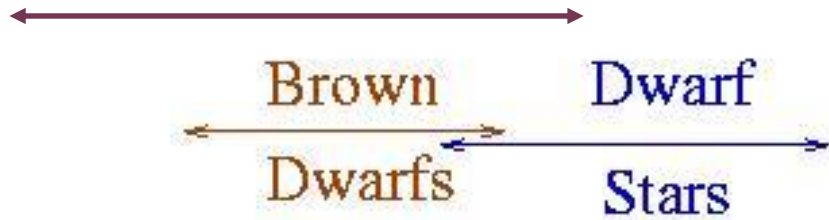
[www.exomol.com](http://www.exomol.com)

J. Tennyson and S.N. Yurchenko, MNRAS, 425, 21 (2012)

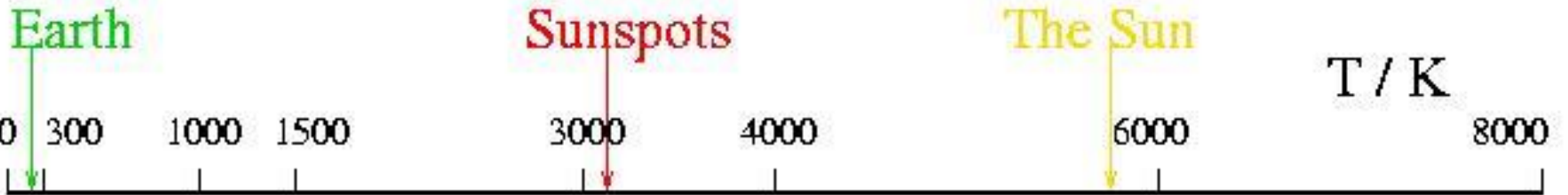
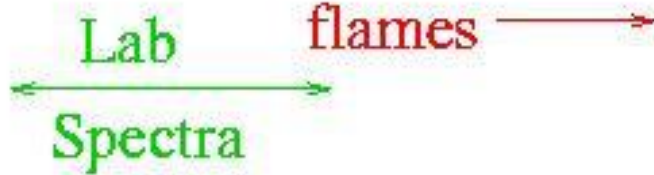


Why theory, not experiment?

Exoplanets



The Molecular Thermometer



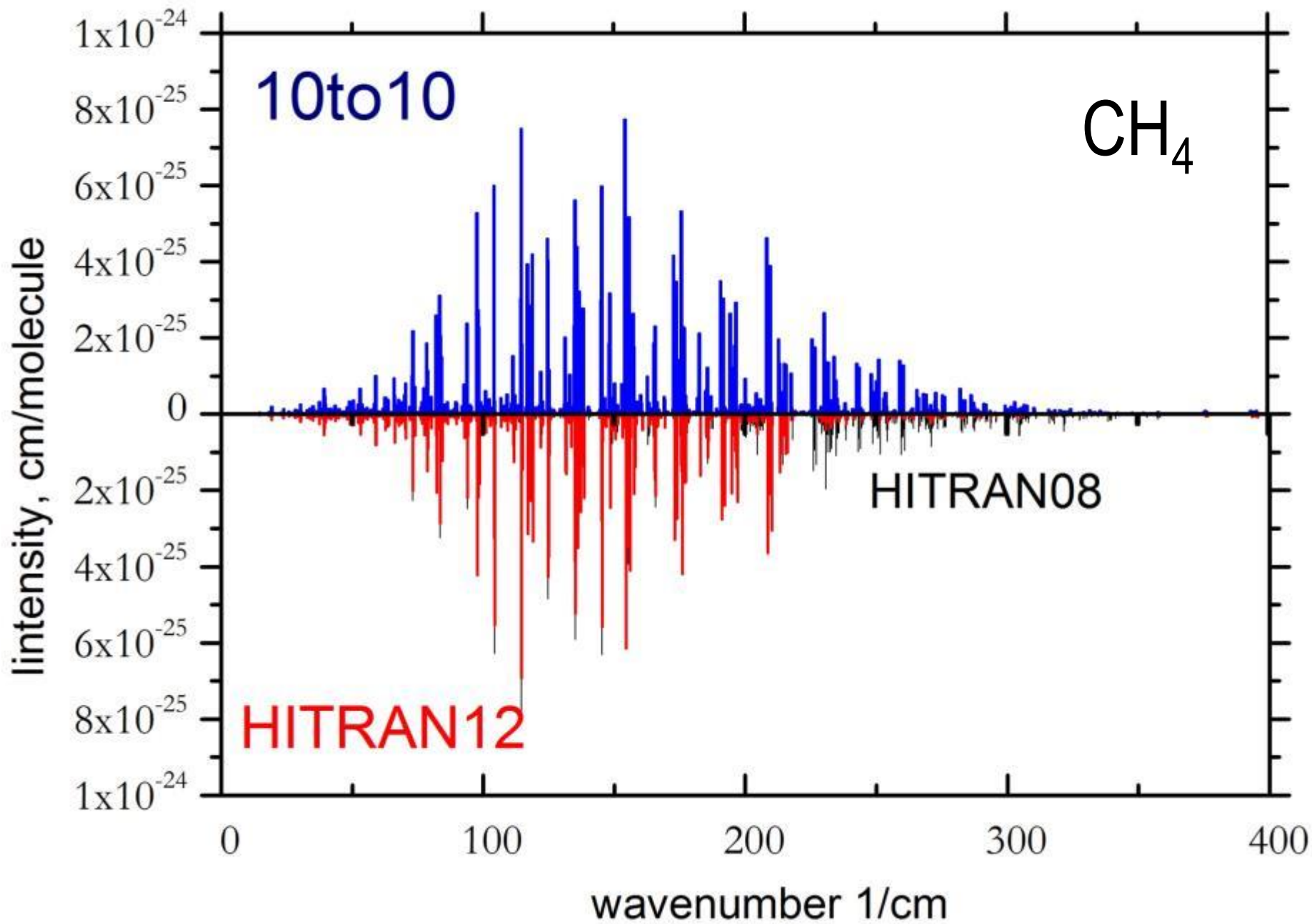
hitran

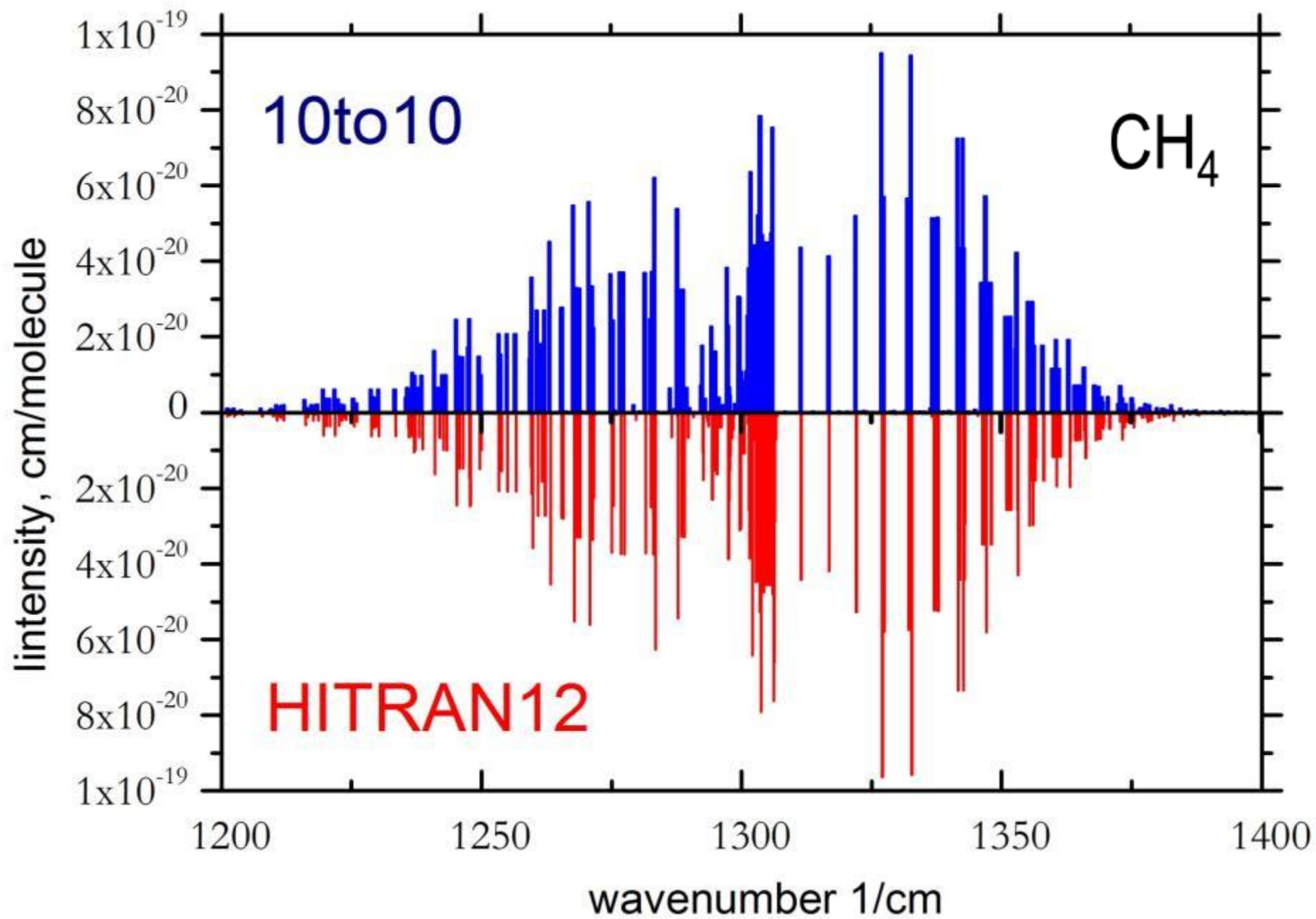
hitemp →

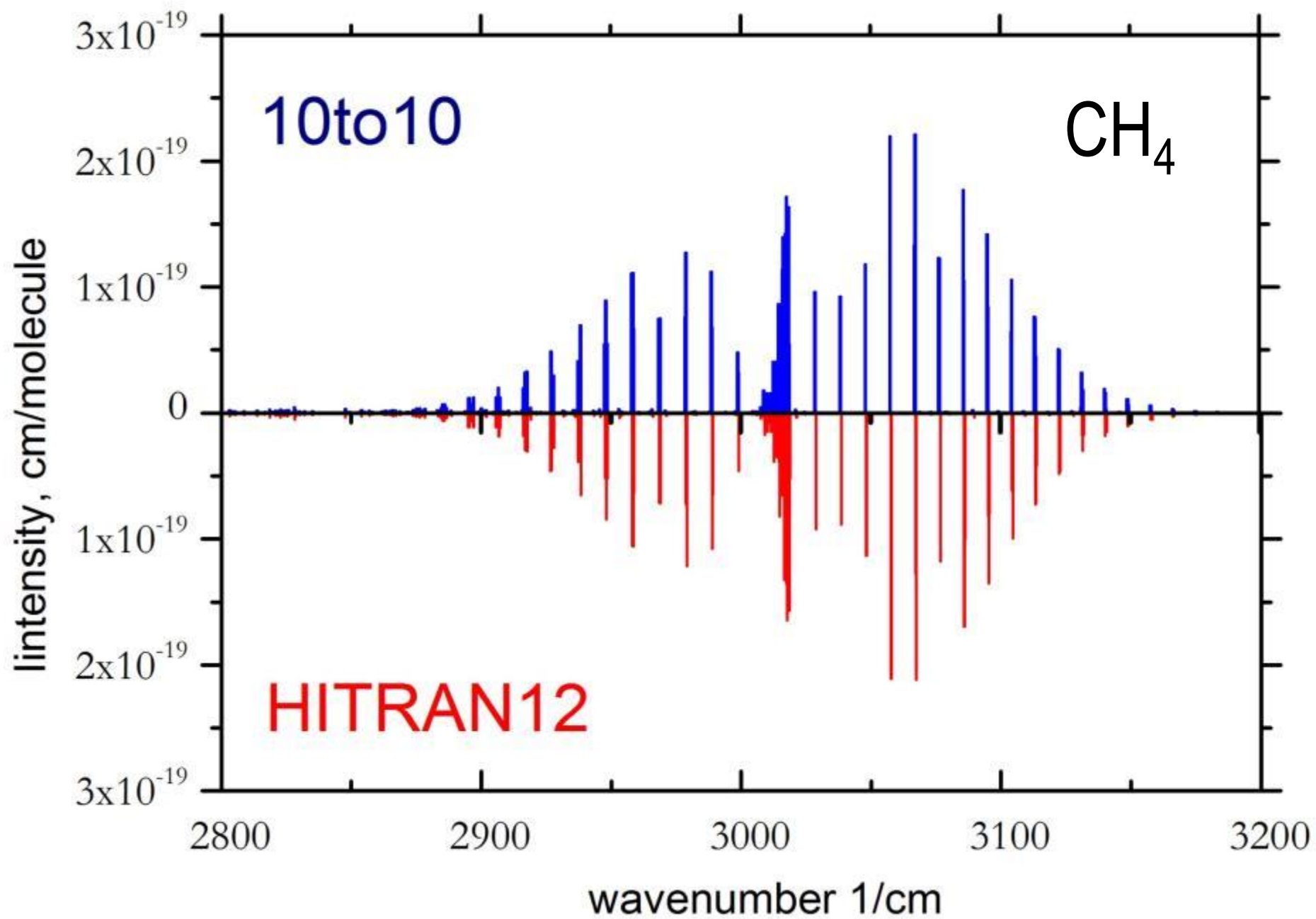
diatomic molecules →

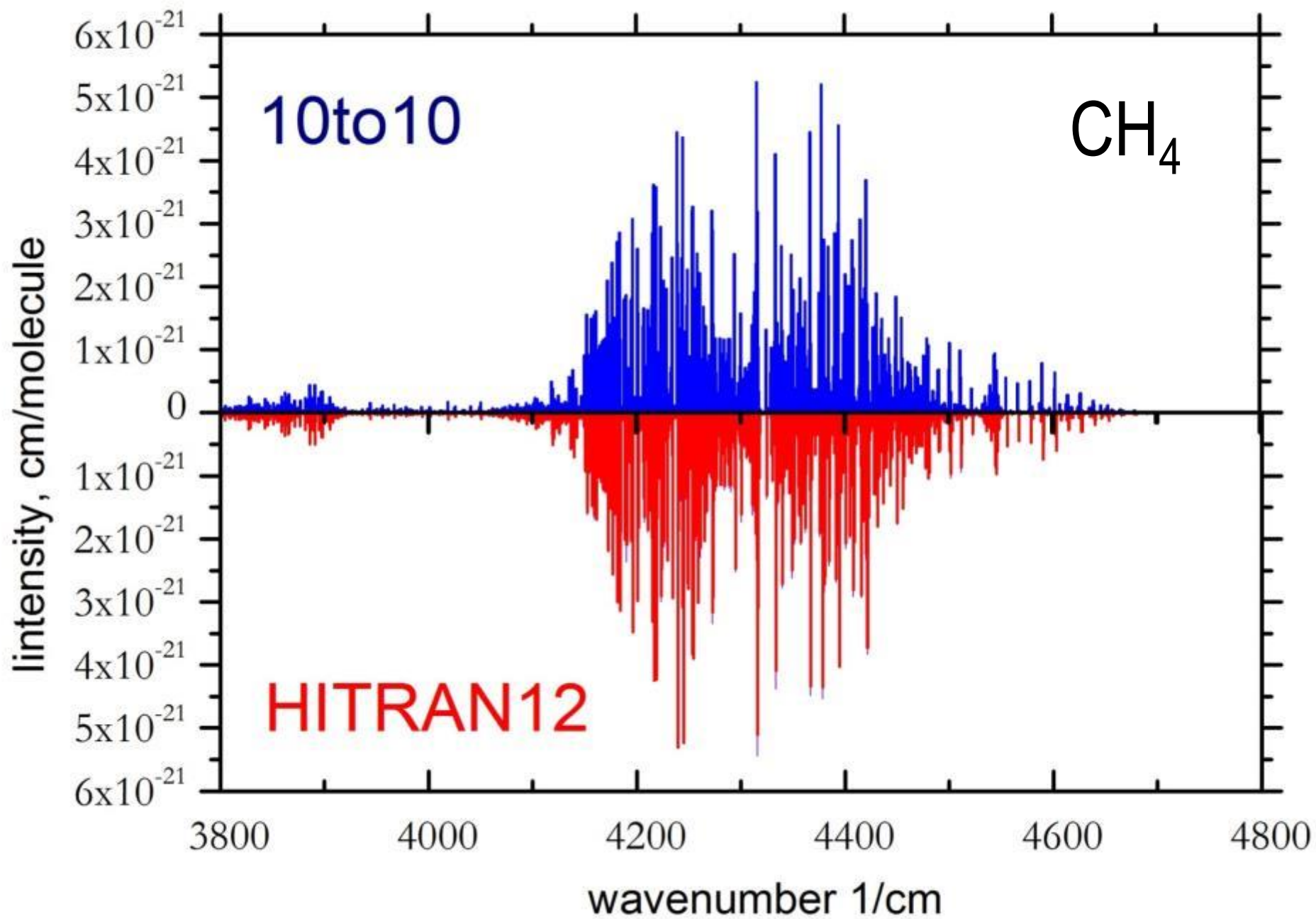
← H<sup>+</sup> →

polyatomic molecules →

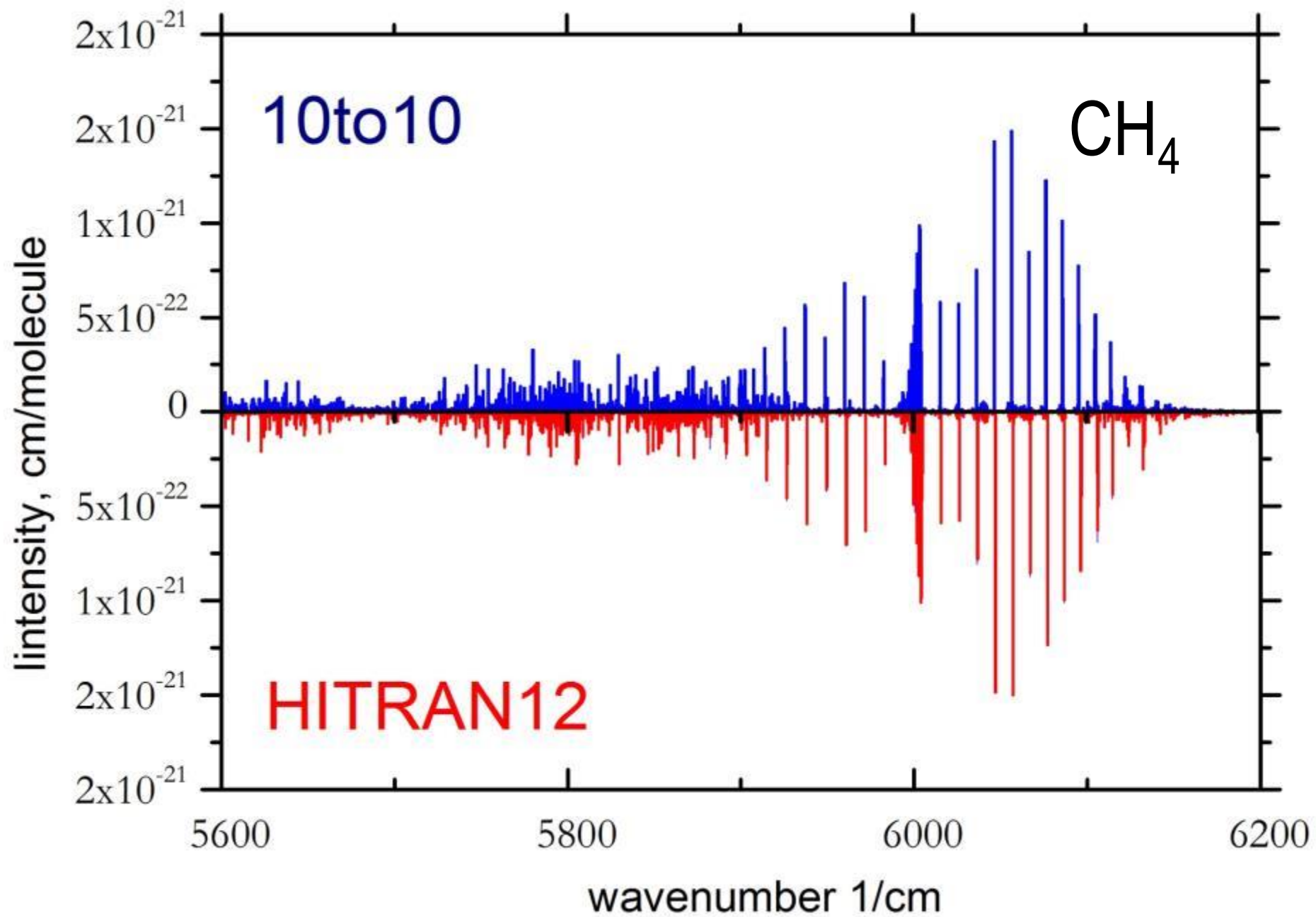




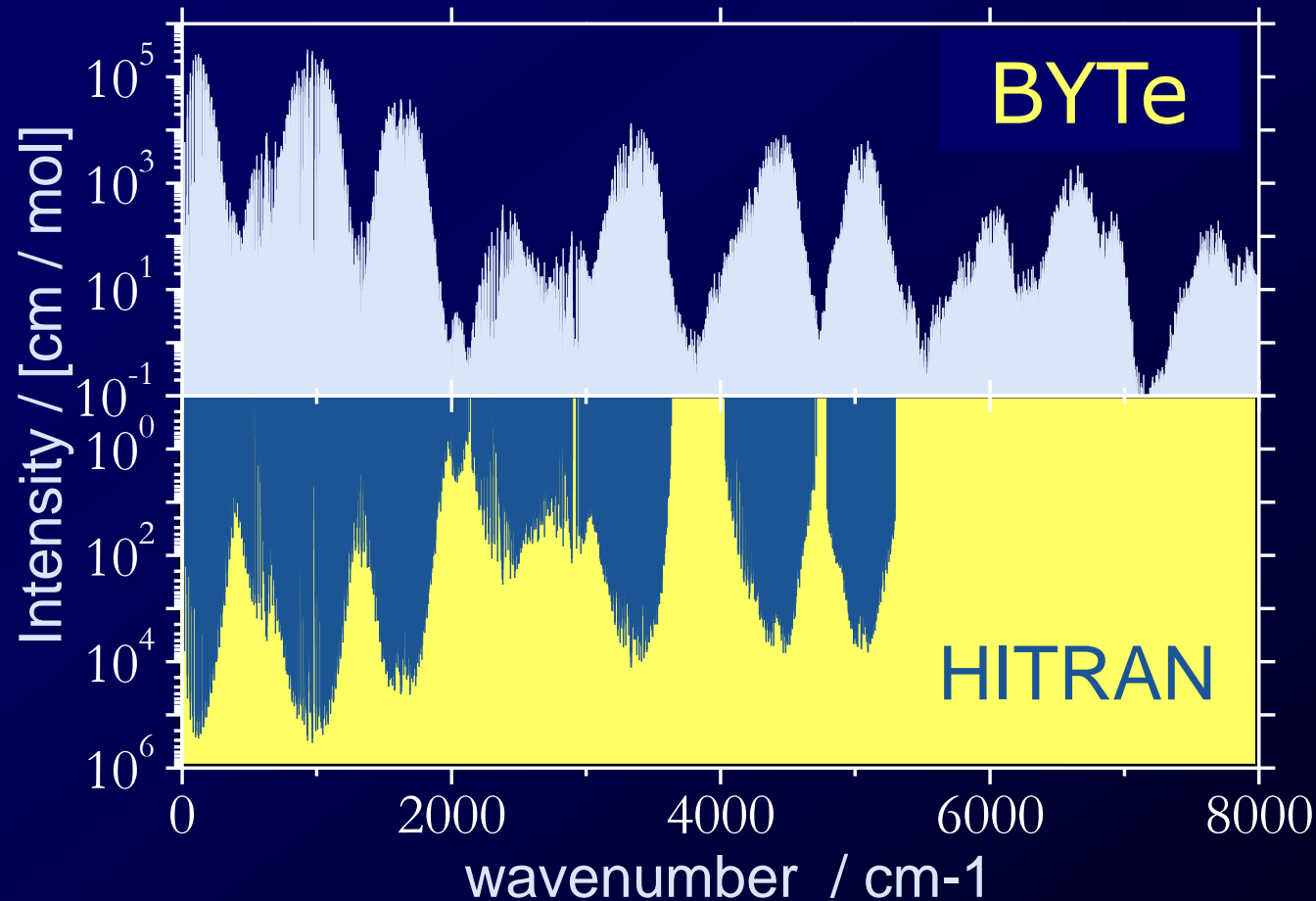






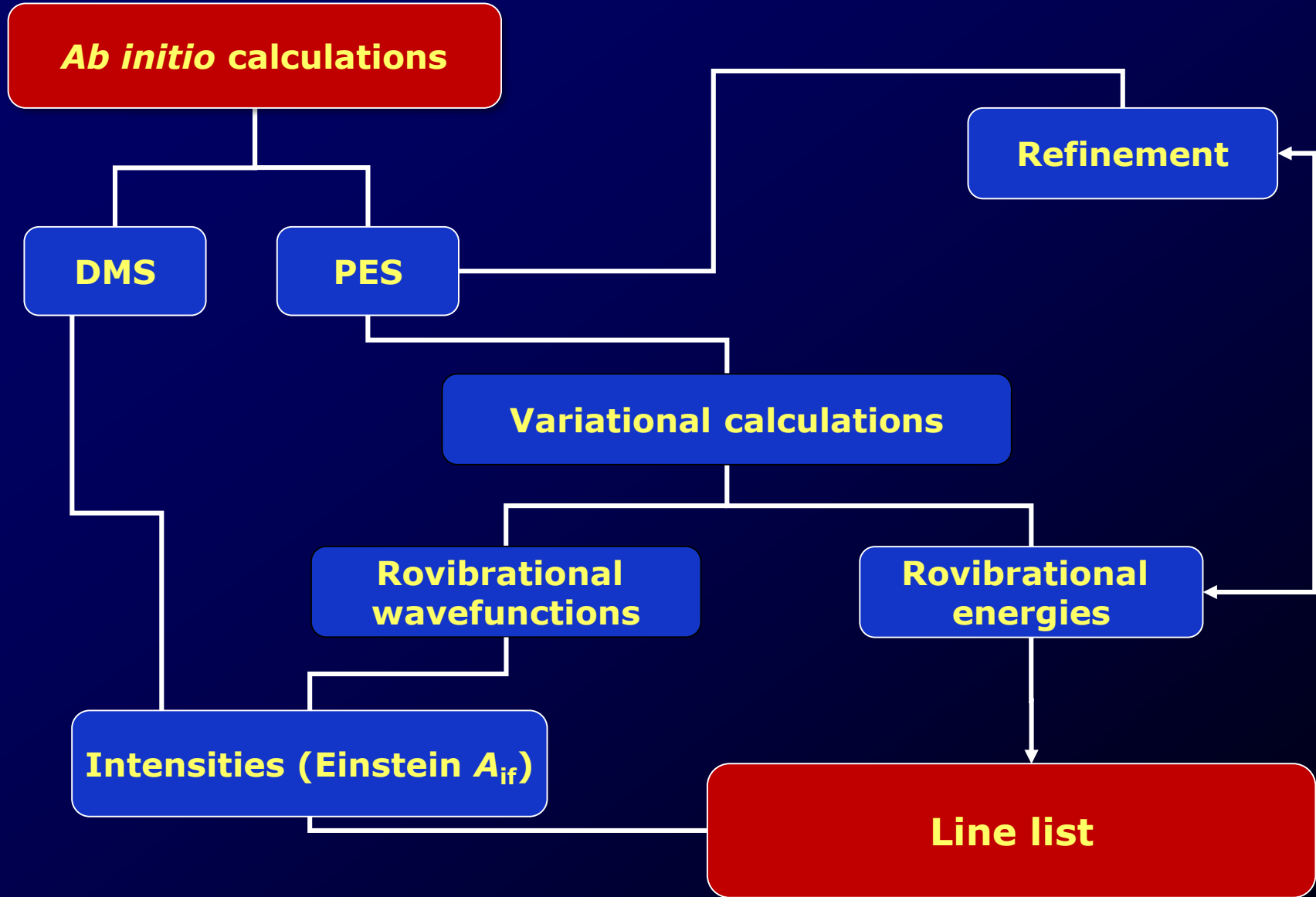


# Completeness: Absorption of ammonia (T=300 K)



Less than 30,000  $\text{NH}_3$  lines known experimentally:  
BYTe contains 1.1 billion lines, about 40,000 times as many!

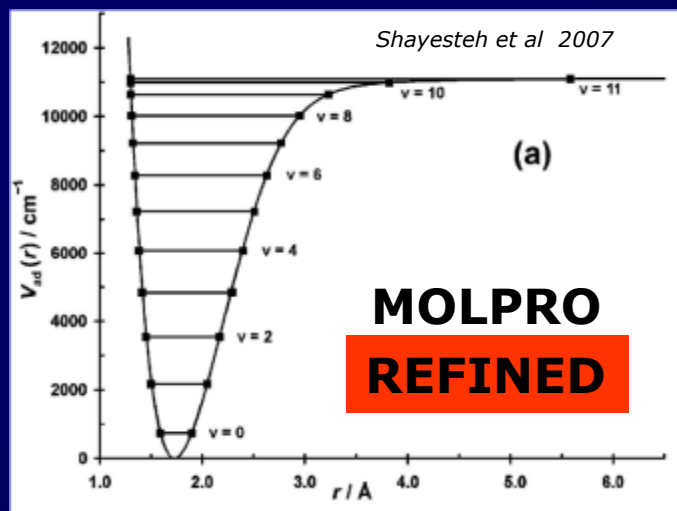
# Method: Spectrum from the “first-principles”



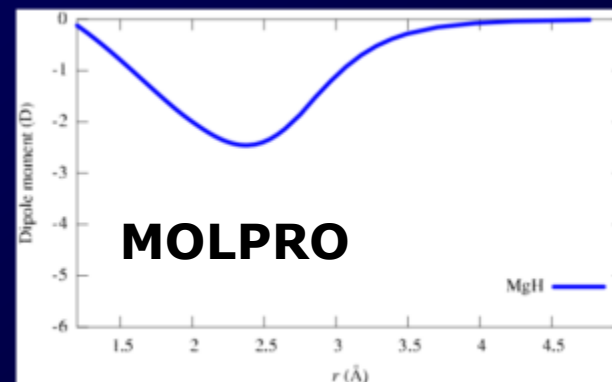
**Ab initio: solve for motion of electrons**

Line list: MgH

**Potential energy curve**



**Dipole moment curve**

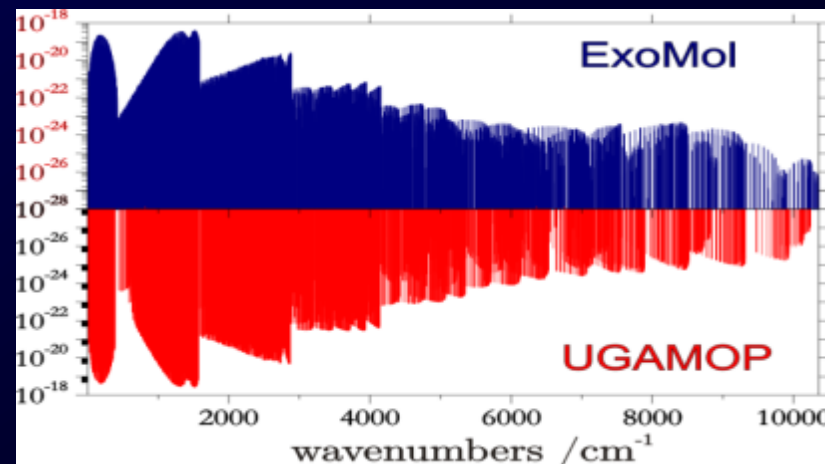


**Line list: 6690 lines,  $N_{\text{max}}=60$**

**Solve for the motion  
of the nuclei**

**LEVEL 8.0**

R. Le Roy,  
Waterloo, Canada



B Yadin et al, MNRAS 425, 34 (2012)

# New general diatomic code

# Duo

Freely available from CCPforge  
[ccpforge.cse.rl.ac.uk](http://ccpforge.cse.rl.ac.uk)

Computer Physics Communications 202 (2016) 262–275

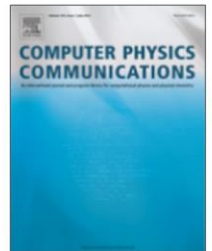
Contents lists available at [ScienceDirect](http://ScienceDirect)

## Computer Physics Communications

journal homepage: [www.elsevier.com/locate/cpc](http://www.elsevier.com/locate/cpc)



ELSEVIER



## Duo: A general program for calculating spectra of diatomic molecules<sup>☆</sup>

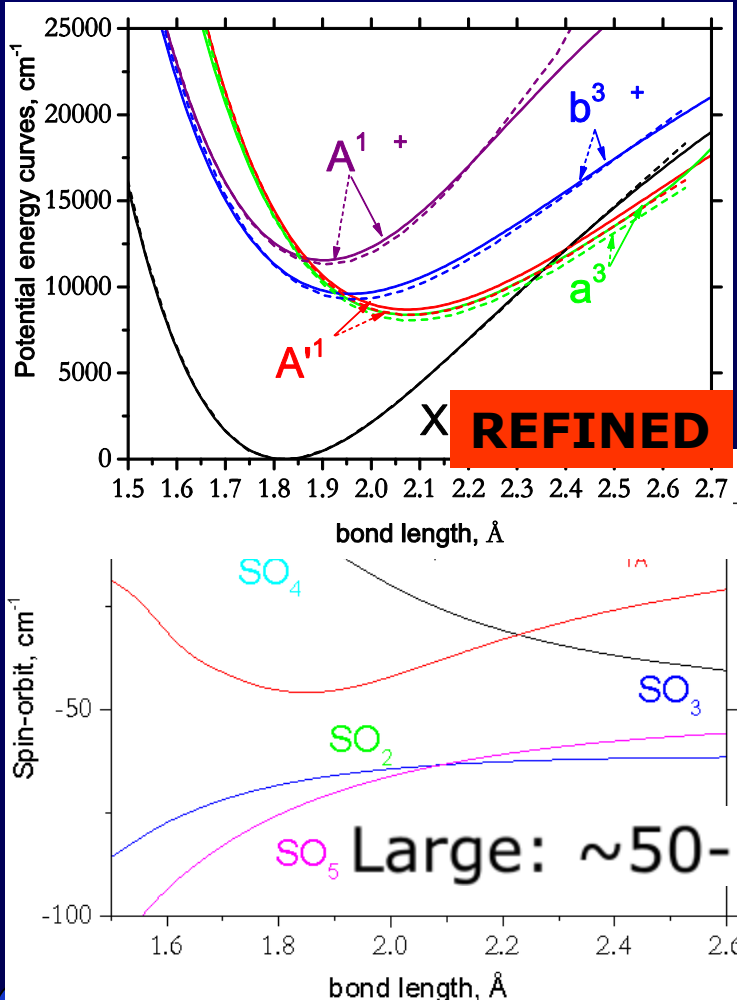


Sergei N. Yurchenko<sup>a,\*</sup>, Lorenzo Lodi<sup>a</sup>, Jonathan Tennyson<sup>a</sup>, Andrey V. Stolyarov<sup>b</sup>

<sup>a</sup> Department of Physics & Astronomy, University College London, Gower Street, London WC1E 6BT, United Kingdom

<sup>b</sup> Department of Chemistry, Lomonosov Moscow State University, Leninskiye gory 1/3, 119992 Moscow, Russia

## Potential energy



Solve for the motion of the nuclei

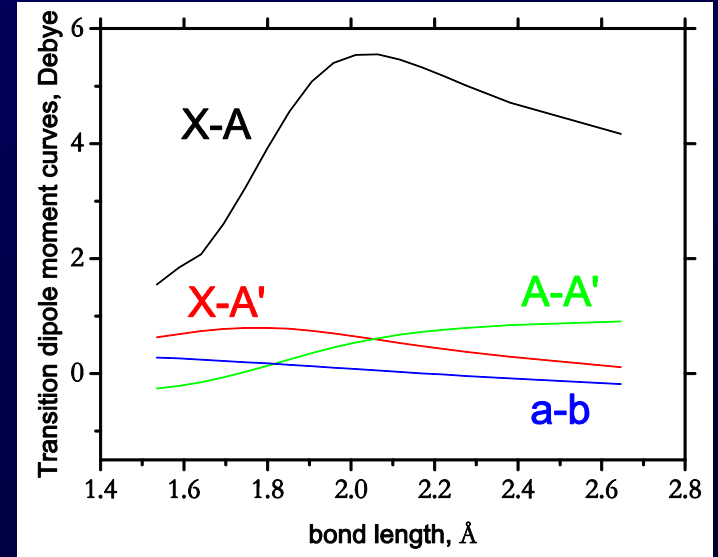
*Duo*

SN Yurchenko et al, MNRAS 456, 4524 (2016)

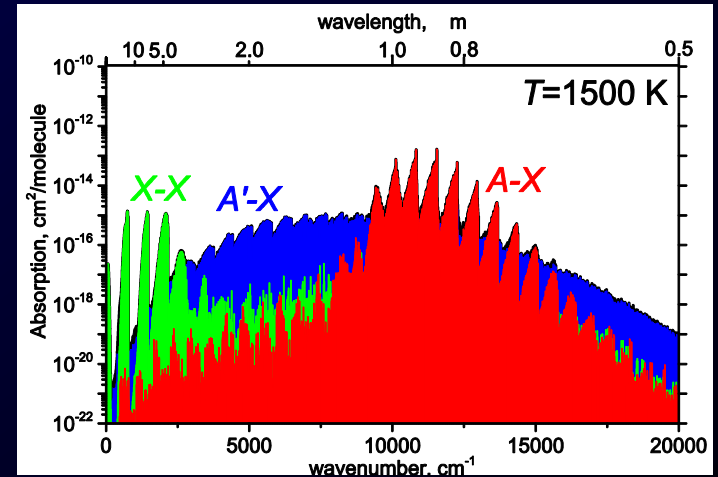
## Line list: CaO

Khalil et al (2011)

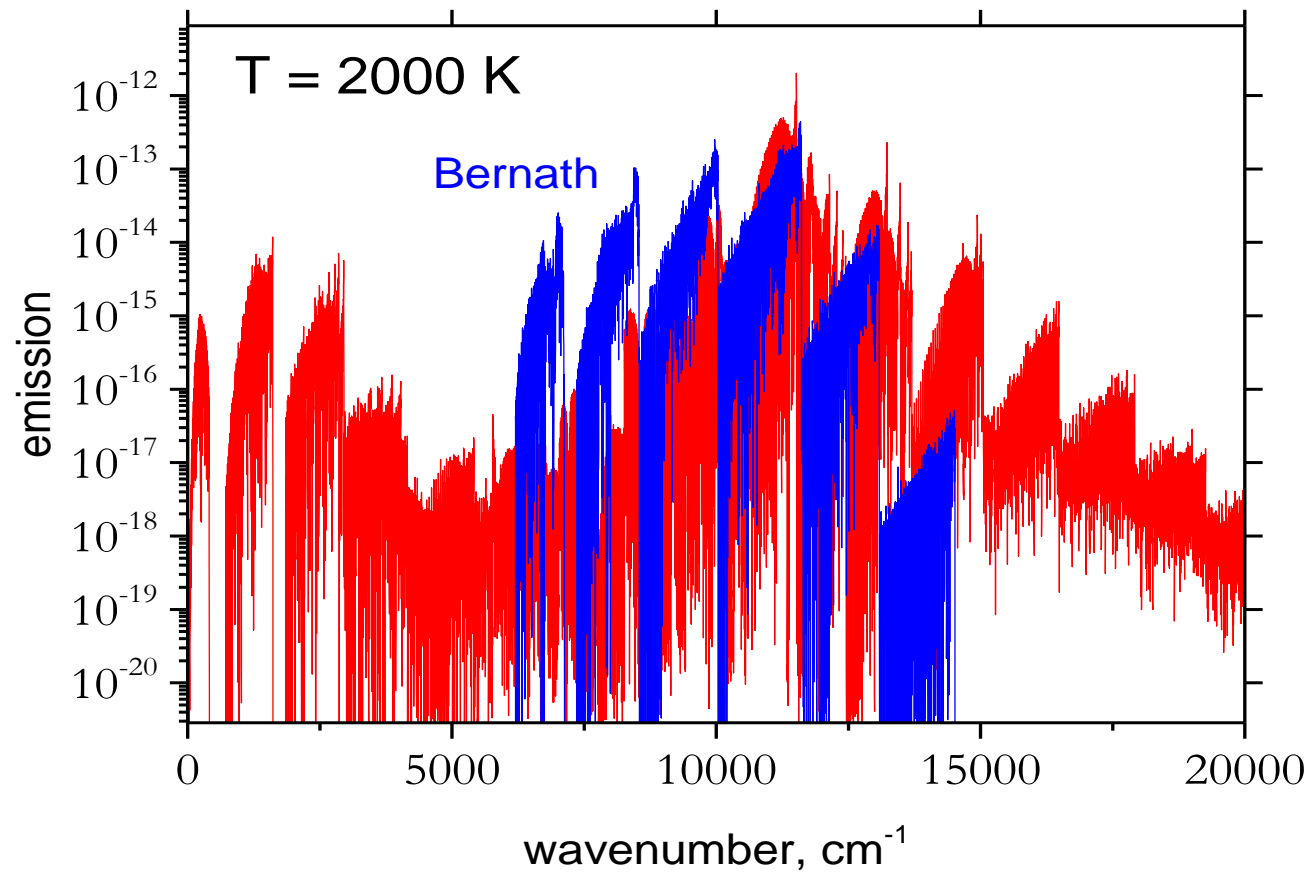
## Dipole moment



## Line list: 22 M lines



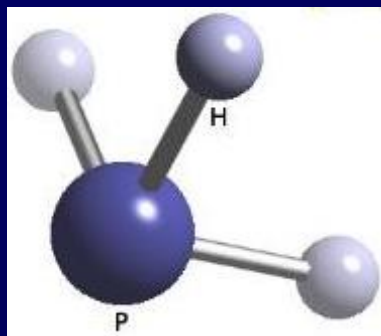




**CrH**

Maire N. Gorman, PhD UCL (2016)

## Potential energy



## Dipole moment



Ab initio PES  
[CCSD(T)/aug-cc-pV(Q+d)Z]

R. I. Ovsyannikov et al.  
J. Chem. Phys 129, 044309 (2008).

Refined using lab spectra

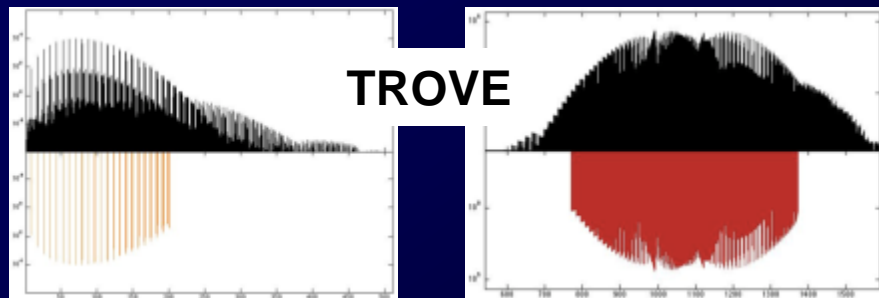
**Tunneling  
motion  
neglected**

Ab initio:  
CCSD(T)/aug-cc-pVTZ

S.N. Yurchenko et al.  
J. Mol. Spectrosc 239, 71 (2006).

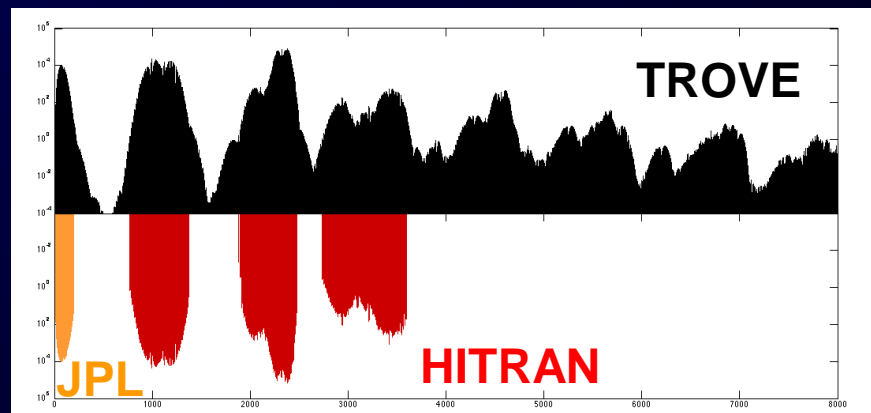
## Solve for the motion of the nuclei

**TROVE:** Yurchenko, Thiel, Jensen



JPL

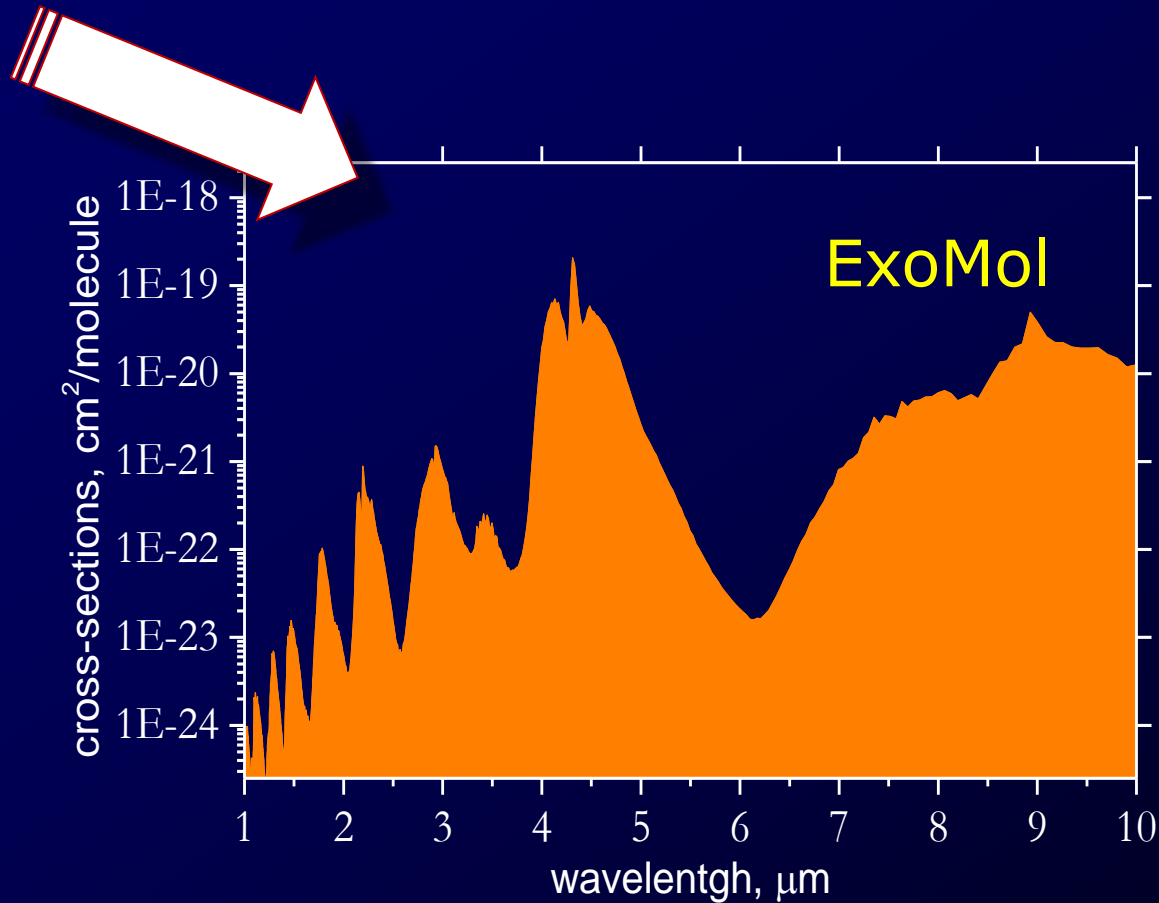
HITRAN



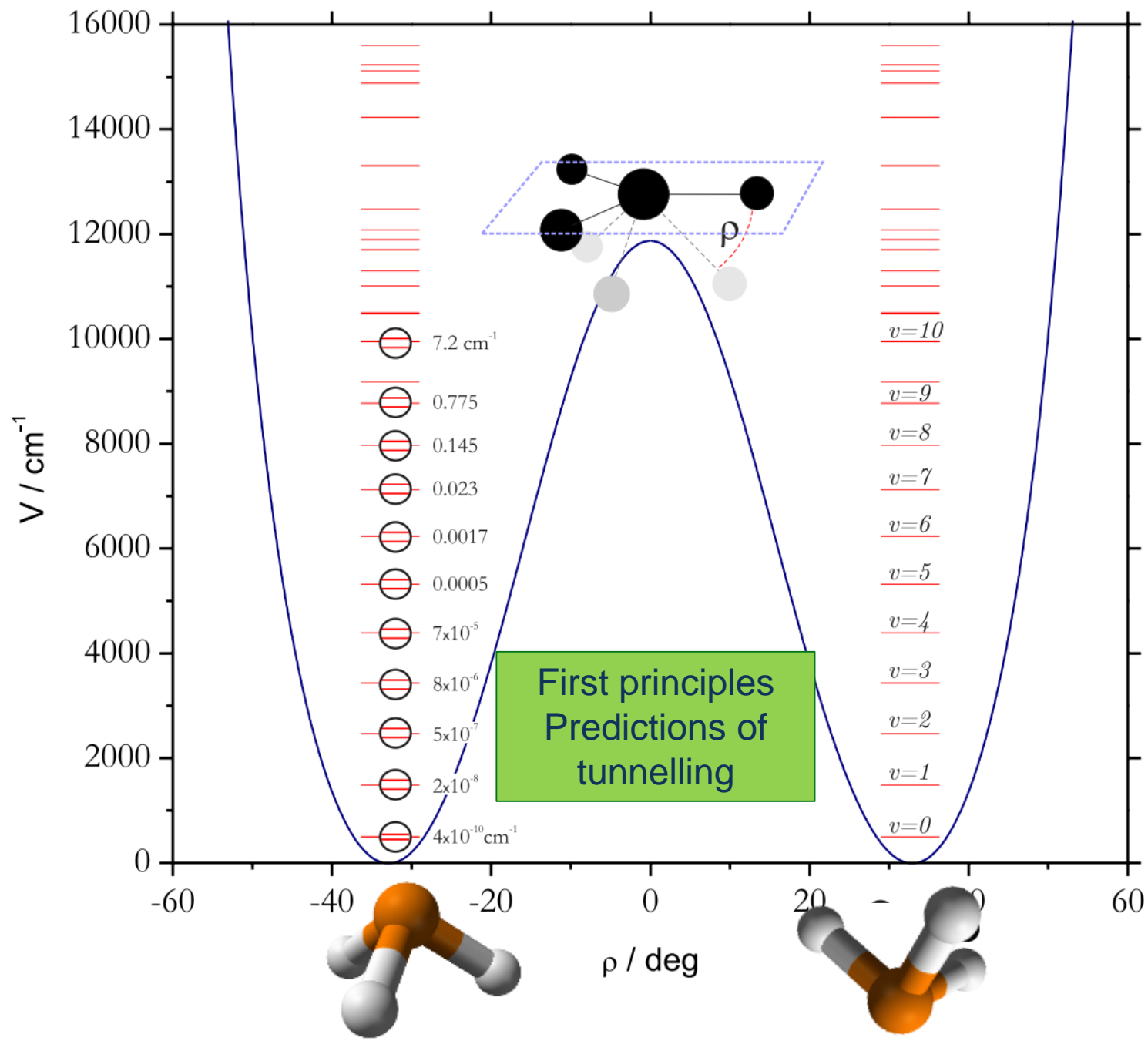
**16.8 billion transitions for  
T up to 1500 K**

C Sousa-Silva et al, MNRAS, 446, 2337 (2015)

It has a nice strong feature at 4.5  $\mu\text{m}$



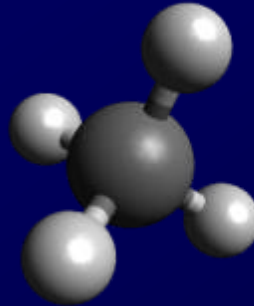
No detection yet of phosphine on exoplanets



**Ab initio: solve for motion of electrons**

CH<sub>4</sub>

Potential energy



Dipole moment

**9D surface**  
**130 000**  
**geometries**

**MOLPRO**  
**CCSD(T)-f12/QZ**

Ab initio  
10 electrons  
Ground  
electronic state

**Three 9D surfaces**  
**130 000**  
**geometries**

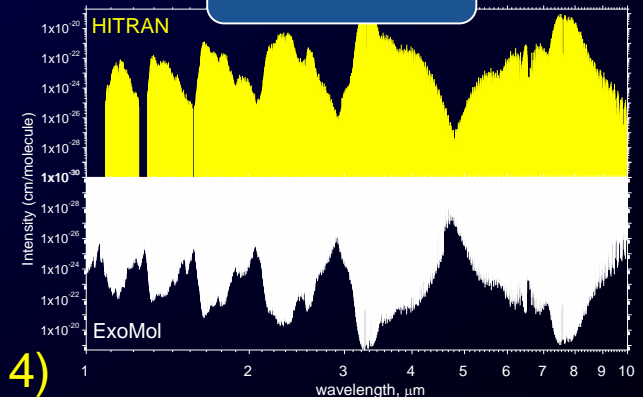
**MOLPRO**  
**CCSD(T)-f12/QZ**

Solve for the motion  
of the nuclei

**TROVE**  
Yurchenko, Thiel, Jensen

**10to10**

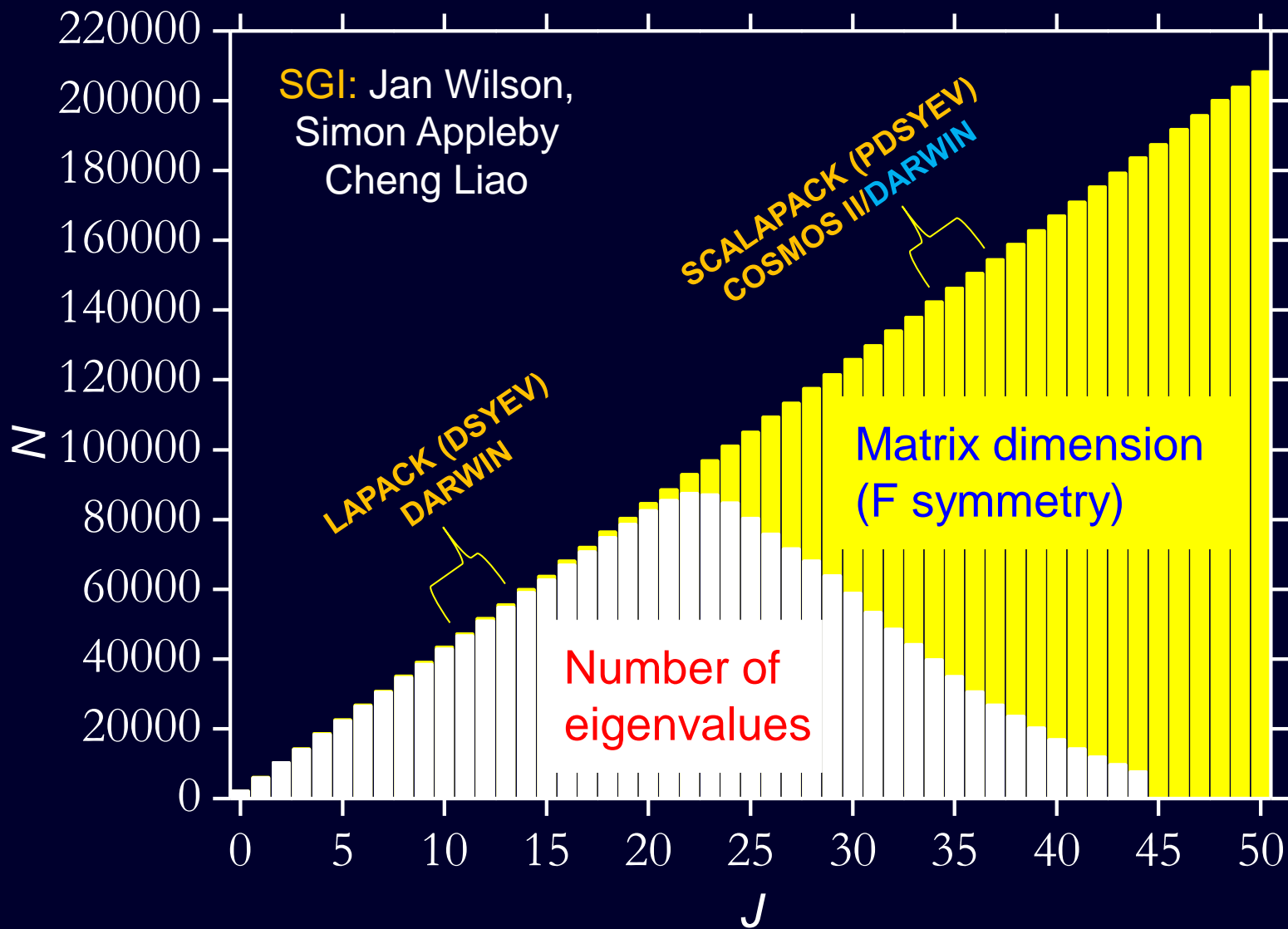
Line list:



**9.8 Billion transitions**

SN Yurchenko & J Tennyson, MNRAS 440, 1649 (2014)

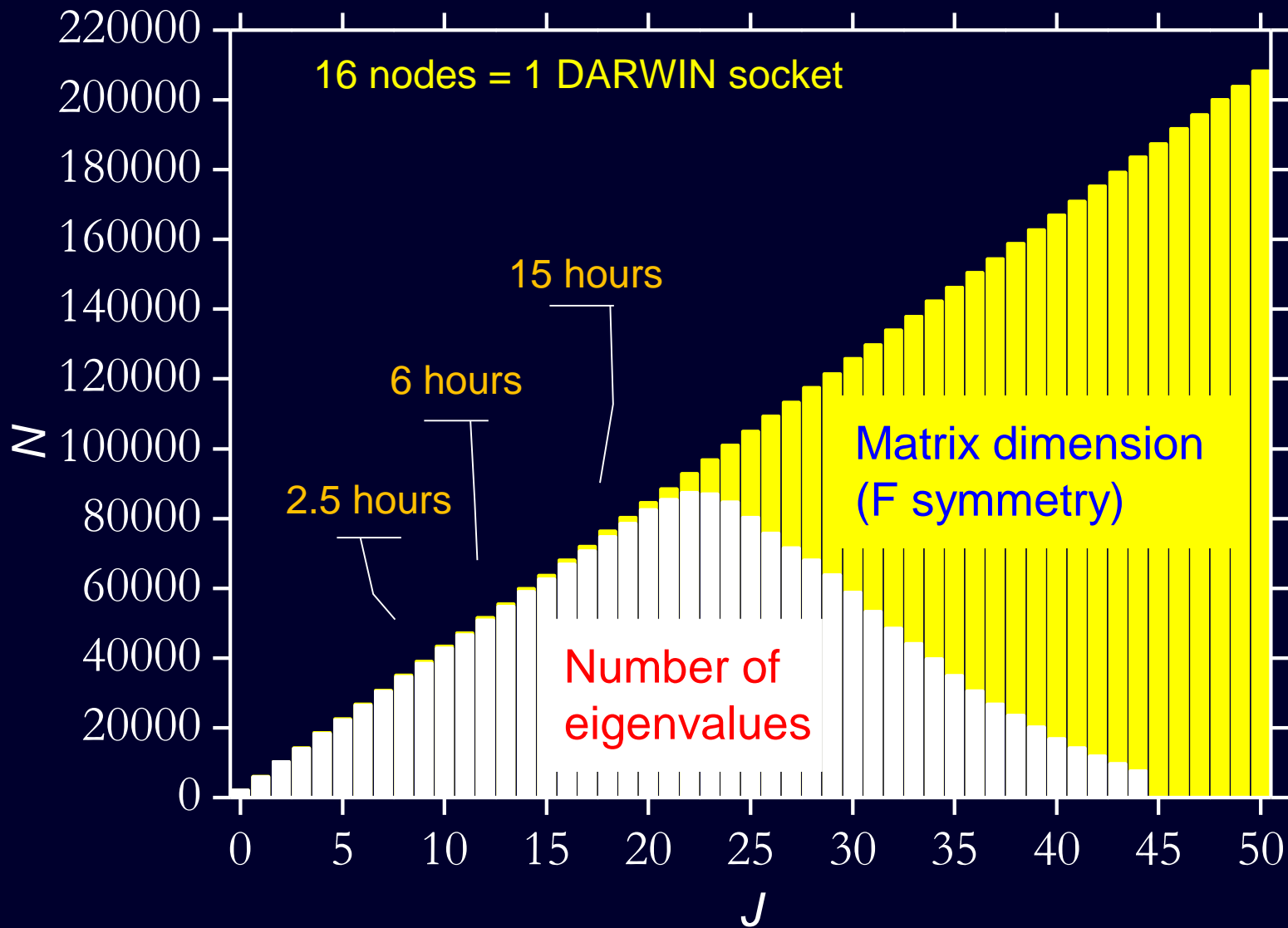
# CH<sub>4</sub> diagonalization: Size of the problem



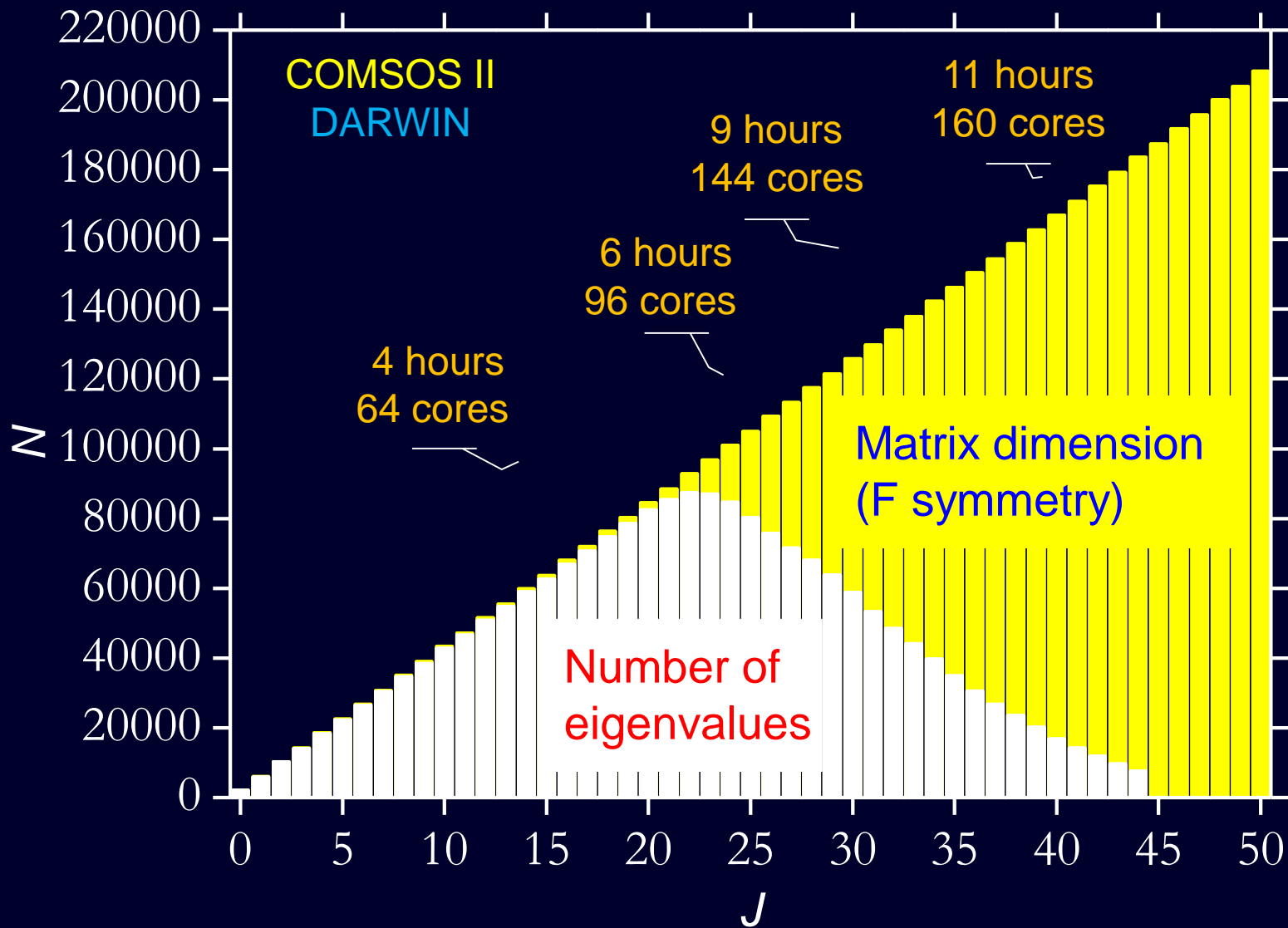
Acknowledgment: Andrey Kaliazin Dirac/COSMOS



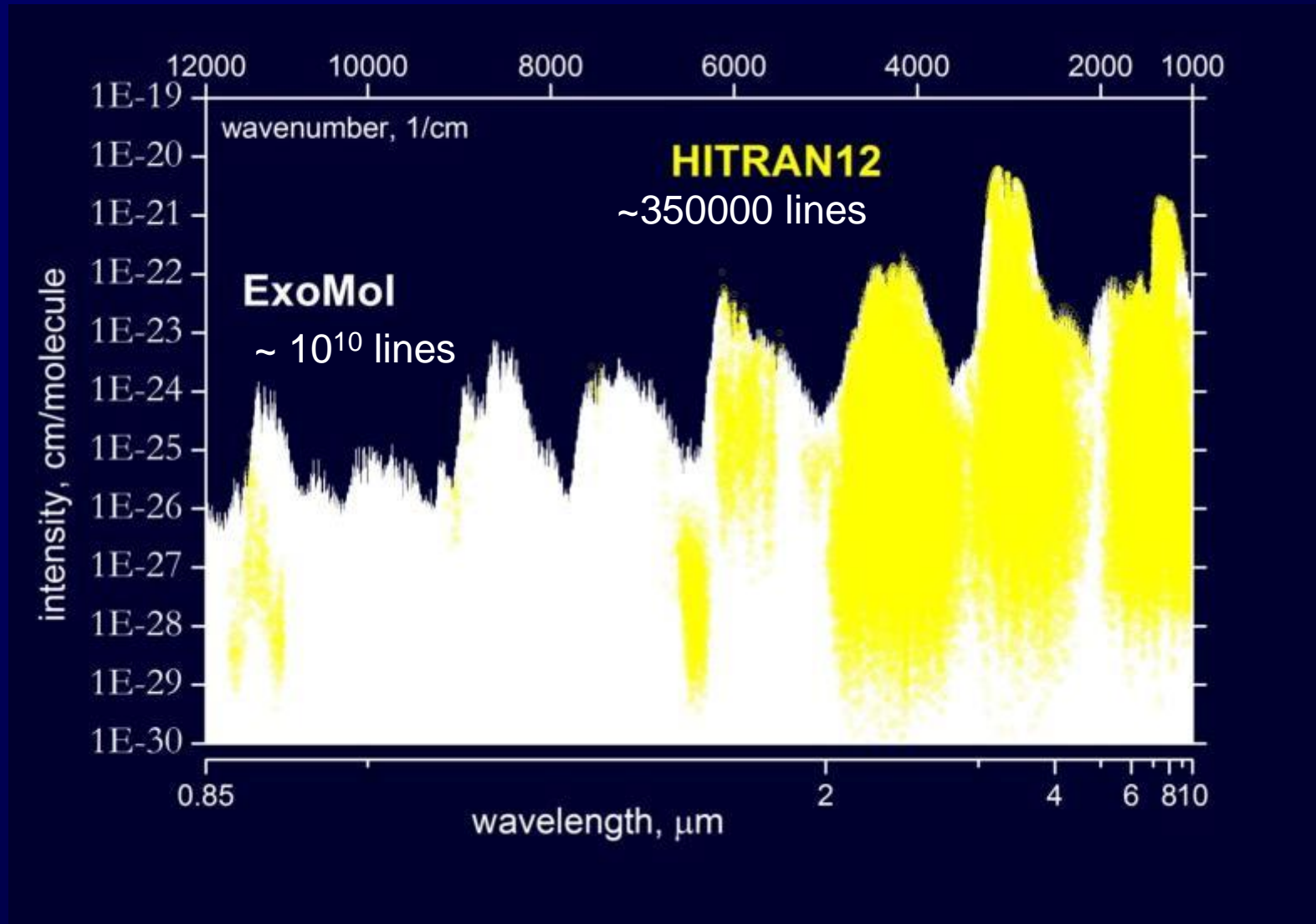
# CH<sub>4</sub> diagonalization: Size of the problem

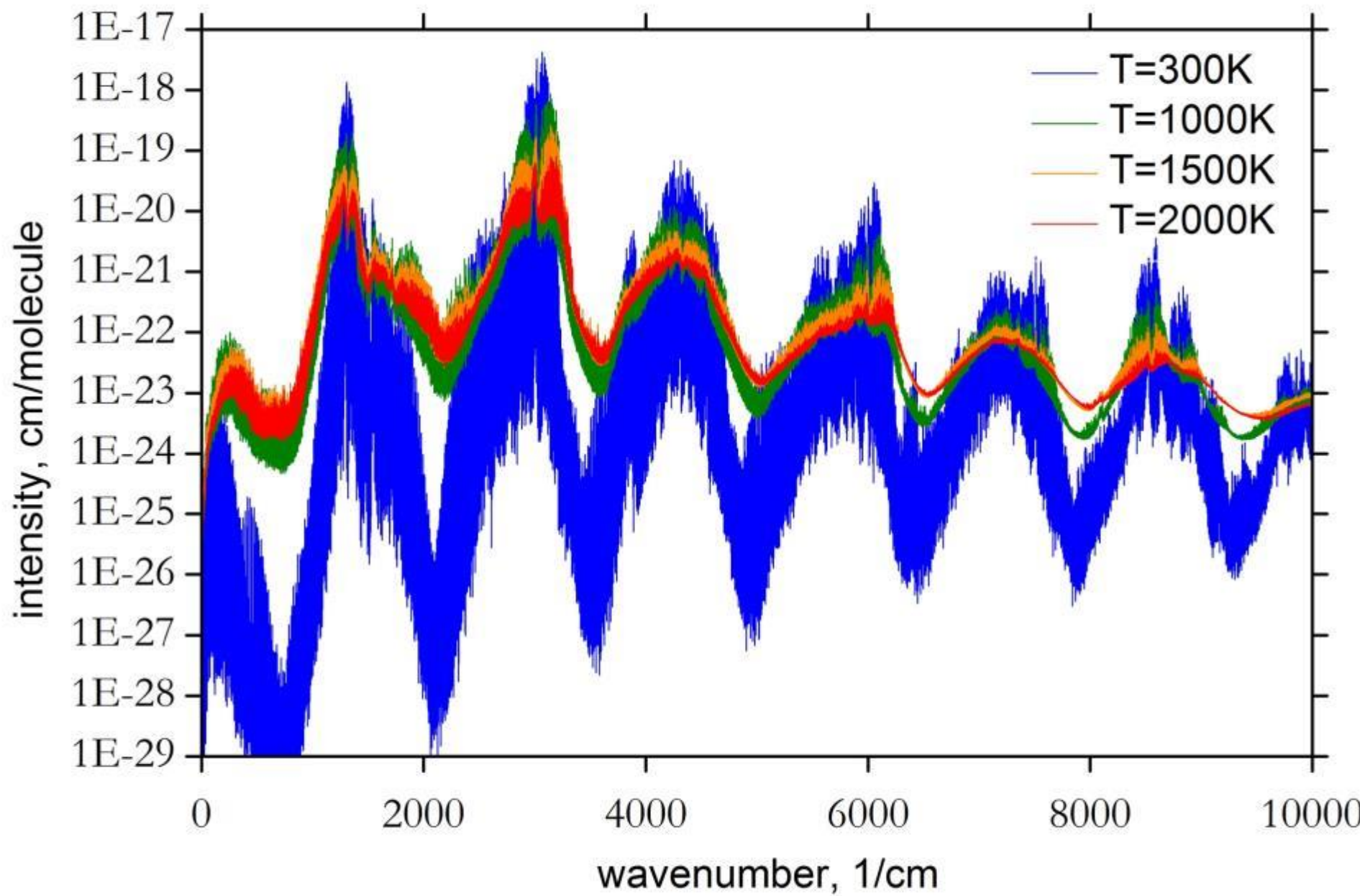


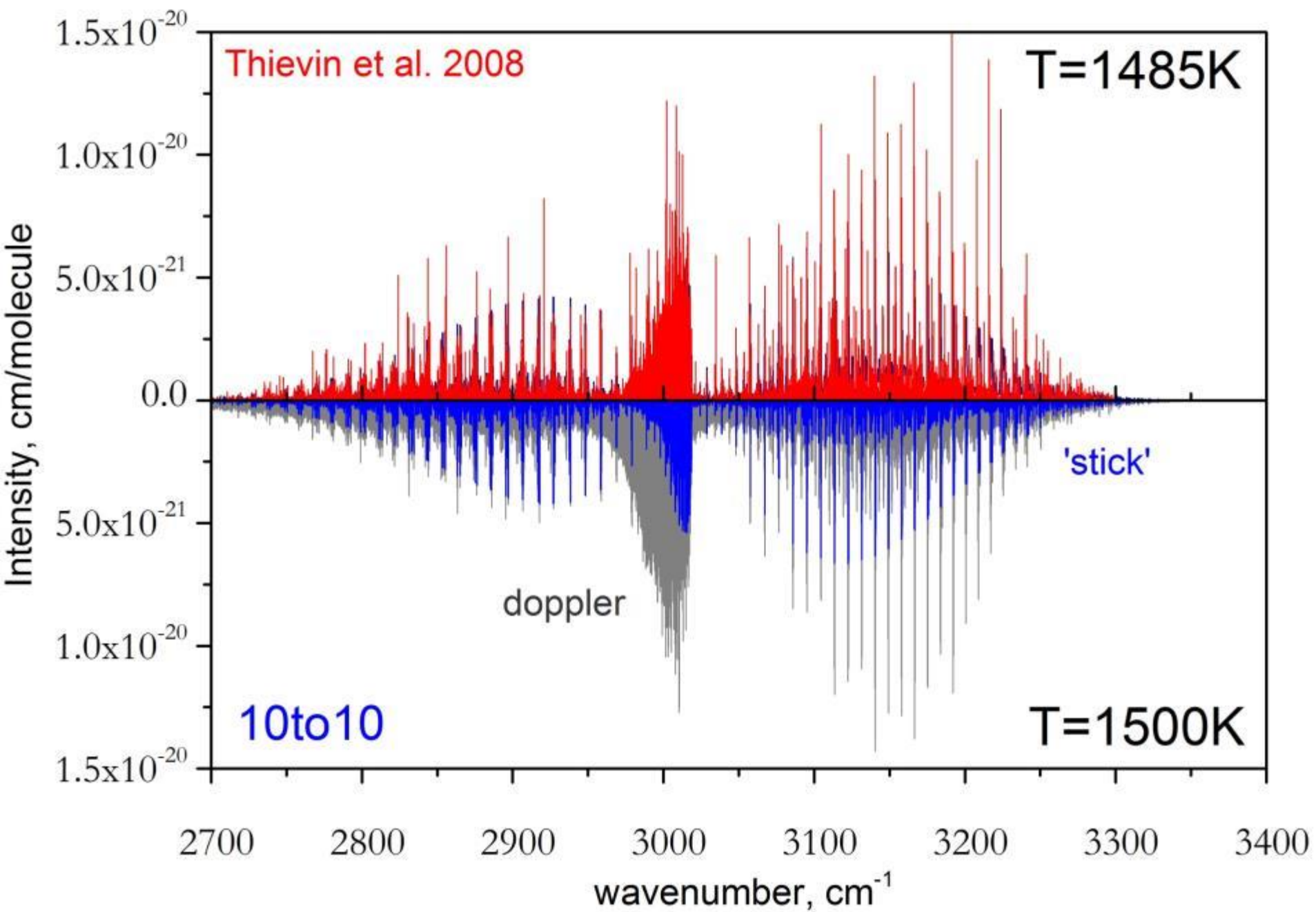
# CH<sub>4</sub> diagonalization: Size of the problem

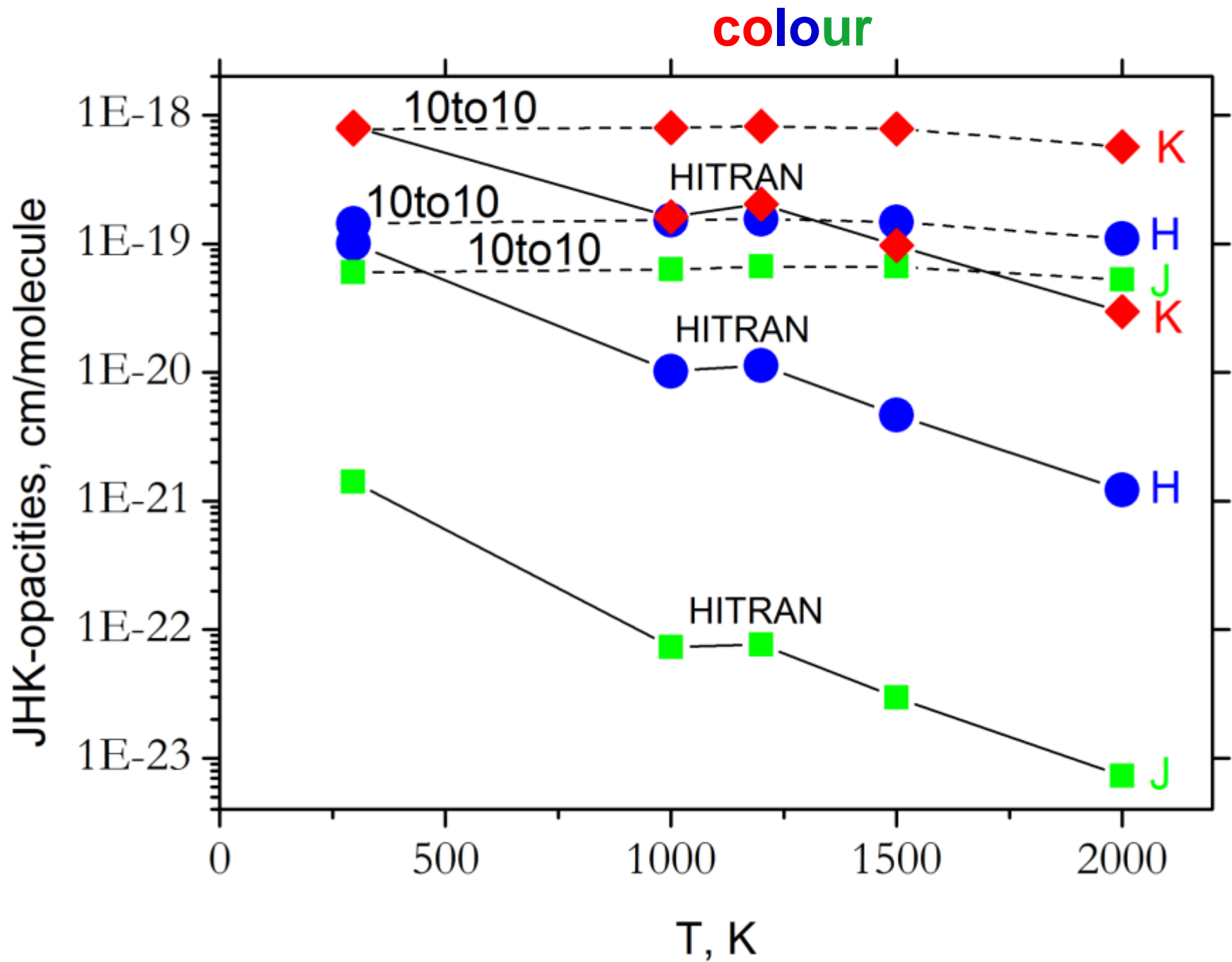


# Absorption spectra of CH<sub>4</sub>: from experimental line list



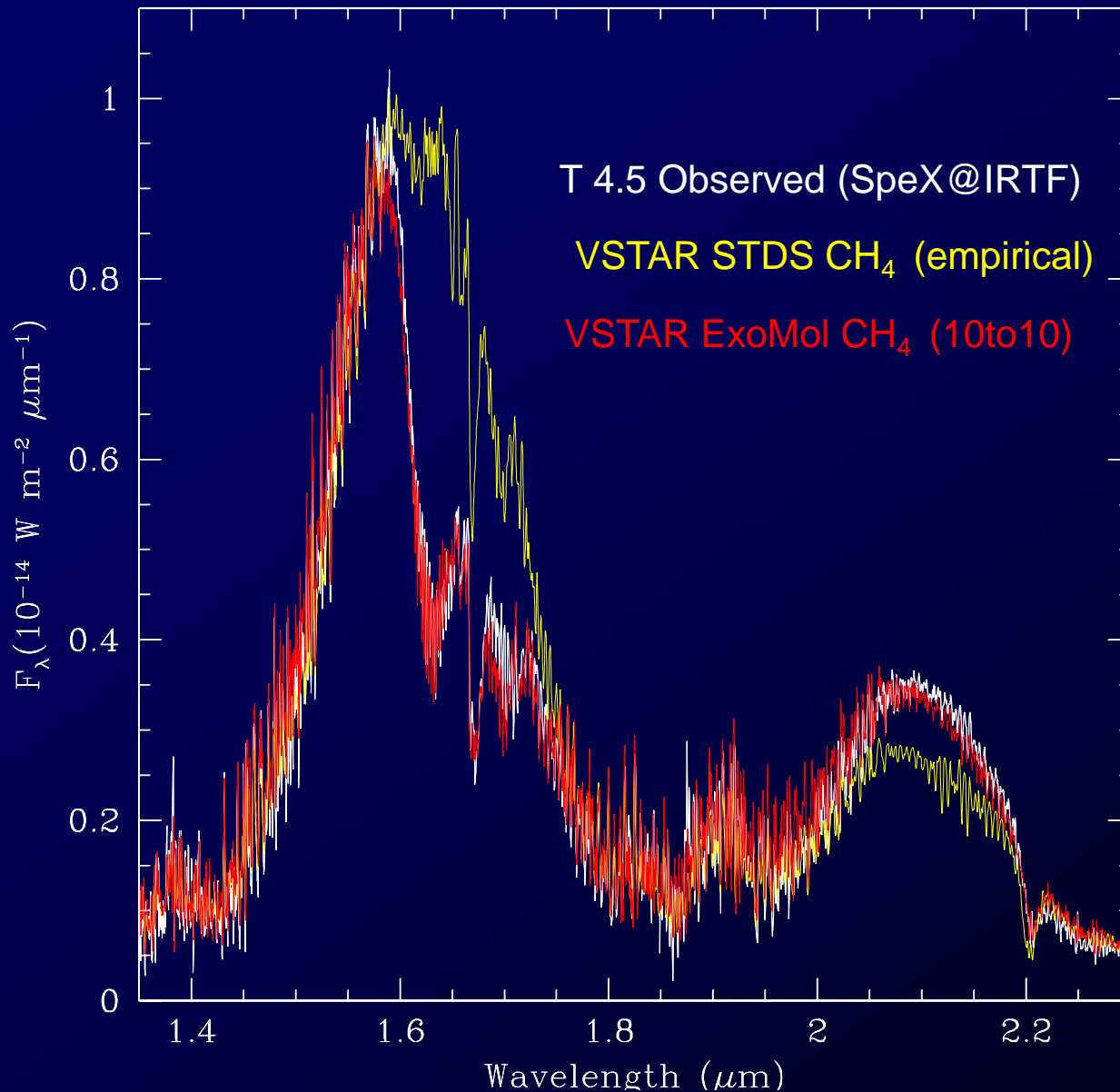








# VSTAR spectra of the T4.5 brown dwarf: a “methane dwarf”



2MASS 0559-14

SN Yurchenko, J  
Tennyson, J Bailey,  
MDJ Hollis, G Tinetti,  
PNAS, 111, 9379 (2014)

Cushing, Rayner, Vacca (2005)

### Published in MNRAS

I. BeH, MgH, CaH

II. SiO

III. HCN/HNC

IV. CH<sub>4</sub>

V. NaCl, KCl

VI. PN

VII. PH<sub>3</sub>

VIII. H<sub>2</sub>CO

IX. AlO

X. NaH

XI. HNO<sub>3</sub>

XII. CS

XIII. CaO

XIV. SO<sub>2</sub>

XV. HOOH

XVI. H<sub>2</sub>S

XVII. SO<sub>3</sub>

XVIII. VO

### (Virtually) Complete

XIX. H<sub>2</sub><sup>18</sup>O, H<sub>2</sub><sup>17</sup>O

XX. H<sub>3</sub><sup>+</sup>

XXI. H<sub>2</sub><sup>16</sup>O

XXII. NO

XXIII. TiO

• NS

• CrH

### In progress

• C<sub>3</sub>

• PH, PO, PS

• TiH

• MnH

• NaO

• SH

• AlH

• C<sub>2</sub>H<sub>4</sub>

• SiH

• HCCH

• SrH

• CH<sub>3</sub>Cl

• SiH<sub>4</sub>

### Planned

• NH<sub>3</sub>

• MgO

• NiH

• FeH

• Larger hydrocarbons

J. Tennyson et al, 327, 73,  
J. Mol. Spectrosc. (2016)

Updated data structure  
New functionality

# ExoMol

Other features:

1. Lifetimes (Tennyson et al, J Phys B, 49, 044002 (2016))
2. Lande g-factors (Semenov et al, J Molec Spectrosc (2016))
3. Pressure broadening (Barton et al, JQSRT (2017) + submitted)  
H<sub>2</sub> and He  
J and T dependence (only)
4. Dipoles for molecular control/orientation effects  
(A Yachmenev, RichMol project)

# First detection of super-Earth atmosphere

Exoplanet 55 Cancri e has a dry atmosphere without any indications of water vapor.

HCN (weakly) detected!

*“Hydrogen cyanide, or prussic acid, is highly poisonous, so it is perhaps not a planet I would like to live on!”*

J. Tennyson, UCL press release



A. Tsiaras, M. Rocchetto, I.P. Waldmann, G. Tinetti, R. Varley, G. Morello, E.J. Barton, S.N. Yurchenko & J. Tennyson, *Astrophys. J.* 820, 99 (2016)



@ExoMol





# ExoMol

Sch, TiH

CO<sub>2</sub>

CrH

NaCl,  
KCl

C<sub>2</sub>H<sub>4</sub>

SO<sub>3</sub> HNO<sub>3</sub>  
SO<sub>2</sub>

NH<sub>3</sub>

CO<sub>2</sub>  
O<sub>3</sub>

CH<sub>4</sub>

C<sub>3</sub>

NH<sub>3</sub>

VO, TiO

HCCH

PH<sub>3</sub>

H<sub>2</sub>CO  
HOOH



@ExoMol





**ExoMol**

**Emma Barton**

**Emil Zak**

**"Zoe" Na**

**Anatoly Pavlyuchko**

**Lorenzo Lodi**

**Maire Gorman**

**Andrey Yachmenev**

**Phillip Coles**

**Dan Underwood**

**Afaf Al-Derzi**

**Oleg Polyansky**

**Sergey Yurchenko**

**Renia Diamantopoulou**

**Laura McKemmish**

**Katy Chubb**

**Clara Sousa-Silva**

**Ahmed Al-Rafaie**



**@ExoMol**



Jonathan Tennyson

# Astronomical Spectroscopy

An Introduction to the Atomic and  
Molecular Physics of Astronomical Spectra

2nd Edition

 World Scientific

[www.worldscibooks.com/physics/7574.html](http://www.worldscibooks.com/physics/7574.html)

About the first edition

*“The best book for anyone who is  
embarking on research in  
astronomical spectroscopy”*

Contemporary Physics (2006)

Published 2011

# Anatoly Pavlyuchko (1956-2015)

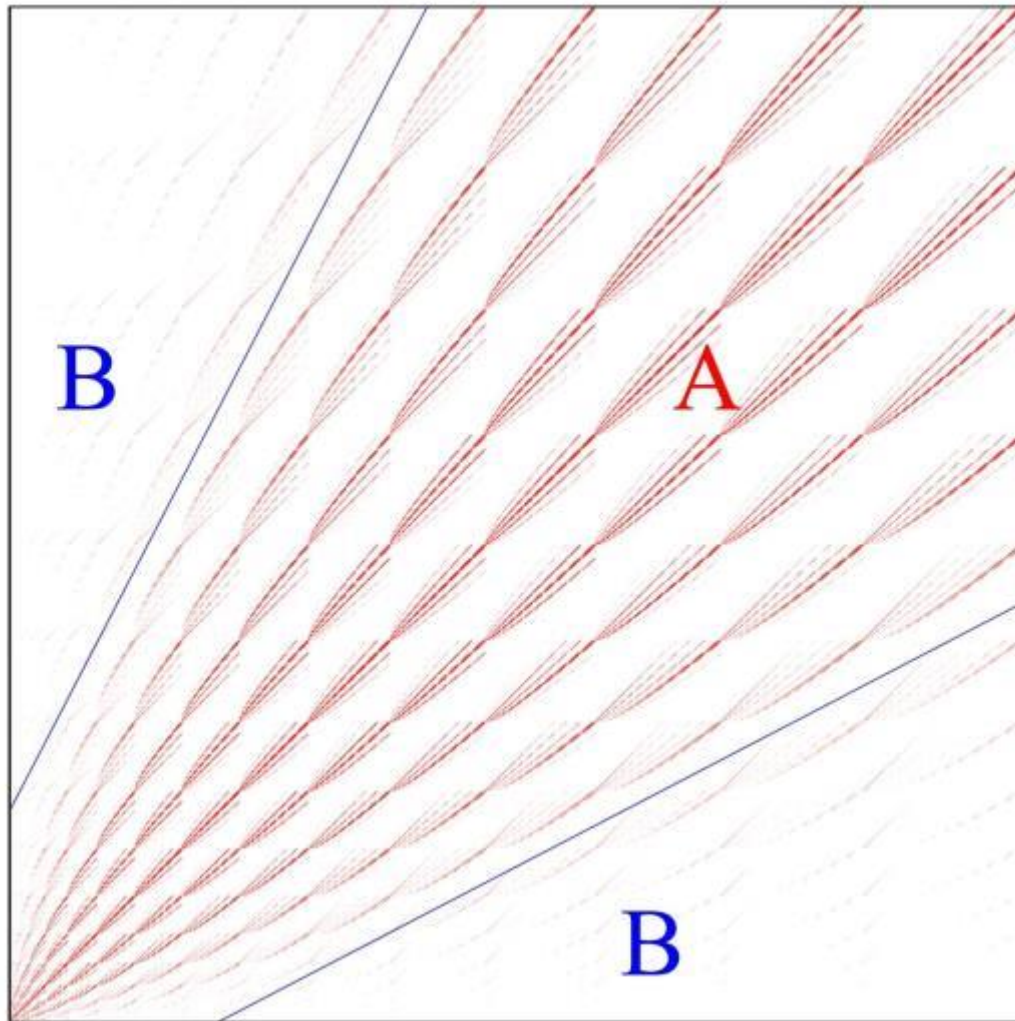
A hybrid variational-perturbation method for calculating rovibrational energy levels of polyatomic molecules, *Mol. Phys.*, 113, 1559 (2015)

A hybrid variational-perturbation calculation of the ro-vibrational spectrum of nitric acid, *J. Chem. Phys.*, 142, 094309 (2015)

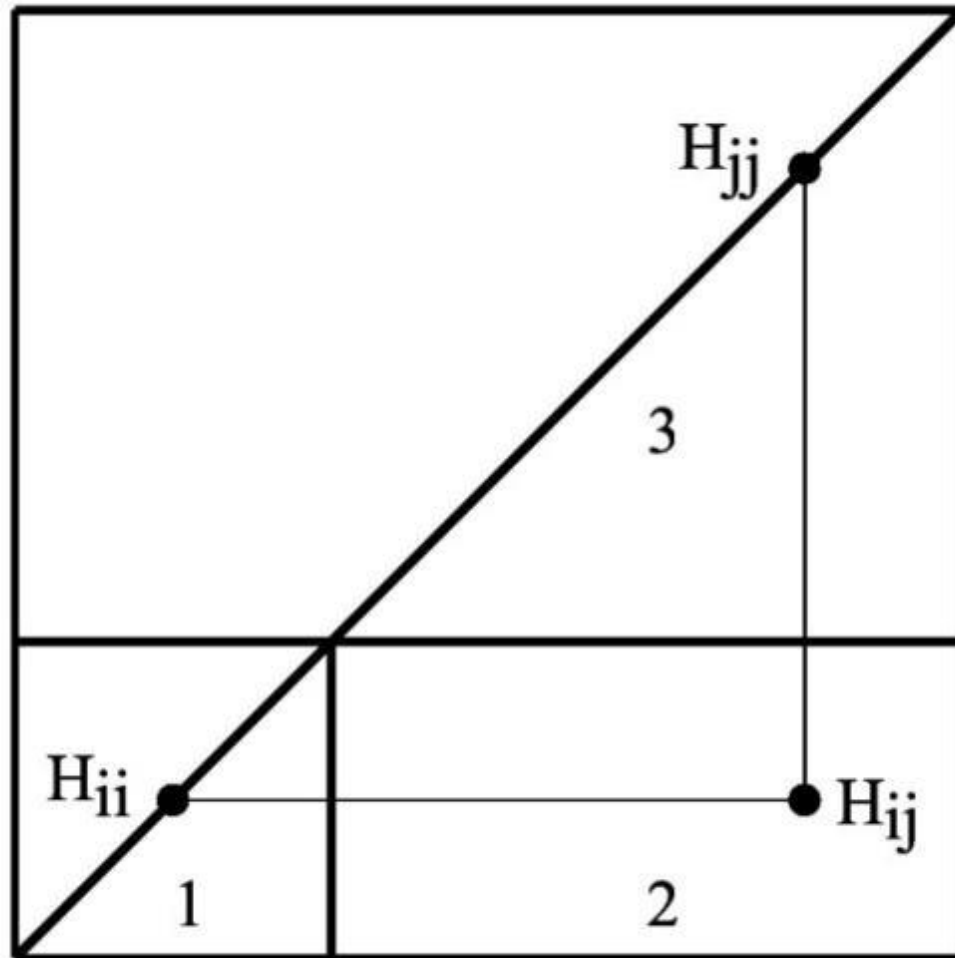
ExoMol molecular line lists XII:  
A Hot Line List for nitric acid, *MNRAS*, 452, 1702 (2015).







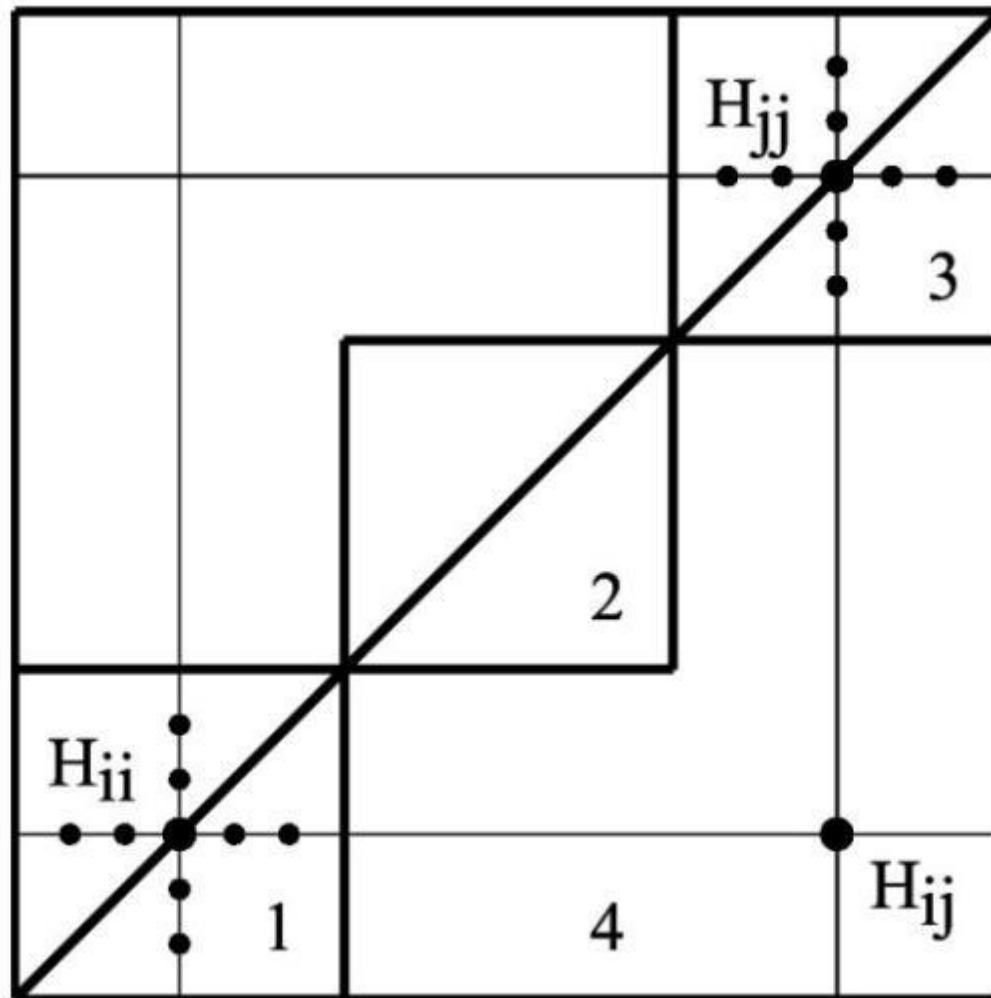
Vibrational Hamiltonian matrix:  
A dense near the diagonal,  
B sparse elsewhere



Vibrational Hamiltonian matrix:

Diagonalise region 1 only

Include effects of region 2 using perturbation theory

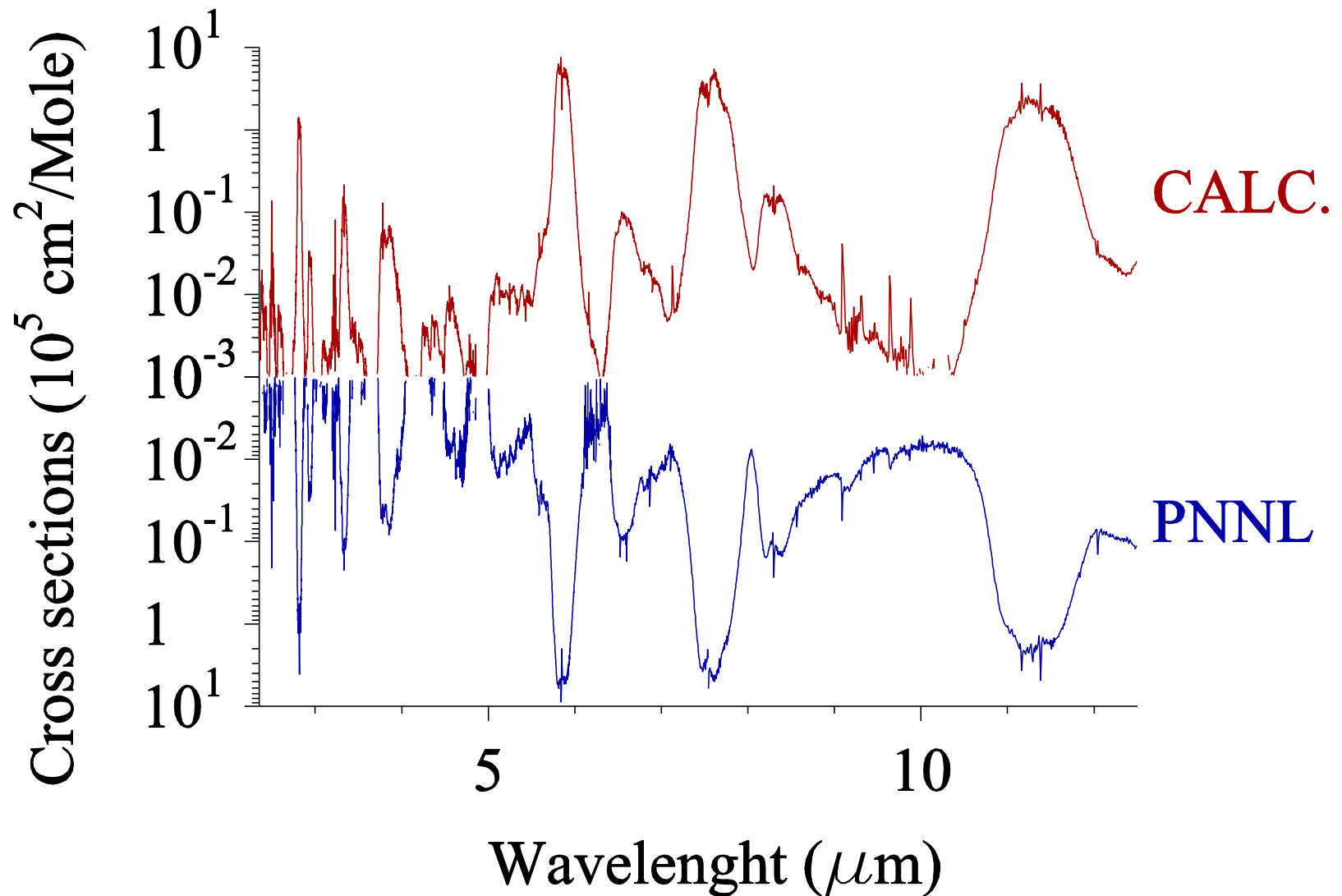


## Rotation-Vibration Hamiltonian:

Treat vib state by vib state

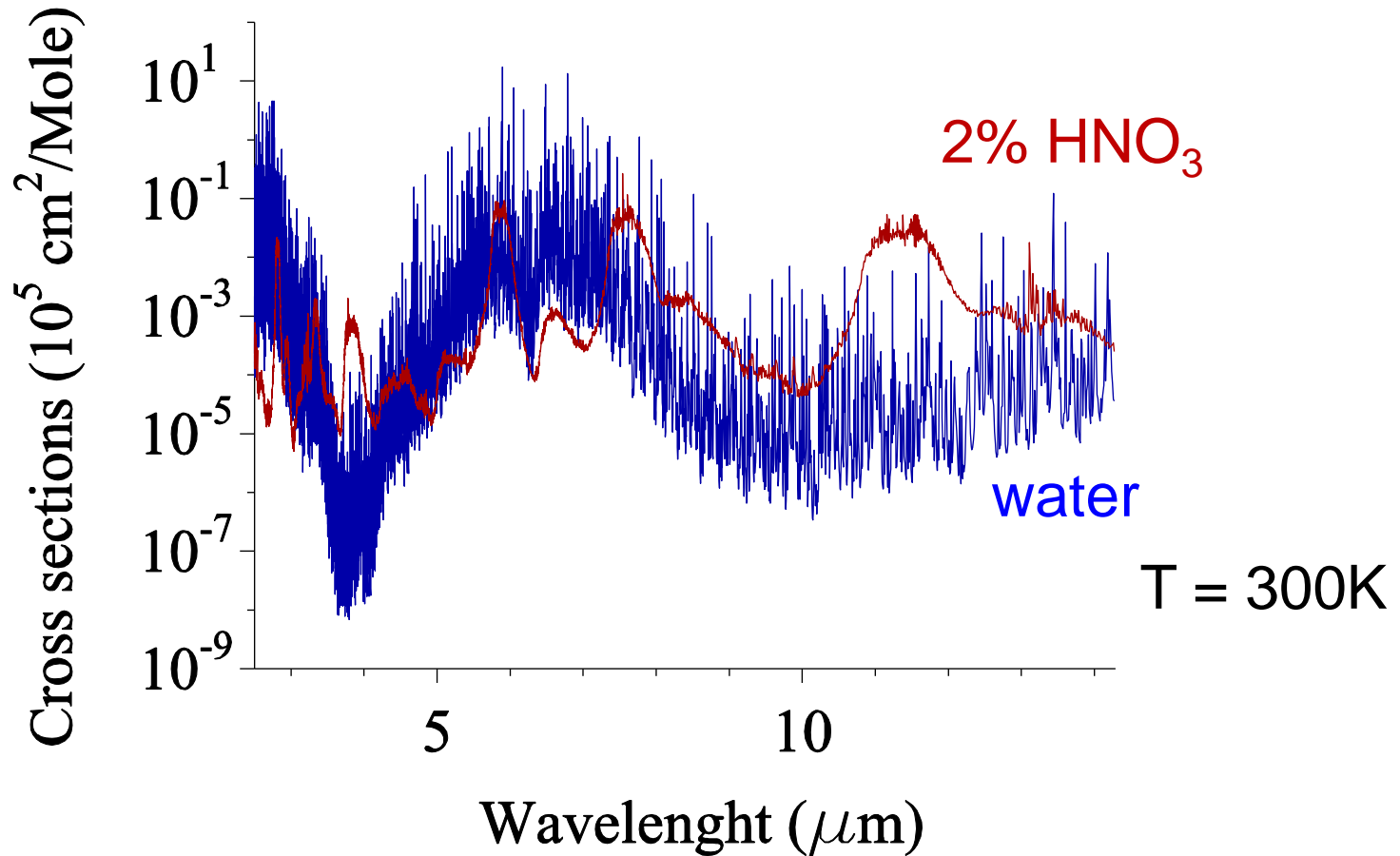
Include Coriolis coupling via perturbation theory

# T = 298 K spectrum of nitric acid (HNO<sub>3</sub>)



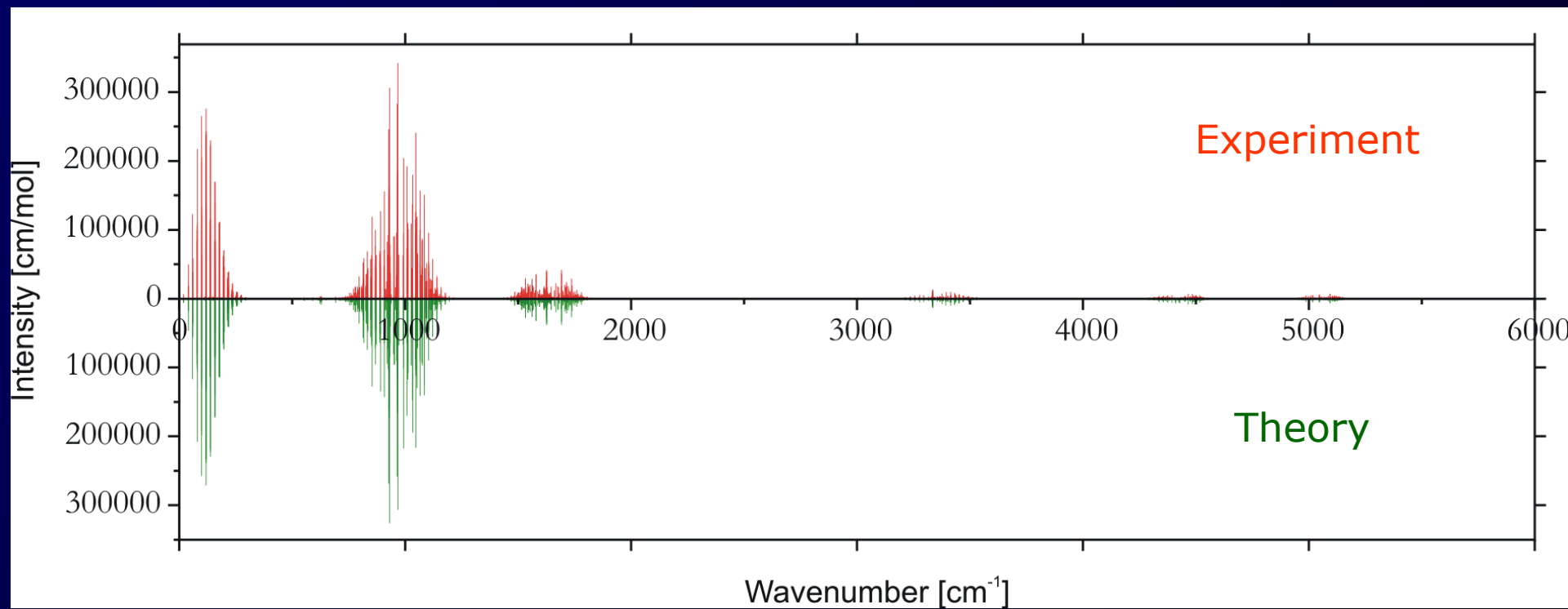


# Nitric acid as a bio-signature?

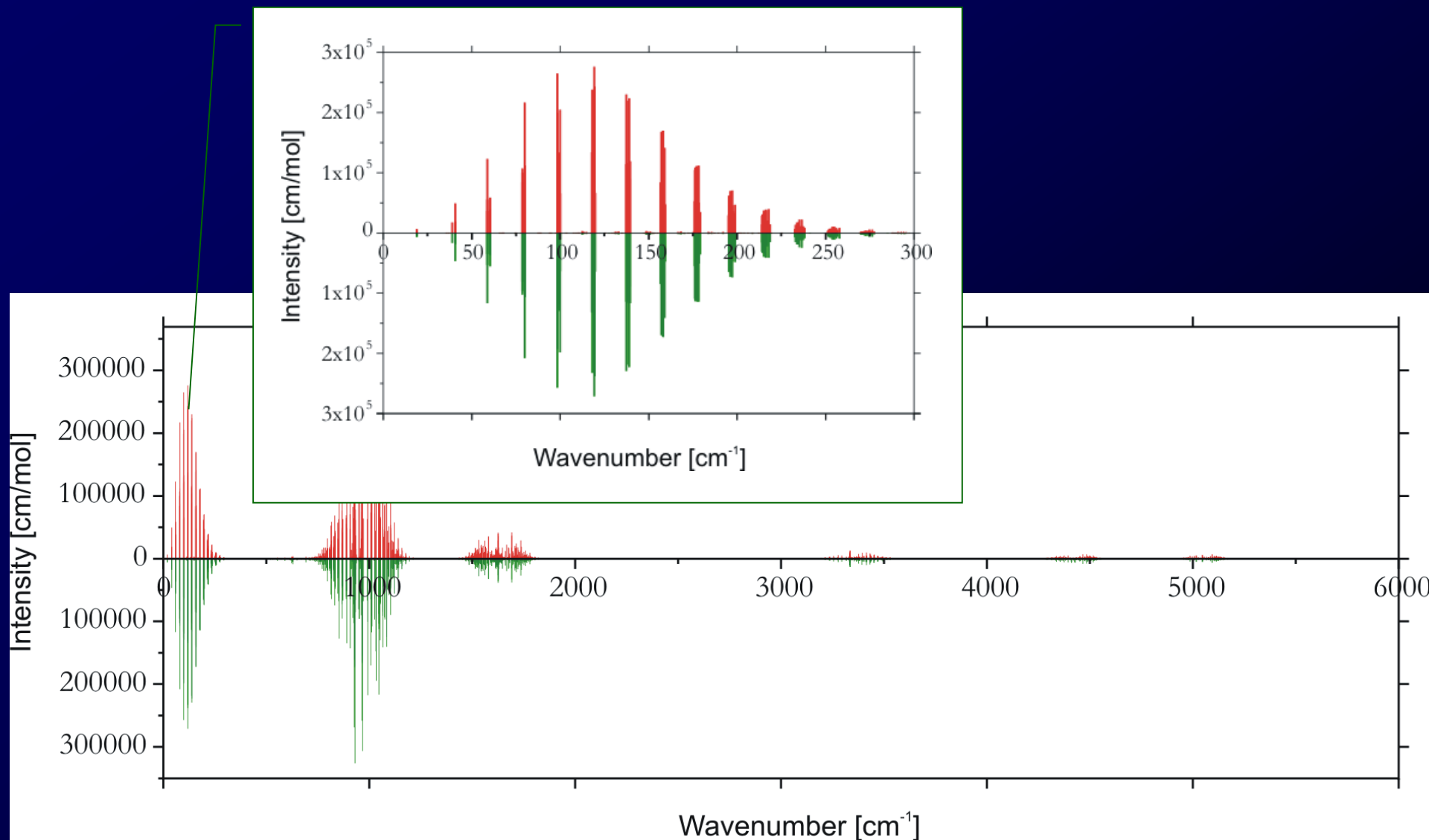


A.I. Pavlyuchko, S.N. Yurchenko & J. Tennyson, Mon. Not. R. astr. Soc., 452, 1702 (2015).

# Absorption ( $T=300\text{K}$ ) spectrum of $\text{NH}_3$ : Accuracy



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